

SYNTHESIS AND CHARACTERIZATION OF LASER-IRRADIATED  
POLYANILINE/RICE HUSK (SILICA) NANOCOMPOSITES FOR ELECTRICAL  
CONDUCTIVITY APPLICATIONS

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## DEDICATION

To my loving and caring mother Qismah Shihab, my father Oudah Mezan, my beloved wife Janan Obaid, and my sons Mohammad Salim, Zahraa Salim and Moqtadah Salim.



PTTA UTHM  
PERPUSTAKAAN TUNKU TUN AMINAH

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## ABSTRACT

Polymer nanocomposites contain inorganic fillers that have metallic particles dispersed in polymer matrices. These are very appealing for optical and electrical conductivity applications. Polyaniline/ silica (rice husk ash CCl (SiO<sub>2</sub>)), nanocomposites (PANI/RHACCl (SiO<sub>2</sub>)) nanocomposites can be manipulated through different methods of fabrication for desired purposes. An innovative method for nanocomposite materials fabrication has been developed using sol-gel that has been applied to polyaniline/RHACCl (Silica) nanocomposites. Laser irradiation is also used as a way to manipulate the structure of the nanocomposite to investigate the effect of enhancing new optical and electrical properties. The laser irradiation power of value 13.17 w/cm<sup>2</sup> leads to the formation of graphite oxide and carbon percentage increment, which is based on laser irradiation time and distribution percentage of RHACCl nanoparticles. The sodium silicate was obtained using hydrochloric acid (HCl) wash to produce silica. Polyaniline was synthesized by chemical oxidative polymerization, then mixed with RHACCl (SiO<sub>2</sub>) in the presence of toluene and trimethylamine to form. The samples were analyzed by the structural, chemical, morphology, optical-electrical conductivity, and thermal properties behaviors. The amorphous nature of silica was confirmed using XRD analysis. The sample size was obtained from (6-9 nm) by FESEM morphology studies. EDX atomic weight was measured at 57.10% of carbon, 26.78% of oxygen, 3.49% of silicon, 10.09% of nitrogen, 0.68% sulfur, and 1.87% of chlorine. UV-Vis absorption wavelength was determined at 382 nm. The best energy band gap of PANI was 1.13 eV and the best laser irradiation time was (40) min. Direct current (DC) electrical conductivity was calculated for the best ratio conductivity values were found to be at PANI / RHACCl (SiO<sub>2</sub>) NCs after laser irradiation (1: 0.5 =  $1.10 \times 10^{-1} \text{ S cm}^{-1}$ , 2: 0.5 =  $1.36 \times 10^{-1} \text{ S cm}^{-1}$  and 3: 0.5 =  $2.08 \times 10^{-1} \text{ S cm}^{-1}$ ) in the time 40 min.

## ABSTRAK

Polimer Nanokomposit mengandungi bahan pengisi tidak organik yang mempunyai zarah logam terserak di dalam matriks polimer. Ianya usaha yang sangat menarik terutama dalam aplikasi kekonduksian optik dan elektrik. Polianilin/silika (sekam padi terproses CCl (SiO<sub>2</sub>)) nanokomposit (PANI/RHACCl (SiO<sub>2</sub>) nanokomposit) boleh dimanipulasi melalui kaedah fabrikasi yang berbeza untuk tujuan yang diinginkan. Kaedah inovatif fabrikasi bahan nanokomposit melalui teknik sol-gel telah digunakan dalam kajian aplikasi PANI/RHACCl (SiO<sub>2</sub>) nanokomposit. Penyinaran laser juga digunakan sebagai kaedah memanipulasi struktur nanokomposit dalam menyiasat kesan peningkatan sifat optik dan elektrik baharu. Kuasa penyinaran laser dengan nilai 13.17 w/cm<sup>2</sup> membawa kepada pembentukan grafit oksida dan kenaikan peratusan karbon, yang berdasarkan masa penyinaran laser dan peratusan pengedaran nanopartikel RHACCl. Natrium silikat diperoleh menggunakan pencucian asid hidroklorik (HCl) untuk menghasilkan silika. Polianilin telah disintesis oleh pempolimeran oksidatif kimia yang mengandungi kumpulan berfungsi aktif dalam molekul, kemudian dicampur dengan RHACCl (SiO<sub>2</sub>) dengan kehadiran toluena dan trimetilamin untuk membentuk. Sampel oleh struktur, kimia, morfologi, kekonduksian optik-elektrik, dan kelakuan sifat terma. Sifat amorf silika telah disahkan menggunakan analisis XRD. Saiz sampel diperoleh daripada (6-9 nm) oleh kajian morfologi FESEM. Berat atom EDX diukur pada 57.10% karbon, 26.78% oksigen, 3.49% silikon, 10.09% nitrogen, 0.68% sulfur, dan 1.87% klorin. UV-Vis ditentukan pada 382 nm. jurang jalur tenaga terbaik PANI ialah 1.13 eV dan masa penyinaran laser terbaik ialah (40) min. Kekonduksian elektrik arus terus (DC) dikira pada nilai kekonduksian nisbah terbaik adalah pada penyinaran laser dengan nisbah (PANI / RHACCl (SiO<sub>2</sub>) NCs (1: 0.5 = 1.10x10<sup>-1</sup> S cm<sup>-1</sup>, 2: 0.5 = 1.36x10<sup>-1</sup> S cm<sup>-1</sup> dan 3: 0.5 = 2.08x10<sup>-1</sup> S cm<sup>-1</sup>) adalah pada masa sinaran 40 min.

## TABLE OF CONTENTS

	<b>TITLE</b>	<b>i</b>
	<b>DECLARATION</b>	<b>ii</b>
	<b>DEDICATION</b>	<b>iii</b>
	<b>ACKNOWLEDGEMENT</b>	<b>iv</b>
	<b>ABSTRACT</b>	<b>v</b>
	<b>ABSTRAK</b>	<b>vi</b>
	<b>TABLE OF CONTENTS</b>	<b>vii</b>
	<b>LIST OF TABLES</b>	<b>xiii</b>
	<b>LIST OF FIGURES</b>	<b>xv</b>
	<b>LIST OF SYMBOLS AND ABBREVIATIONS</b>	<b>xx</b>
	<b>LIST OF APPENDICES</b>	<b>xxii</b>
<b>CHAPTER 1</b>	<b>INTRODUCTION</b>	<b>1</b>
	1.1 Background of the study	1
	1.2 Polyaniline (PANI)	2
	1.3 Silica Nanoparticles (SiO <sub>2</sub> NPS)	3
	1.4 Polyaniline/Rice Husk Ash Silica Nanocomposites	4
	1.5 Problem statement	4
	1.6 Aim and objectives of the study	6
	1.7 Scope of the study	7
	1.8 Thesis organization	8
<b>CHAPTER 2</b>	<b>LITERATURE REVIEW</b>	<b>9</b>
	2.1 Introduction	9
	2.2 Polyaniline (PANI)	10
	2.2.1 Polyaniline (PANI) chemical structure	11
	2.2.2 Chemical properties of polyaniline by Oxidative polymerization	12
	2.2.3 Conductivity of Polyaniline (PANI)	14
	2.2.4 Interconversion of various Polyaniline (PANI)	

	oxidation state	15
2.3	Silica (SiO <sub>2</sub> )	17
2.3.1	Silica nanoparticles	19
2.3.2	Rice Husk Ash (RHA)	20
2.3.3	Structure of silica preparation from Rice Husk Ash	21
2.3.4	Extraction of Amorphous Rice Husk Ash Silica	21
2.3.5	Modification of silica	24
2.3.6	Preparation Sol-gel techniques	25
2.3.7	Surface modification of silica with 3 (chloropropyl) triethoxysillin	27
2.3.8	Modified from Rice Husk	27
2.4	Synthesis Polyaniline/ Silica nanocomposites (PANI/SiO <sub>2</sub> ) NCs	29
2.5	Fabrication Methods and application of polyaniline/silica nanocomposites	34
2.6	Techniques used for the preparation of Polyaniline /Rice Husk Ash Silica nanocomposites (PANI/RHACCI (SiO <sub>2</sub> ))	36
2.7	Polymer nanocomposite	40
2.7.1	Solution mixing method	41
2.7.2	Ex-situ technique for nanocomposite fabrication	41
2.7.3	In-situ technique for nanocomposite fabrication	42
2.8	Carbon material	42
2.9	Phonon and Energy band gap	43
2.10	Effect of Solvent Properties	44
2.10.1	Effect of environmental parameters	45
2.11	Conductivity enhancement using laser irradiation	46
2.11.1	Characteristics of lasers	46
2.11.2	Laser operation	47
2.12	Summary	47
<b>CHAPTER 3</b>	<b>METHODOLOGY</b>	<b>48</b>
3.1	Introduction	48
3.2	Experimental design	48

3.3	Raw materials and apparatus	50
3.4	Determination of the concentration of the solution for Polyaniline (PANI) and Rice Husk Ash (Silica)	51
3.5	Synthesis of Polyaniline (PANI) nanoparticles	51
3.6	Chemical Synthesis for Oxidative polymerization	53
3.7	Synthesis of Rice Husk Ash (Silica) Nanoparticles by Sol-Gel Method	53
3.8	Experimental procedure	54
3.8.1	Amorphous Silica Extracted from Rice Husk	55
3.9	Preparation of Polyaniline/ Rice Husk Ash silica nanocomposite (PANI/ RHACCl (SiO <sub>2</sub> ) NCPs	57
3.10	Ultrasonic agitation	58
3.11	Sample Palletizing	59
3.11.1	Laser bombardment treatment for the pellet	60
3.12	Sample characterization technique	61
3.12.1	X-ray Diffraction XRD	62
3.12.2	Fourier Transmission Infrared FTIR	64
3.12.3	Field emission scanning electron microscopy (FESEM)	67
3.12.4	Energy-Dispersive X-ray spectroscopy (EDX)	68
3.12.5	Ultraviolet-Visible spectroscopy (UV-Vis)	69
3.12.6	Electrical conductivity Test using Four-point Probe	73
3.12.7	Thermogravimetry analysis (TGA) of polymer nanocomposites	77
<b>CHAPTER 4</b>	<b>RESULTS AND DISCUSSION</b>	<b>78</b>
4.1	Introduction	78
4.2	Analysis of synthesized and optimization Polyaniline (PANI) nanoparticles	78
4.2.1	Crystalline pattern of Polyaniline (PANI) Nanoparticles	79
4.2.2	Functional group analysis of Polyaniline (PANI) nanoparticles	79
4.2.3	Field-Effect Scanning Electron Microscopy	



(FESEM) Analysis of Polyaniline (PANI)	80
4.2.4 The elemental analysis (EDX) of Polyaniline (PANI)	81
4.2.5 Optical analysis of the Polyaniline (PANI) nanoparticles	82
4.2.6 Optical energy band gap analysis of Polyaniline (PANI) nanoparticles	83
4.2.7 D.C conductivity of irradiated Polyaniline (PANI) nanoparticle analysis	84
4.2.8 Thermal (TGA) properties of polyaniline	85
4.3 Laser irradiation of Polyaniline (PANI) analysis	86
4.3.1 Structural analysis of irradiated Polyaniline (PANI)	86
4.3.2 Functional group analysis of irradiated Polyaniline (PANI)	89
4.3.3 Surface morphology analysis of irradiated Polyaniline (PANI)	90
4.3.4 Elemental composition analysis of irradiated Polyaniline (PANI) nanoparticle	92
4.3.5 Optical properties analysis of irradiated Polyaniline (PANI)	94
4.3.6 D.C conductivity of irradiated Polyaniline (PANI) nanoparticle analysis	96
4.4 Analysis of synthesized Rice Husk Ash silica nanoparticles	99
4.4.1 Structural analysis of Rice Husk Ash Silica Particle	99
4.4.2 Chemical bonding analysis of synthesized silica particles	100
4.4.3 Surface morphology analysis of Rice Husk Ash silica particles	101
4.4.4 Elemental composition analysis of RHACCl (SiO <sub>2</sub> ) nanoparticle	102
4.4.5 Optical energy band gap analysis of synthesized Polyaniline (PANI) particles	103



4.4.6	Thermal properties analysis (TGA) of RHACCl (SiO <sub>2</sub> ) nanoparticles	103
4.5	Analysis of synthesized polyaniline/Rice Husk silica nanocomposite	104
4.5.1	Structural analysis of Polyaniline/Rice Husk Ash Silica nanocomposite	104
4.5.2	Chemical bonding analysis of synthesized PANI/RHACCl silica particles	106
4.5.3	Surface morphology analysis of Polyaniline/Rice Husk Ash silica nanocomposite	108
4.5.4	Elemental composition analysis of PANI/RHACCl silica nanocomposite	109
4.5.5	Study of UV-Vis Absorption Spectra of PANI/RHACCl	111
4.5.6	Calculation of energy band gap of PANI/RHACCl nanocomposites	112
4.5.7	The Electrical Conductivity studies of (PANI/RHACCl NCs)	113
4.5.8	Thermal properties (GTA) analysis of PANI/RHACCl (SiO <sub>2</sub> ) nanocomposites	114
4.6	Analysis of laser-irradiated PANI/RHACCl nanocomposites	115
4.6.1	Structural analysis of irradiated PANI/RHACCl nanocomposites	116
4.6.2	Chemical bonding analysis of irradiated PANI/RHACCl nanocomposites	119
4.6.3	Surface morphology analysis of irradiated PANI/RHACCl nanocomposites	123
4.6.4	Elemental composition analysis of irradiated PANI/RHACCl nanocomposite	125
4.6.5	D.C conductivity of irradiated PANI/RHACCl nanocomposite analysis	129

**CHAPTER 5 CONCLUSIONS AND RECOMMENDATIONS 131**

5.1	Conclusions	131
5.2	Future recommendations	133
	<b>REFERENCES</b>	<b>134</b>
	<b>APPENDICES</b>	<b>151</b>
	<b>VITA</b>	<b>160</b>



## LIST OF TABLES

2.1	The different forms of Polyaniline (PANI) on its color and conductivity	13
2.2	Chemical composition of RHA after burning out at (700-800) °C for 6 h	20
2.3	Silica types with their advantage, structure, polymer nanocomposite, and techniques	35
2.4	Some applications of PANI-RHACCl (SiO <sub>2</sub> ) nanocomposite	36
2.5	An overview of polymers exposed to different radiation sources research work conducted from 2014 to 2020	39
3.1	Basic raw materials for Rice Husk Ash Silica (RHACCl (SiO <sub>2</sub> )) nanoparticles	50
3.2	Basic raw materials for Polyaniline (PANI) nanoparticles	50
3.3	Basic raw materials for Polyaniline/Rice Husk Ash Silica (PANI/RHACCl (SiO <sub>2</sub> )) nanocomposites	51
3.4	PANI and PANI/SiO <sub>2</sub> with different concentrations and different times of laser irradiation exposure	60
3.5	List of instruments with specifications employed for characterization	61
3.6	The infrared spectrum region represents an energy absorption pattern	65
3.7	Assignment of IR absorption peaks	67
4.1	Elemental composition of Polyaniline (PANI) nanoparticles	82
4.2	The conductivity of Polyaniline (PANI) nanoparticles	84

4.3	Structural parameters of irradiated Polyaniline (PANI)	88
4.4	Positions and assignments of the FTIR vibration bands of Polyaniline (PANI)	91
4.5	The average size D (nm) from FESEM analysis	92
4.6	Estimated values of energy band gap and number of carbon atoms (N) in a cluster of irradiated and non-irradiated Polyaniline (PANI)	97
4.7	The conductivity of Pure Polyaniline (PANI) and irradiated polyaniline with Diode laser	98
4.8	Structural Parameter of the prepared RHACCl (SiO <sub>2</sub> ) nanoparticles	99
4.9	Elemental composition of RHA and RHACCl (SiO <sub>2</sub> ) nanocomposites	102
4.10	The average size D (nm) from FESEM analysis of PANI/RHACCl	108
4.11	Elemental composition for the synthesized PANI/RHACCl (SiO <sub>2</sub> ) nanocomposite	110
4.12	Estimated values of energy band gap for polyaniline and PANI/RHACCl nanocomposites	112
4.13	Conductivity of Polyaniline (PANI) contains RHACCl (SiO <sub>2</sub> ) nanoparticles	114
4.14	PANI/RHACCl (1:0.5) the average size D (nm) from FESEM analysis	124
4.15	PANI/RHACCl (2:0.5) the average size D (nm) from FESEM analysis	125
4.16	D.C electrical conductivity of irradiated PANI/RHACCl nanocomposite	130
4.17	Comparison of conductivity for the current work (PANI/RHACCl SiO <sub>2</sub> ) with previous studies.	130

## LIST OF FIGURES

2.1	General structure of Polyaniline (PANI)	12
2.2	Diverse potential Polyaniline (PANI) oxidation states	12
2.3	Interconversion of different oxidation states of Polyaniline (PANI) (redox procedure)	13
2.4	Homo polymerization of Polyaniline (PANI)	14
2.5	Emeraldine salt (ES) is protonated to emeraldine base (EB) in the alkaline form. A-is a random molecule, for example, the chloride	16
2.6	Energy band gap in materials	17
2.7	Top-Down and Bottom-Up approaches	25
2.8	Phonon band gap of graphene	44
3.1	Flow chart of the research methodology	49
3.2	(a) Synthesized Polyaniline (PANI) with the stirred in 0 °C. (b) Filtration and washing of the Polyaniline	52
3.3	Synthesis of Polyaniline (PANI)	52
3.4	Burning stages RH (a) RH. (b) RHA less than 800 °C (c) RHA of 800°C (d) RHACCl silica.	54
3.5	The simple reaction sequence and the possible structures for RHACCl (a) three siloxane bonds to silica (b) two siloxane bonds to silica	55
3.6	Synthesis of RHACCl (SiO <sub>2</sub> )	56
3.7	The reaction sequences for the synthesis of PANI-RHACCl	57
3.8	Synthesis of PANI/RHACCl (SiO <sub>2</sub> ) NCs	58
3.9	Ultrasonic in biology a laboratory of FAST faculty in UTHM	59
3.10	The pellets of Polyaniline (PANI) and PANI/RHACCl	59
3.11	Laser irradiation setup	60

3.12	X-ray diffraction (XRD) with $\text{CuK}\alpha$ radiation of 1.542 Å (PANalytical XPERT-PRO MPD X-ray diffractometer system)	62
3.13	The diffraction conditions of the Bragg equation	64
3.14	The infrared spectrum region represents an energy absorption pattern	65
3.15	Fourier transformation infrared spectroscopy (ATR-FTIR) Perkin Elmer FTIR spectrometer	66
3.16	Typical Infrared Absorption Regions	67
3.17	FESEM and EDS Spectroscopy (FESEM/EDS-JOEL)	68
3.18	UV-3101PC Shimadzu UV-VIS spectrophotometer	71
3.19	Four-point probe	74
3.20	Sketch of four-point probe	75
4.1	X-ray Diffraction of Polyaniline (PANI) FTIR Spectrum of	79
4.2	pure Polyaniline (PANI)	80
4.3	FESEM image morphology of pure Polyaniline (PANI) nanoparticles	81
4.4	EDX spectrum of Polyaniline (PANI) nanoparticles	82
4.5	UV-vis Analysis of Polyaniline (PANI) nanoparticles	83
4.6	Plot of Polyaniline (PANI) nanoparticles $(\alpha h\nu)^2$ versus the photon energy (hv)	84
4.7	TGA/DTG curve of Polyaniline (PANI) nanoparticles showing mass loss	86
4.8	XRD pattern for irradiated polymer Polyaniline (PANI) nanocomposite with Diode laser	87
4.9	FTIR spectra for irradiated Polyaniline (PANI) with Diode laser at laser at (a) non-irradiated (0 min) (b) 10 min (c) 20 min (d) 30 min (e) 40 min	90
4.10	FESEM images for irradiated Polyaniline (PANI) with Diode laser at (a) 10 min (b) 20 min (c) 30 min (d) 40 min (30-50K magnification)	92
4.11	EDX spectrum of irradiated Polyaniline (PANI) nanoparticle with Diode laser at (a)10 min (b) 20 min (c)	

	30 min (d) 40 min	93
4.12	EDX quantitative result of irradiated Polyaniline (PANI) nanoparticle with Diode laser at (a) 10 min (b) 20 min(c) 30 min (d) 40 min	94
4.13	UV-Vis absorption spectra for irradiated Polyaniline (PANI) with Diode laser at (a)10 min (b) 20 min (c) 30 min (d) 40 min	95
4.14	Energy band gap for irradiated Polyaniline (PANI) with Diode laser at (a)10 min (b) 20 min (c) 30 min (d) 40 min	97
4.15	Conductivity with time laser irradiation nanoparticle analysis	97
4.16	XRD analysis of the (a) RHA and (b) RHACCl (SiO <sub>2</sub> ) NPs	100
4.17	FTIR spectra of the (a) RHA and (b) RHACCl (SiO <sub>2</sub> )	101
4.18	FESEM of (a, b) RHA and (c, d) RHACCl (SiO <sub>2</sub> )	101
4.19	EDX spectrums of the (a) RHA, (b) RHACCl (silica) nanoparticle	102
4.20	UV-Vis spectra of RHACCl	103
4.21	TGA curve of RHACCl (SiO <sub>2</sub> ) nanoparticles showing mass loss	104
4.22	XRD analysis of RHACCl, PANI-RHACCl (SiO <sub>2</sub> ) (1:0.5), PANIRHACCl (SiO <sub>2</sub> ) (2:0.5), PANI/RHACCl (SiO <sub>2</sub> ) (3:0.5) nanocomposite shows the variation of polyaniline	105
4.23	FTIR spectra of the (a) RHACCl, (b) Polyaniline (PANI), (c) PANI-RHACCl (SiO <sub>2</sub> ) (1:0.5), (d) PANI-RHACCl (SiO <sub>2</sub> )(2:0.5), PANI/RHACCl (SiO <sub>2</sub> ) (3:0.5) nanocompsite	107
4.24	FESEM images of the (a, b) PANI-RHACCl (SiO <sub>2</sub> ) (1:0.5), (c, d) PANI-RHACCl (SiO <sub>2</sub> ) (2:0.5), (e, f) PANI/RHACCl (SiO <sub>2</sub> ) (3:0.5) nanocompsite	109
4.25	EDX spectrums of the (a) PANI-RHACCl (SiO <sub>2</sub> ) (1:0.5), (b) PANIRHACCl (SiO <sub>2</sub> ) (2:0.5), (c) PANI/RHACCl (SiO <sub>2</sub> ) (3:0.5) nanocompsite	111
4.26	UV-vis analysis of UV-Vis spectra of PANI/ RHACCl (SiO <sub>2</sub> ) (a) 1:0.5), (b) PANI-RHACCl (SiO <sub>2</sub> )(2:0.5), (c)	



	PANI/RHACCl (SiO <sub>2</sub> ) (3:0.5) nanocomposite	111
4.27	Energy band gap for PANI-RHACCl (a) 1:0.5 (b) 2:0.5 (c) 3:0.54	113
4.28	TGA and DTG curve of PANI /RHACCl (SiO <sub>2</sub> ) nanocomposites showing mass loss and thermal stability of nanocomposites	115
4.29	XRD pattern for irradiated PANI/RHACCl(SiO <sub>2</sub> ) (1:0.5) nanocomposite with Diode laser at (a) 10 min (b) 20 min (c) 30 min (d) 40 min	116
4.30	XRD pattern for irradiated PANI/RHACCl (2:0.5) nanocomposite with Diode laser at (a) 10 min (b) 20 min (c) 30 min (d) 40 min	118
4.31	XRD pattern for irradiated PANI/RHACCl(SiO <sub>2</sub> ) (3:0.5) nanocomposite with diode laser at (a) 10 min (b) 20 min (c) 30 min (d) 40 min (e)	119
4.32	FTIR spectra for irradiated PANI/ARHA (1:0.5) nanocomposite with Diode laser at (a) 0 min (b) 10 min (c) 20 min (d) 30 min (e) 40 min	120
4.33	FTIR spectra for irradiated PANI/ARHA (2:0.5) nanocomposite with Diode laser at (a) 0 min (b) 10 min (c) 20 min (d) 30 min (e) 40 min	122
4.34	TIR spectra for irradiated PANI/ARHA (3:0.5) nanocomposite with Diode laser at (a) 0 min (b) 10 min (c) 20 min (d) 30 min (e) 40 min	123
4.35	FESEM for irradiated PANI/RHACCl (1:0.5) nanocomposite with Diode laser at (a) 10 min (b) 20 min (c) 30 min (d) 40 min, (100K magnification)	124
4.36	FESEM for irradiated PANI/RHACCl (2:0.5) nanocomposite with Diode laser at (a) 10 min (b) 20 min (c) 30 min (d) 40 min (100K magnification)	125
4.37	EDX spectrum of irradiated PANI/RHACCl (1:0.5) nanocomposite with Diode laser at (a) 10 min (b) 20 min (c) 30 min (d) 40 min	126

4.38	EDX Elemental composition for irradiated PANI/RHACCl (1:0.5) nanocomposites	127
4.39	EDX spectrum of irradiated PANI/RHACCl (2:0.5) nanocomposite with Diode laser at (a) 10 min (b) 20 min (c) 30 min (d) 40 min	128
4.40	EDX Elemental composition for irradiated PANI/RHACCl (3:0.5) nanocomposites	128



## LIST OF SYMBOLS AND ABBREVIATIONS

%	-	Percentage
$\pi$	-	Pi
$\Omega$	-	Ohms
=	-	Equal to
$\leq$	-	Less than equal to
$^{\circ}$	-	Degree
$^{\circ}\text{C}$	-	Degree celsius
3D	-	3 dimensional
$\text{\AA}$	-	Angstrom
$\beta$	-	Beta
$\theta$	-	Angle
$\sigma$	-	Conductivity
A	-	Area
(a-C)	-	amorphous carbon
at%	-	Atomic percentage
C	-	Capacitance of dielectric constant
cm	-	The capacitance
CPE	-	Constant phase element
Cu	-	Copper
D	-	Diameter
DC	-	Direct current
EDX	-	Energy Dispersive X-ray
FESEM	-	Field emission scanning electron microscope
FTIR	-	Fourier transforms infrared
$f$	-	Frequency
FWHM	-	Full width at half maximum
g	-	Graem
H	-	Hour

Hz	-	Hertz
IS	-	Impedance spectroscopy
IT	-	Information Technology
k	-	Rate constant ( $\text{min}^{-1}$ )
K	-	Kelvin
Kg	-	Kilogram
L	-	Liter
M	-	Mass of tested sample (g)
m	-	Meter
mg	-	Milgram
MHz	-	Megahertz
mL	-	Mililiter
NCs	-	Nanocomposite
NPs	-	Nanoparticles
$^{\circ}\text{C}/\text{min}$	-	Degree Celsius per minute
PANI	-	Polyaniline
PVA	-	Polyvinyl Alcohol
RH	-	Rice Husk
RHA	-	Rice Husk Ash
RHACCl	-	Rice Husk Ash(Silica)
RT	-	Room temperature
s	-	Second
S	-	Siemens
s.g	-	Space group
SC	-	Scandium
Si	-	Silicon
$\text{SiO}_2$	-	Silica
T	-	Thickness
XRD	-	X-Ray diffraction

**LIST OF APPENDICES**

APPENDIX	TITLE	PAGE
A	Calculation of mass for each compound	151
B	List of powder diffraction (xrd) file (pdf) before and after laser	153
C	List of publications	157
D	List of equipment and chemical used analytical instruments	158
E	Analytical instruments	159



PTTA UTHM  
PERPUSTAKAAN TUNKU TUN AMINAH

## CHAPTER 1

### INTRODUCTION

#### 1.1 Background of the study

A nanomaterial is the cornerstones of nanoscience, nanotechnology, and nanoscale that have contributed to a vast and multidisciplinary research area that is increasing dramatically worldwide in the last few years (Liu *et al.*, 2017). The future for advanced nanomaterials is faced by four fundamental problems in nanotechnology, given in (Manikandan *et al.*, 2020). The first is the precise fabrication of nanomaterials (Wang *et al.*, 2020), the second one is the precise positioning of materials and repeated at ease for various manipulation strategies (James *et al.*, 2019). Thirdly, nanomaterials are interconnecting for functional devices fabrication (Karim *et al.*, 2019), and finally is the mass production of nano-based devices for commercialization (Selvaraju, Sundaresan, & Dharmar, 2019). Due to its benefits of being easily synthesized, being environmentally stable in air and water, and possessing high conductivity, polyaniline is an interesting conducting polymer (Zhao *et al.*, 2019). However, polyaniline (PANI) has its own significance in the materials for electrochemical capacitors, with a higher precise capacity, a higher energy capacitance, and a faster energy density than typical capacitors (Roy *et al.*, 2018; Mazzeu *et al.*, 2017).

It comprises one silicon atom and 2 atoms of oxygen that yields  $\text{SiO}_2$  that contributes to today's technology revolution that is applied to computer plastics and provides the raw material for silicon chips (Roosz *et al.*, 2017b). The Silica is widely used and exists uncommonly in amorphous state and often in crystalline state (Roosz *et al.*, 2017a).

It is observed that 95% of the rice husk is made up of  $\text{SiO}_2$ , obtained at 800 °C (Liu *et al.*, 2020) known as silica RHACCl (Adam *et al.*, 2010). Several metal oxide nanoparticles (NPs) are attached to the conductive polymers to create nanocomposites (NCs) applied in the manufacture from PANI/ RHACCl (silica) nanocomposites in the presence of triethylamine ( $\text{Et}_3\text{N}$ ) that has been synthesized by  $\text{Et}_3\text{N}$ . It is a new method for synthesizing silica (RHACCl) with polyaniline to form a nanocomposite used sol-gel.

The term polymer is widely used today to indicate the sense of "plastic" or "resin" (Roosz *et al.*, 2017a). A polymer is a chemical compound where long repeating chains, molecules are bound together. Such polymers have specific properties and can be modified according to their intended purpose (Roosz *et al.*, 2017b). The backbone description of the polymer chain is categorized into two (inorganic and organic) polymers. The polymer whose backbone chain is composed of carbon atoms is referred to as an organic polymer, typically hydrogen, hydrogen, oxygen, oxygen, nitrogen, nitrogen, etc.) (Liu *et al.*, 2013). On the other hand, most plastic polymers are organic, with no carbon atom in the chain backbone. However, glass is referred to as inorganic polymers, while silicone rubber is an exception (Kim & Kang, 2013). It is also possible to classify organic polymers as conductors polyaniline (PANI) (Ruiz-Pérez *et al.*, 2020), polypyrrole (PPy), (Abolghasemi *et al.*, 2018), polythiophene, polyimide (Chen *et al.*, 2017) and non-conductive, such as polystyrene (Balint *et al.*, 2014).

One of the reasons for this study is the ease of linking polyaniline to the modified silica obtained from rice husks due to the ease of removing the chlorine atom from the modified silica and linking the polyaniline in place to form a nanocomposite. This is a new method for preparing nanocomposites. In addition to the electrical properties.(Chen *et al.*, 2017).

## 1.2 Polyaniline

Due to its unique characteristics like low cost, lightweight, environmental, height stability, dielectric properties, workability, resilience, thermal stability, oxidative stability, and sensitivity, polyaniline is recognized as being the most attractive conducting polyaniline (Chen *et al.*, 2017). Ease from synthesis, good electrical conductivity, including interesting redox properties is connected with the chain

nitrogen (Song & Choi, 2013). The Polyaniline (PANI) uniqueness defines its great impact to be used in technology such as material involved in electromagnetic shielding, anti-corrosion protection, sensor devices, anti-fouling protection, including antistatic materials, among others (Caldas *et al.*, 2017). The reversible tunable redox features define the electrical conductivity control with a great range over protonation and charge-transfer doping that are distinctive features of PANI compared to other conjugated polymers (Soares, 2018).

### 1.3 Silica Nanoparticles (SiO<sub>2</sub>NPs)

Silica is a cluster of minerals made of silicon and oxygen i.e. two copious elements in the earth layer. Silica formula of SiO<sub>2</sub> that is pragmatic on earth since quartz which is greater than 10% by mass from the earth layer. The rapidly evolving nanotechnology fields and the versatile processes used to generate, manipulate and utilize nanomaterial, which is recently being emphasized leading to a promising era in bio-analytical, biotechnological, and biomedical applications (Ionita *et al.*, 2008). A round of work activity has focused on the preparation and functionalization of the nanoparticles (NPs), owing to their unique, chemical (catalytic), physical (structural, magnetic, optical and electronic), including electrochemical characteristics (Lee *et al.*, 2010).

Rice husk (RH) is fibre based on cellulose, ideal for recycling. Rice is grown in over 80 nations. The world's annual rice production ranges from four hundred to fifty-five hundred million metric tons, more than 10 % than rice husk because of its high silica content. Using this free raw material makes economic sense (Adam *et al.*, 2012). Rice husk consists of 38% cellulose, 20% ash, 18% pentose, 22% lignin, 2% water, and other organic ingredients. The burning of rice husk in nature has caused environmental pollution. Consequently, serious steps have been taken to burn the peels at reasonable heat and controlled pressure and these wastes are used to produce scientific materials that have commercial use within the recycling of materials (Shen, 2017). It is worth to note that rice husk charred produced the white ash, that porous silica by high specific surface area and this can be accumulated. This RHA comprises of over 95% silica (Wang *et al.*, 2019).



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**APPENDIX C:****LIST OF PUBLICATIONS**

- i. Mezan, S. O., Jabbar, A. H., Hamzah, M. Q., Tuama, A. N., Hasan, N. N., & Agam, M. A. (2018). Synthesis and Characterization of Zinc Sulphide (ZnS) Thin Film Nanoparticle for Optical Properties. *Journal of Global Pharma Technology*, 10(07), 369-373.
- ii. Mezan, S. O., Hello, K. M., Jabbar, A. H., Hamzah, M. Q., Tuama, A., Roslan, M. S., & Agam, M. A. (2020). A review on synthesis of conducting with polyaniline rice husk ash silica nanocomposites and application. *International Journal of Psychosocial Rehabilitation*, 24(03). *International Journal of Psychosocial Rehabilitation*, 24(03).
- iii. Mezan, S. O., Hello, K. M., Jabbar, A. H., Hamzah, M. Q., Tuama, A. N., Roslan, M. S., & Agam, M. A. (2020). Synthesis and Characterization of Enhanced Polyaniline Nanoparticles by Oxidizing Polymerization. *Solid State Technology*, 63(1), 256-266.
- iv. Mezan, S. O., Jabbar, A. H., Hamzah, M. Q., Tuama, A. N., Hasan, N. N., Roslan, M. S., & Agam, M. A. (2019, August). Synthesis, characterization, and properties of polystyrene/SiO<sub>2</sub> nanocomposite via sol-gel process. *In AIP Conference Proceedings (Vol. 2151, No. 1, p. 020034)*.
- v. Mezan, S. O., Al Absi, S. M., Jabbar, A. H., Roslan, M. S., & Agam, M. A. (2021). Synthesis and characterization of enhanced silica nanoparticle (SiO<sub>2</sub>) prepared from rice husk ash immobilized of 3-(chloropropyl) triethoxysilane. *Materials Today: Proceedings*.

**APPENDIX D****LIST OF EQUIPMENT AND CHEMICAL USED**

- i. aniline monomer  $C_6H_7N$ .
- ii. Ammonium peroxydisulfate (APS)  $(NH_4)_2S_2O_8$ .
- iii. Saturated hydrochloric acid HCl.
- iv. 3-(chloropropyl) triethoxysilane  $C_9H_{23}NO_3Si$
- v. Sodium hydroxide NaOH
- vi. Rice Husk Ash RHA.
- vii. Nitric acid  $HNO_3$
- viii. Aston  $CH_3COCH_3$
- ix. Deionized water (DI)
- x. Polyaniline  $([C_6H_4NH]_2[C_6H_4N]_2)_n$
- xi. 50- 60 °C Conventional Oven.
- xii. Air dryer cupboard.
- xiii. Toluene  $C_7H_8$  or  $C_6H_5CH_3$
- xiv. Ammonium hydroxide  $NH_4OH$ .
- xv. Diode Laser Generator Power 1.265 W, 450 nm wavelength to expose nanocomposite of PANI and PANI/RHACCl ( $SiO_2$ ) nanoparticles that have already been prepared.
- xvi. Commonly used analytical glass devices in a laboratory experimental tests include 10 mL and 25 mL volume measurement glasses; 5mL, 10 mL, 50 ML, and 700 mL fixed volume volumetric flasks; beaker glasses; 25 mL volume microbursts; and volumetric pipettes.
- xvii. Micro 200 Hettich Zentrifugen D-78532 Mini Ultracentrifuge Computer fabricated in Germany.
- xviii. Vortex homogenizer machine, Uzusio VTX-3000L, Made in Japan.
- xix. Small Erlenmeyer flask
- xx. Large dish of ice
- xxi. Stirrer hotplate
- Xxii. 1" stir bar

**APPENDIX E**

## ANALYTICAL INSTRUMENTS

- i. (0.0001)-gram sensitivity analytical balance.
- ii. Magnetic Stirrer Hotplate Machine: IKA RCT basic safety control mode, temperature range up to 310 oC, motor rotation ranges up to 1500 rpm.
- ii. The furnace at range 1200 °C.
- iii. Driven Bench-Top Ultrasonic Cleaner Micro Process Made by
- iv. Seoul, Korea, Hwashin Technology Co. Model: Powersonic 405, Input/output electrical energy: AC 230 volt, 50 Hz 350 watt for samples to agitate/irradiate.



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## VITA

The author was born in 3 April 1975 and completed his secondary school in Ibin Hian high school for boys, Al Muthanna, Iraq. He then joined the bachelor's degree in 1996 at the University of Qadisiyah - College of Education - Department of Physics. He has worked as a physics teacher in high schools since 2001 and until now.

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