

APPLICATION OF ELECTROKINETIC REMEDIATION ON CONTAMINATED  
ALLUVIAL SOILS IN SRI GADING, BATU PAHAT, JOHOR

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*Ilmu adalah panduan  
Harta adalah alat  
Kedua-duanya adalah untuk menegakkan kebenaran  
Bukan untuk mencari kepentingan  
Dan jauh sekali untuk tujuan kesombongan dan kebanggaan*



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## ABSTRACT

The research described in this thesis is a comprehensive account to remove excessive heavy metals and radionuclides contained in the most radiation-contaminated soil by using the most optimised electrokinetic remediation technique. This novel study that saved time and costs based on electrokinetic optimisation was the ultimate vision to be discovered on this thesis. In addition, reduction in terrestrial gamma radiation dose rate (TGRD) by reducing soil pH level during electrokinetic processes was somewhat new and exciting in this study. The TGRD and soil pH of various soil type within Batu Pahat and Kluang districts were measured. Soil with the highest TGRD was selected to be remediated by various types of electrokinetics setup, to list out the best electrokinetic parameters. Results obtained from these setup led to the development of the most optimised electrokinetic setup, especially in remediating radionuclides. Data analysis showed that Holyrood-Lunas soil species taken from Sri Gading recorded the highest TGRD of  $476.8 \pm 41.9$  nGy h<sup>-1</sup>. The pre-electrokinetic examination by X-ray fluorescence (XRF) indicated that 10 heavy metals contained in the sample were significantly higher than the Environmental Protection Agency (EPA) contamination indices limit, in decreasing order, were Th>As>Sb>Sn>U>Cr>Pb>Ni>V>Zr. The post-electrokinetic analysis suggested the most optimised parameters for Th and U removals were 3.0 hours remediation time, 30 volts electrode voltage, 22 cm electrode spacing, plate-shaped electrodes by 8×8 cm and in 1-D electrodes configuration. Applications of these optimised parameters by high-pure germanium (HPGe) analysis gave a very low <sup>232</sup>Th and <sup>238</sup>U removals at  $2.74 \pm 23.78$  and  $0.23 \pm 2.64$  ppms, respectively, while for the pilot scale electrokinetic setup shows average removals at  $1.47 \pm 93.09$  and  $0.21 \pm 4.11$  ppms, respectively. The relationships between <sup>232</sup>Th and <sup>238</sup>U concentrations with soil pH changes supported the initial hypotheses that reducing TGRD can be obtained by reducing soil pH level during the electrokinetic remediation process. Thus it can be concluded that the findings of this research can contribute towards the best soil decontamination.

## ABSTRAK

Penyelidikan dalam tesis ini adalah satu kajian menyeluruh untuk menyingkirkan lebihan logam berat dan radionuklid dalam tanah paling tercemar secara radiasi menggunakan teknik pemulihan elektrokinetik paling optimum. Kajian terbaru ini yang menjimatkan masa dan kos berdasarkan pengoptimuman elektrokinetik adalah visi muktamad di dalam tesis ini. Tambahan pula pengurangan kadar dos sinar gamma terrestrial (TGRD) dengan cara mengurangkan pH tanah semasa elektrokinetik adalah agak baru dan menarik dalam kajian ini. Pelbagai jenis tanah sekitar Batu Pahat dan Kluang telah diukur TGRD dan pH tanah. Tanah dengan TGRD tertinggi telah dipilih untuk dirawat melalui pelbagai jenis persediaan elektrokinetik, bagi menyenaraikan parameter elektrokinetik terbaik. Keputusan persediaan elektrokinetik ini menjurus kepada pembangunan persediaan elektrokinetik yang paling optimum, terutamanya dalam memulihkan radionuklid. Analisa data mendapati spesies tanah Holyrood-Lunas yang diambil dari Sri Gading mencatatkan TGRD tertinggi iaitu  $476.8 \pm 41.9$  nGy jam<sup>-1</sup>. Pemeriksaan pra-elektrokinetik melalui analisa X-ray pendarflouran (XRF) mendapati 10 logam berat dalam sampel tersebut melampaui had indeks pencemaran Agensi Perlindungan Alam Sekitar (EPA), dalam susunan menurun: Th>As>Sb>Sn>U>Cr>Pb>Ni>V>Zr. Analisa pasca-elektrokinetik mencadangkan parameter paling optimum bagi nyah-singkir Th dan U adalah 3.0 jam masa rawatan, 30 volt beza keupayaan elektrod, 22 cm jarak elektrod, elektrod berbentuk plat bersaiz 8×8 cm dan konfigurasi elektrod 1-D. Aplikasi parameter optimum menerusi analisa germanium berketulenan tinggi (HPGe) memberikan nyah-singkir  $^{232}\text{Th}$  dan  $^{238}\text{U}$  yang sangat rendah iaitu  $2.74 \pm 23.78$  dan  $0.23 \pm 2.64$  ppm, masing-masingnya, sedangkan elektrokinetik pada skala perintis menunjukkan purata nyah-singkir  $1.47 \pm 93.09$  dan  $0.21 \pm 4.11$  ppm, masing-masingnya. Hubungan di antara kepekatan  $^{232}\text{Th}$  dan  $^{238}\text{U}$  dengan perubahan pH tanah menyokong hipotesis awal iaitu pengurangan TGRD boleh dilakukan dengan cara mengurangkan aras pH tanah semasa proses pemulihan elektrokinetik. Oleh itu adalah dirumuskan dapatan-dapatan di dalam penyelidikan ini boleh menyumbang kepada nyah-kontaminasi tanah yang terbaik.

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## LIST OF SYMBOLS

$^{40}\text{K}$ (or K-40)	-	Naturally occurring potassium isotope with $A = 40$
$^{226}\text{Ra}$ (or Ra-226)	-	Naturally occurring radium isotope with $A = 226$
$^{220}\text{Rn}$ (or Rn-220)	-	Naturally occurring radon isotope with $A = 220$
$^{222}\text{Ra}$ (or Ra-222)	-	Naturally occurring radium isotope with $A = 222$
$^{232}\text{Th}$ (or Th-232)	-	Naturally occurring thorium isotope with $A = 232$
$^{238}\text{U}$ (or U-238)	-	Naturally occurring uranium isotope with $A = 238$
$\alpha$	-	Alpha particle
$\beta$	-	Beta particle
$\Delta_m$	-	The absolute difference between $c_m$ and $c_{CRM}$
$\gamma$	-	Gamma radiation
$\eta$	-	Fluid viscosity
$\tau$	-	Tortuosity
$\zeta$	-	Zeta potential
$A_K$	-	Activity mass concentration of $^{40}\text{K}$ (in $\text{Bq}\cdot\text{kg}^{-1}$ )
$A_{\text{Ra}}$	-	Activity mass concentration of $^{226}\text{Ra}$ (in $\text{Bq}\cdot\text{kg}^{-1}$ )
$A_s$	-	Activity mass concentration (in $\text{Bq}\cdot\text{kg}^{-1}$ )
$A_{\text{Th}}$	-	Activity mass concentration of $^{232}\text{Th}$ (in $\text{Bq}\cdot\text{kg}^{-1}$ )
Al	-	Aluminium
As	-	Arsenic
At%	-	Soil element concentration in atomic percentage
Ba	-	Barium
$\text{Bq kg}^{-1}$	-	Becquerel per kilogram
$c_m$	-	The average of measured heavy metal concentrations
CRM	-	Certified values

$C$	-	Coulomb
$C_D$	-	Post-electrokinetic deposition concentration
$C_F$	-	Dose conversion factor (absorbed dose rate in air per unit of activity concentration in soil)
$C_R$	-	Post-electrokinetic residual concentration
$C_M$ background	-	The basic value of heavy metals (in ppm).
$C_M$ sample	-	The concentration of heavy metals in contaminated soils (in ppm)
$C_{\text{smp}}$	-	The concentrations of the soil sample.
$C_{\text{std}}$	-	The concentrations of the standard sample
$CF$	-	Pollution factors
Ca	-	Calcium
Cd	-	Cadmium
Ce	-	Cerium
Co	-	Cobalt
Cu	-	Copper
Cr	-	Chromium
Cs	-	Cesium
d	-	Day
$D$	-	Dielectric constant
$e^-$	-	Electron
$E$	-	Electrical field strength (in V/m)
$E_z$	-	Applied voltage gradient
$E_{\text{eff}}$	-	Annual effective dose (in $\text{mSv}\cdot\text{y}^{-1}$ )
$EF$	-	Enrichment Factor
$F$	-	Faraday constant (96.487 C/mol electrons)
$F$	-	Absorbed-to-effective dose conversion factor (in $\text{mSv}$ per $\text{nGy}$ )
Fe	-	Iron
Ga	-	Gallium
h	-	Hour
$H_2$	-	Chemical form of hydrogen gas
$H_2O$	-	Chemical form of water

$H^+$	-	Hydrogen ion
$H_{\text{ext}}$	-	External hazard index
$H_{\text{int}}$	-	Internal hazard index
$\text{Hg}$	-	Mercury
$I_{\text{geo}}$	-	Index of Geo-accumulation
$k$	-	coverage factor
$\text{K}$	-	Potassium
$\text{La}$	-	Lanthanum
mass%	-	Soil element concentration in mass percentage
$\text{mSv}\cdot\text{y}^{-1}$	-	Millisievert per year
$\text{mGy}$	-	Milligray
$\text{Mg}$	-	Magnesium
$\text{Mn}$	-	Manganese
$\text{Mo}$	-	Molybdenum
$n$	-	Number of measurement data
$N_{\text{smp-bg}}$	-	The net counts of the photopeak area (after the background subtraction) of the soil sample
$N_{\text{std-bg}},$	-	The net counts of the photopeak area (after the background subtraction) of the standard sample
$\text{Na}$	-	Sodium
$\text{nGy}\cdot\text{h}^{-1}$	-	Nanogray per hour
$\text{Nb}$	-	Niobium
$N$	-	Soil porosity
$\text{Ni}$	-	Nickel
$\text{O}_2$	-	Chemical form of oxygen gas
$\text{OH}^-$	-	Hydroxide ion
$\text{Pb}$	-	Lead
$\text{pH}$	-	Hydrogen ion concentration
$PLI$	-	Pollution Load Index
$\text{Ra}_{\text{eq}}$	-	Radium equivalent activity
$\text{Rb}$	-	Rubidium
$RSD$	-	Relative standard deviation
$s$	-	Standard deviation

## REFERENCES

- Ab Razak, N. H., Praveena, S. M., Aris, A. Z., & Hashim, Z. (2015). Drinking water studies: A review on heavy metal, application of biomarker and health risk assessment (a special focus in Malaysia). *Journal of Epidemiology and Global Health*, 5(4), 297–310. <https://doi.org/10.1016/j.jegh.2015.04.003>
- Abdul Rahman, A. T., & Ramli, A. T. (2007). Radioactivity levels of 238U and 232Th, the alpha and beta activities and associated dose rates from surface soil in Ulu Tiram, Malaysia. *Journal of Radioanalytical and Nuclear Chemistry*, 273(3), 653–657. <https://doi.org/10.1007/s10967-007-0926-2>
- Abdullah, M. P., Saadati, N., Wan Mohd Khalik, W. M. A., & Zakaria, Z. (2015). Pattern recognition of the presence and distribution of organochlorine pesticides in sediment of Cameron Highlands, Malaysia. Retrieved April 7, 2019, from [http://www.ukm.my/mjas/v19\\_n4/html/19\\_4\\_6.html](http://www.ukm.my/mjas/v19_n4/html/19_4_6.html)
- Abdullahi, S., Ismail, A. F., & Samat, S. (2019). Determination of indoor doses and excess lifetime cancer risks caused by building materials containing natural radionuclides in Malaysia. *Nuclear Engineering and Technology*, 51(1), 325–336. <https://doi.org/10.1016/j.net.2018.09.017>
- Absar, N., Rahman, M. M., Kamal, M., & Siddique, N. (2014). Natural and anthropogenic radioactivity levels and the associated radiation hazard in the soil of Oodalia Tea Estate in the hilly region of Fatickchari in Chittagong , Bangladesh, (July), 1075–1080. <https://doi.org/10.1093/jrr/rru054>
- Acar, Y. B., & Alshawabkeh, A. (1993). Principles of Electrokinetic Remediation. *Environmental Science and Technology*, 27(13), 2638–2647.
- Acar, Y. B., Gale, R. J., Alshawabkeh, A. N., Marks, R. E., Puppala, S., Bricka, M., & Parker, R. (1995). Electrokinetic remediation: Basics and technology status. *Journal of Hazardous Materials*, 40(2), 117–137. [https://doi.org/10.1016/0304-3894\(94\)00066-P](https://doi.org/10.1016/0304-3894(94)00066-P)
- Adham, M. I., Shirazi, S. M., Othman, F., Zardari, N. H., Yusop, Z., & Ismail, Z.

- (2015). Soil Erodibility for Water Pollution Management of Melaka Watershed in Peninsular Malaysia. *Environment Asia*, 8(2), 63–69.
- Adriano, D. C. (1986). *Trace Elements in the Terrestrial Environment*. New York: Springer-Verlag.
- Agamuthu, P., & Fauziah, S. H. (2010). Heavy metal pollution in landfill environment: A Malaysian case study. *4th International Conference on Bioinformatics and Biomedical Engineering (ICBBE)*.
- Agar, O., Boztosun, I., Korkmaz, M. E., & Ozmen, S. F. (2014). Measurement of radioactivity levels and assessment of radioactivity hazards of soil samples in Karaman, Turkey. *Radiation Protection Dosimetry*, 162(4), 630–637.
- Agnew, K., Cundy, A. B., Hopkinson, L., Croudace, I. W., Warwick, P. E., & Purdie, P. (2011). Electrokinetic remediation of plutonium-contaminated nuclear site wastes: Results from a pilot-scale on-site trial. *Journal of Hazardous Materials*, 186(2–3), 1405–1414. <https://doi.org/10.1016/j.jhazmat.2010.12.016>
- Ahmad, N., Jaafar, M. S., Bakhsh, M., & Rahim, M. (2015). An overview on measurements of natural radioactivity in Malaysia. *Journal of Radiation Research and Applied Sciences*, 8, 136–141. <https://doi.org/10.1016/j.jrras.2014.12.008>
- Akinci, A., & Artir, R. (2008). Characterization of trace elements and radionuclides and their risk assessment in red mud. *Material Characterization*, 59, 417–421. <https://doi.org/10.1016/j.matchar.2007.02.008>
- Akoglu, H. (2018). User's guide to correlation coefficients. *Turkish Journal of Emergency Medicine*, 18(3), 91–93. <https://doi.org/10.1016/j.tjem.2018.08.001>
- Aliyu, A. S., & Ramli, A. T. (2015). The world's high background natural radiation areas (HBNRAs) revisited: A broad overview of the dosimetric, epidemiological and radiobiological issues. *Radiation Measurements*, 73, 51–59. <https://doi.org/10.1016/j.radmeas.2015.01.007>
- Almayahi, B. A., Tajuddin, A. A., & Jaafar, M. S. (2014). Measurements of natural radionuclides in human teeth and animal bones as markers of radiation exposure from soil in the Northern Malaysian Peninsula. *Radiation Physics and Chemistry*, 97, 56–67. <https://doi.org/10.1016/j.radphyschem.2013.10.016>
- Alnour, I. A., Wagiran, H., Ibrahim, N., Laili, Z., Omar, M., Hamzah, S., & Idi, B. Y. (2012). Natural radioactivity measurements in the granite rock of quarry sites, Johor, Malaysia. *Radiation Physics and Chemistry*, 81(12), 1842–1847.

- <https://doi.org/10.1016/j.radphyschem.2012.08.005>
- Alshawabkeh, A. N., Gale, R. J., Ozsu-Acar, E., & Bricka, R. M. (1999). Optimization of 2-D Electrode Configuration for Electrokinetic Remediation. *Soil and Sediment Contamination*, 8(6), 617–635. <https://doi.org/10.1080/1058833991339504>
- American Galvanizers Association. (2013). *Soil Environment*. Retrieved from [www.galvanizeit.org](http://www.galvanizeit.org)
- Antoniadis, V., Shaheen, S. M., Levizou, E., Shahid, M., Niazi, N. K., Vithanage, M., ... Rinklebe, J. (2019). A critical prospective analysis of the potential toxicity of trace element regulation limits in soils worldwide: Are they protective concerning health risk assessment? - A review. *Environment International*, 127(November 2018), 819–847. <https://doi.org/10.1016/j.envint.2019.03.039>
- Anuar, A. R., & Che Fauziah, I. (2008). Background concentrations of trace elements in soils for risk assessment. In *Proceeding of the Soils Science Conference of Malaysia 2008: Sustaining Soil Ecosystems with Emphasis on Coastal Soils: Perak (Malaysia)* (pp. 67–68). Perak, Malaysia: Malaysia Society of Soil Science. Retrieved from agris.upm.edu.my:8080
- Aoyagi, C., & Ganelli, G. (2015). Asia's quest for inclusive growth revisited. *Journal of Asian Economics*, 40, 29–46. <https://doi.org/10.1016/j.asieco.2015.06.005>
- AQCS Intercomparison Runs Reference Material. (2010). Radionuclides and some Trace Elements in Soil. Vienna, Austria: Analytical Quality Control Services, IAEA.
- Arai, Y. (2010). Arsenic and Antimony. In P. S. Hooda (Ed.), *Trace elements in soils* (1st ed., p. 384). London: Blackwell Publishing Ltd.
- Babai, K. S., Poongothai, S., & Punniyakotti, J. (2013). Determination of environmental radioactivity ( $^{238}\text{U}$ ,  $^{232}\text{Th}$  and  $^{40}\text{K}$ ) and indoor natural background radiation level in Chennai city (Tamilnadu State), India. *Radiation Protection Dosimetry*, 153(4), 457–466. <https://doi.org/10.1093/rpd/ncs121>
- Baek, K., & Yang, J.-S. (2009). Electrokinetic Removal of Nitrate and Fluoride. In K. R. Reddy & C. Cameselle (Eds.), *Electrochemical Remediation Technologies for Polluted Soils, Sediments and Groundwater* (1st ed., p. 141). Hoboken, New Jersey: Wiley.
- Bani Baker, M., Elektorowicz, M., & Hanna, A. (2018). Electrokinetic nondestructive in-situ technique for rehabilitation of liners damaged by fuels. *Journal of*

- Hazardous Materials*, 359, 510–515. <https://doi.org/10.1016/j.jhazmat>. 2018.07. 113
- Baranowski, R., Rybak, A., & Baranowska, I. (2002). Speciation Analysis of Elements in Soil Samples by XRF. *Polish Journal of Environmental Studies*, 11(5), 473–482.
- Bártová, H., Kučera, J., Musílek, L., Trojek, T., & Gregorová, E. (2017). Determination of U, Th and K in bricks by gamma-ray spectrometry, X-ray fluorescence analysis and neutron activation analysis. *Radiation Physics and Chemistry*, 140(September 2016), 161–166. <https://doi.org/10.1016/j.radphyschem.2017.01.035>
- Belivermis, M., Kılıç, Ö., Çotuk, Y., & Topcuoğlu, S. (2010). The effects of physicochemical properties on gamma emitting natural radionuclide levels in the soil profile of Istanbul. *Environmental Monitoring and Assessment*, 163(1–4), 15–26. <https://doi.org/10.1007/s10661-009-0812-1>
- Biesinger, M. C., Lau, L. W. M., Gerson, A. R., & Smart, R. S. C. (2010). Resolving surface chemical states in XPS analysis of first row transition metals, oxides and hydroxides: Sc, Ti, V, Cu and Zn. *Applied Surface Science*, 257(3), 887–898. <https://doi.org/10.1016/j.apsusc.2010.07.086>
- Biesinger, M. C., Payne, B. P., Grosvenor, A. P., Lau, L. W. M., Gerson, A. R., & Smart, R. S. C. (2011). Resolving surface chemical states in XPS analysis of first row transition metals, oxides and hydroxides: Cr, Mn, Fe, Co and Ni. *Applied Surface Science*, 257(7), 2717–2730. <https://doi.org/10.1016/j.apsusc.2010.10.051>
- Bradford, G. R., Change, A. C., Page, A. L., Bakhtar, D., Frampton, J. A., & Wright, H. (1996). *Background Concentrations of Trace and Major Elements in California Soils*. Retrieved from <http://www.envisci.ucr.edu/downloads/chang/kearney/kearneytext.html>
- Brady, J. E. (1990). *General chemistry : principles and structure* (5th ed.). New York: Wiley. Retrieved from <https://uthm.on.worldcat.org/external-search>
- Bruker. (2011). *Geo-Quant T User Manual*. Karlsruhe, Germany: Bruker AXS GmbH.
- Bruker AXS. (2006). *Introduction to X-ray Fluorescence (XRF)*. Karlsruhe, West Germany: Bruker AXS GmbH.
- Bruker AXS GmbH. (2014). *Minerals Solutions*. Karsruhe, Germany: Bruker AXS GmbH.

- Buhrke, V. E., Jenkins, R., & Smith, D. K. (1998). *A Practical Guide For The Preparation Of Specimens For X-Ray Flourescence And X-Ray Diffraction Analysis*. New York: Wiley-VCH.
- Cameselle, C., Gouveia, S., Eddine, D., & Belhadj, B. (2013). Advances in Electrokinetic Remediation for the Removal of Organic Contaminants in Soils. In M. N. Rashed (Ed.), *Organic Pollutants - Monitoring, Risk and Treatment* (1st ed., pp. 209–229). InTech. <https://doi.org/10.5772/54334>
- Cappelen, T., Unhjem, J. F., & Watson, R. (2009). Use of 131-Iodine and the Risk of Radiation Exposure. In *Comprehensive Handbook of Iodine* (pp. 965–978). Elsevier. <https://doi.org/10.1016/B978-0-12-374135-6.00100-X>
- Cauwenberghe, L. Van. (1997). *Electrokinetics: Technology Overview Report. Ground-Water Remediation Technologies Analysis Center*. Pittsburg.
- Chakraborty, S. R., & Alam, M. K. (2014). Countrywide radiation dose in different locations, dwellings and free spaces of Bangladesh. *Radiation Protection Dosimetry*, 162(4), 638–648.
- Chandrasekaran, A., Rajalakshmi, A., Ravisankar, R., Vijayagopal, P., & Venkatraman, B. (2015). Measurements of Natural Gamma Radiations and Effects of Physico-Chemical Properties in Soils of Yelagiri Hills, Tamilnadu India with Statistical Approach. *Procedia Earth and Planetary Science*, 11, 531–538. <https://doi.org/10.1016/j.proeps.2015.06.055>
- Chaney, R. L. (2010). Cadmium and Zinc. In P. S. Hooda (Ed.), *Trace elements in soils* (1st ed., p. 409). Chichester, West Sussex: Blackwell Publishing Ltd.
- Chapman, P. M. (2007). Determining when contamination is pollution - Weight of evidence determinations for sediments and effluents. *Environment International*, 33(4), 492–501. <https://doi.org/10.1016/J.ENVINT.2006.09.001>
- Chilvers, D. C., & Peterson, P. J. (1987). Lead, mercury, cadmium and arsenic in the environment. In T. C. Hutchinson & K. M. Meema (Eds.), *Global Cycling of Arsenic* (pp. 279–301). New York: John Wiley & Sons.
- Clesceri, L., Greenberg, A., & Eaton, A. (1998). Part 3000 Metals. In *Standard Methods for the Examination of Water and Wastewater* (20th ed., pp. 3–1). Baltimore: American Public Health Association.
- Clifford, M. J., Hilson, G. M., & Hodson, M. E. (2010). Tin and Mercury. In P. S. Hooda (Ed.), *Trace elements in soils* (1st ed., p. 501). London: Blackwell Publishing Ltd. <https://doi.org/10.1017/CBO9781107415324.004>

- Couto, N., Guedes, P., Zhou, D.-M., & Ribeiro, A. B. (2015). Integrated perspectives of a greenhouse study to upgrade an antimony and arsenic mine soil - Potential of enhanced phytotechnologies. *Chemical Engineering Journal*, 262, 563–570. <https://doi.org/10.1016/j.cej.2014.09.021>
- Desimoni, E., Casella, G. I., Salvi, A. M., Cataldi, T. R. I., & Morone, A. (1992). XPS investigation of ultra-high-vacuum storage effects on carbon fibre surfaces. *Carbon*, 30(4), 527–531. [https://doi.org/10.1016/0008-6223\(92\)90171-R](https://doi.org/10.1016/0008-6223(92)90171-R)
- Díez, M., Simón, M., Martín, F., Dorronsoro, C., García, I., & Van Gestel, C. A. M. (2009). Ambient trace element background concentrations in soils and their use in risk assessment. *Science of the Total Environment*, 407(16), 4622–4632. <https://doi.org/10.1016/j.scitotenv.2009.05.012>
- Director General of Minerals and Geoscience Malaysia. (2012). Geological Map of Peninsular Malaysia. Retrieved April 27, 2018, from [http://www.jmg.gov.my/add\\_on/mt/smnjg/tiles/](http://www.jmg.gov.my/add_on/mt/smnjg/tiles/)
- DPlot. (2014). DPlot Graph Software for Scientists and Engineers. Hydesoft Computing, LLC. Retrieved from <http://www.dplot.com/evaluation.htm>
- Eaton, A. D., Clesceri, L. S., Greenberg, A. E., & Franson, M. A. H. (1998a). General Precaution. In *Standard methods for the examination of water and wastewater* (20th ed.). Washington, DC: American Public Health Association, American Water Works Association and Water Environment Federation.
- Eaton, A. D., Clesceri, L. S., Greenberg, A. E., & Franson, M. A. H. (1998b). Inductively Coupled Plasma/Mass Spectrometry (ICP/MS) Method. In *Standard methods for the examination of water and wastewater* (20th ed.). Washington, DC: American Public Health Association, American Water Works Association and Water Environment Federation.
- Eaton, A. D., Clesceri, L. S., Greenberg, A. E., & Franson, M. A. H. (1998c). Nitric Acid Digestion. In *Standard methods for the examination of water and wastewater* (20th ed.). Washington, DC: American Public Health Association, American Water Works Association and Water Environment Federation.
- Eaton, A. D., Clesceri, L. S., Greenberg, A. E., & Franson, M. A. H. (1998d). Standard methods for the examination of water and wastewater. In *Standard methods for the examination of water and wastewater* (20th ed.). Washington, DC: American Public Health Association, American Water Works Association and Water Environment Federation. Retrieved from <http://www.worldcat.org/title/standard-methods-for-the-examination-of-water-and-wastewater-20th-edition/oclc/1000000000>

- methods-for-the-examination-of-water-and-wastewater/oclc/40733179
- Eaton, A. D., Clesceri, L. S., Greenberg, A. E., & Franson, M. A. H. (1998e). Turbidity. In *Standard methods for the examination of water and wastewater* (20th ed.). Washington, DC: American Public Health Association, American Water Works Association and Water Environment Federation.
- Eisenbud, M. (1987). *Environmental Radioactivity* (Third edit). New York: Academic Press Inc. <https://doi.org/10.1016/B978-0-12-235154-9.X5000-2>
- ELTurk, M., Abdullah, R., Mohamad Zakaria, R., & Abu Bakar, N. K. (2018). Evaluation of heavy metals and environmental risk assessment in the Mangrove Forest of Kuala Selangor estuary, Malaysia. *Marine Pollution Bulletin*, 136(November), 1–9. <https://doi.org/10.1016/j.marpolbul.2018.08.063>
- Evans, L. J., & Barabash, S. J. (2010). Molybdenum, Silver, Thallium and Vanadium. In P. S. Hooda (Ed.), *Trace elements in soils* (1st ed., p. 515). London: Blackwell Publishing Ltd. <https://doi.org/10.1017/CBO9781107415324.004>
- Excel. (2013). Microsoft Office Professional Plus 2013.
- Fauziah, S. H., Izzati, M. N., & Agamuthu, P. (2013). Toxicity on Anabas Testudineus: a case study of sanitary landfill leachate. *Procedia Environmental Sciences*, 18, 14–19. <https://doi.org/10.1016/j.proenv.2013.04.003>
- Feinendegen, L. E., Pollicove, M., & Sondhaus, C. A. (2004). Responses to low doses of ionizing radiation in biological systems. *Nonlinearity in Biology, Toxicology, and Medicine*, 2(2004), 143–171. <https://doi.org/10.1080/15401420490507431>
- Floris, B., Galloni, P., Sabuzi, F., & Conte, V. (2017). Metal systems as tools for soil remediation. *Inorganica Chimica Acta*, 455, 429–445. <https://doi.org/10.1016/j.ica.2016.04.003>
- Fowler, M. E., & Miller, R. E. (2003). *Zoo and Wild Animal Medicine* (5th edn). Philadelphia: W.B. Saunders.
- Galán, E., Romero-Baena, A., Aparicio, P., & González, I. (2019). A methodological approach for the evaluation of soil pollution by potentially toxic trace elements. *Journal of Geochemical Exploration*, 203(August), 96–107.
- Gao, S., Tanji, K. K., & Goldberg, S. (1994). Potentially toxic trace elements in soils and sediments. In L. Dudley & J. Guitjens (Eds.), *Symposium on Sources, Control, and Remediation of Oxyanions*. San Francisco.
- Garmin. (2010). *GPSMAP 62 series ® User Manual*.
- Gary W vanLoon, & Stephen J Duffy. (2011). *Environmental Chemistry: A Global*

- Perspective* (3rd ed.). New York: Oxford University Press. Retrieved from [https://books.google.com.my/books/about/Environmental\\_Chemistry.html](https://books.google.com.my/books/about/Environmental_Chemistry.html)
- Gatti, E., Saidin, M., Talib, K., Rashidi, N., Gibbard, P., & Oppenheimer, C. (2013). Depositional processes of reworked tephra from the Late Pleistocene Youngest Toba Tuff deposits in the Lenggong Valley, Malaysia, 79, 228–241.
- Gavrilescu, M., Pavel, L. V., & Cretescu, I. (2009). Characterization and remediation of soils contaminated with uranium. *Journal of Hazardous Materials*, 163(2–3), 475–510. <https://doi.org/10.1016/j.jhazmat.2008.07.103>
- Geert, P. (2013). *Marine Pollution* (1st editio). Oxford Press University. Retrieved from [http://site.iugaza.edu.ps/elnabris/files/2015/09/1\\_What-is-pollution.pdf](http://site.iugaza.edu.ps/elnabris/files/2015/09/1_What-is-pollution.pdf)
- Goher, M. E., Farhat, H. I., Abdo, M. H., & Salem, S. G. (2014). Metal pollution assessment in the surface sediment of Lake Nasser, Egypt. *Egyptian Journal of Aquatic Research*, 40, 213–224. <https://doi.org/doi.org/10.1016/j.ejar.2014.09.004>
- Google Earth. (2013). Google Earth. Retrieved May 9, 2018, from <https://earth.google.com/web/@1.86231886,102.99983894,20.88268978a,4478.55449394d,35y,-0h,0t,0r>
- GW Instek. (2014). *DC Power Supply User Manual*. Taipei: GW Instek.
- Hagedorn, B. (2008). Acid digestion of waters for total recoverable metals (following EPA Method 3005). Retrieved January 8, 2017, from [file:///D:/Downloads/ACID\\_DIGESTION\\_OF\\_WATERS\\_FOR\\_TOTAL\\_REC\\_OVERABLE\\_ME \(1\).pdf](file:///D:/Downloads/ACID_DIGESTION_OF_WATERS_FOR_TOTAL_REC_OVERABLE_ME (1).pdf)
- Hamby, D. M., & Tynybekov, A. K. (2002). Uranium, Thorium, and Potassium in Soils Along the Shore of Lake Issyk-Kyol in the Kyrgyz Republic. *Environmental Monitoring and Assessment*, 73(2), 101–108. <https://doi.org/10.1023/A:1013071414970>
- Hashemi, S. E., Rezaee, A., Mousavi, S. M., Nikodel, M. R., & Ganjidoust, H. (2014). Optimization of pyrene removal from contaminated soil by electrokinetic remediation process. *Iranian Journal of Health, Safety and Environment*, 1(1), 16–22.
- Hassan, A. (2006). Extent Of Peatlands And ‘C’ Contents Of Soils In Peninsular Malaysia. In *Workshop On Vulnerability of Carbon Pools of Tropical Peatlands In Asia*. Pekanbaru, Riau, Sumatra, Indonesia.
- Hawkesworth, C. J., & Kemp, A. I. S. (2006). Evolution of the continental crust.

- Nature*, 443(October), 811–817. <https://doi.org/10.1038/nature05191>
- Hooda, P. S. (2010). Introduction. In P. S. Hooda (Ed.), *Trace elements in soils* (1st ed., p. 3). London: Blackwell Publishing Ltd. <https://doi.org/10.1017/CBO9781107415324.004>
- Hough, R. L. (2010). Copper and lead. In P. S. Hooda (Ed.), *Trace elements in soils* (1st ed., p. 441). London: Blackwell Publishing Ltd.
- ICRP. (2018a). *ICRP 2018 Annual Report. International Commission on Radiological Protection*. Ontario. Retrieved from [www.icrp.org](http://www.icrp.org)
- ICRP. (2018b). ICRPAEDIA: Dose Limits. Retrieved May 26, 2018, from <http://www.icrp.org/icrpaedia/limits.asp>
- Imanishi, Y., Bando, A., Komatani, S., Wada, S., & Tsuji, K. (2010). Experimental Parameters for XRF Analysis of Soils, 248–255. <https://doi.org/10.4028/www.scientific.net/MSF.278-281.151>
- Institute of Geophysical and Geochemical Exploration. (2012). Certificate of Certified Reference Material. Langfang China: Institute of Geophysical and Geochemical Exploration.
- Instruction JEOL. (2000). Mini Cup EDS Detector.
- Jayanthi, B., Emenike, C. U., Agamuthu, P., Simarani, K., Mohamad, S., & Fauziah, S. H. (2016). Selected microbial diversity of contaminated landfill soil of Peninsular Malaysia and the behavior towards heavy metal exposure. *Catena*, 147, 25–31. <https://doi.org/10.1016/j.catena.2016.06.033>
- Jayanthi, B., Emenike, C. U., Auta, S. H., Agamuthu, P., & Fauziah, S. H. (2017). Characterization of induced metal responses of bacteria isolates from active non-sanitary landfill in Malaysia. *International Biodeterioration & Biodegradation*, 119(April), 467–475. <https://doi.org/10.1016/j.ibiod.2016.10.053>
- JEOL. (2000). *JSM-6390 Series*.
- Joel, G. S. C., Penabei, S., Ndontchueng, M. M., Chene, G., Mekontso, E. J. N., Ebongue, A. N., ... David, S. (2017). Precision measurement of radioactivity in gamma-rays spectrometry using two HPGe detectors (BEGe-6530 and GC0818-7600SL models) comparison techniques: Application to the soil measurement. *MethodsX*, 4, 42–54. <https://doi.org/10.1016/j.mex.2016.12.003>
- Kalnicky, D. J., & Singhvi, R. (2001). Field portable XRF analysis of environmental samples. *Journal of Hazardous Materials*, 83, 93–122.
- Kampa, M., & Castanas, E. (2008). Human health effects of air pollution.

- Environmental Pollution*, 151(2), 362–367. <https://doi.org/10.1016/j.envpol.2007.06.012>
- Keshavarzifard, M., Zakaria, M. P., Tan, S. H., Yusuff, F. M., Mustafa, S., Vaezzadeh, V., ... Abootalebi-Jahromi, F. (2014). Polycyclic aromatic hydrocarbons (PAHs) in sediments from Prai and Malacca Rivers, Peninsular Malaysia. *Marine Pollution Bulletin*, 88, 366–372. <https://doi.org/10.1016/j.marpolbul.2014.08.014>
- Kim, G.-N., Kim, S.-S., Park, U.-R., & Moon, J.-K. (2015). Decontamination of Soil Contaminated with Cesium using Electrokinetic-electrodialytic Method. *Electrochimica Acta*, 3–7. <https://doi.org/10.1016/j.electacta.2015.03.208>
- Kim, G.-N., Park, U.-R., Kim, S.-S., Kim, W.-S., Moon, J.-K., & Hyun, J.-H. (2014). Decontamination of gravels contaminated with uranium. *Annals of Nuclear Energy*, 72, 367–372. <https://doi.org/10.1016/j.anucene.2014.05.031>
- Kim, G. N., Kim, S. S., Moon, J. K., & Hyun, J. H. (2015). Removal of uranium from soil using full-sized washing electrokinetic separation equipment. *Annals of Nuclear Energy*, 81, 188–195. <https://doi.org/10.1016/j.anucene.2015.01.046>
- Kim, G., Shon, D., Park, H., Lee, K., & Chung, U. (2011). Development of pilot-scale electrokinetic remediation technology for uranium removal. *Separation and Purification Technology*, 80(1), 67–72. <https://doi.org/10.1016/j.seppur.2011.04.009>
- Kim Gye-Nam, Kim Seung-Soo, Park Hye-Min, Kim Wan-Suk, Moon Jei-Kwon, & Hyeon Jay-Hyeok. (2012). Development of complex electrokinetic decontamination method for soil contaminated with uranium. *Electrochimica Acta*, 86, 49–56. <https://doi.org/10.1016/j.electacta.2012.06.041>
- Kim, K.-W., Lee, K.-Y., & Kim, S.-O. (2009). Electrokinetic remediation of mixed metal contamination. In K. R. Reddy & C. Cameselle (Eds.), *Electrochemical Remediation Technologies for Polluted Soils, Sediments and Groundwater* (pp. 287–310). Hoboken, New Jersey: Wiley.
- Kingston, H. M., & Walter, P. J. (1998). The Art and Science of Microwave Sample Preparations for Trace and Ultratrace Elemental Analysis. In A. Montaser (Ed.), *Inductively coupled plasma mass spectrometry* (p. 964). New York: J. Wiley. Retrieved from <https://uthm.on.worldcat.org/oclc/37464872>
- Koh Meng Hock, & Ramli, A. T. (2006). Iso dose-rate contour map of terrestrial gamma radiation in Johor Bahru district, Johor. *Jurnal Fizik UTM*, 1, 16–21.

- Korolev, V. A. (2009). Electrokinetic removal of radionuclides. In K. R. Reddy & C. Cameselle (Eds.), *Electrochemical Remediation Technologies for Polluted Soils, Sediments and Groundwater* (1st ed., pp. 127–139). Hoboken, New Jersey: Wiley.
- Krcmar, D., Varga, N., Prica, M., Cveticanin, L., Zukovic, M., Dalmacija, B., & Corba, Z. (2017). Application of hexagonal two dimensional electrokinetic system on the nickel contaminated sediment and modelling the transport behavior of nickel during electrokinetic treatment. *Separation and Purification Technology*, 192, 253–261. <https://doi.org/10.1016/j.seppur.2017.10.008>
- Laing, G. Du. (2010). Analysis and fractionation of trace elements in soils. In P. S. Hooda (Ed.), *Trace elements in soils* (1st ed., pp. 53–74). London: Blackwell Publishing Ltd. <https://doi.org/10.1017/CBO9781107415324.004>
- Lariviere, D., & Guerin, N. (2010). Natural Radioactivity. In D. A. Atwood (Ed.), *Radionuclides in the Environment* (1st ed., pp. 1–18). Chichester: John Wiley & Sons, Ltd.
- Lee, S. K., Wagiran, H., Ramli, A. T., Apriantoro, N. H., & Wood, A. K. (2009). Radiological monitoring : terrestrial natural radionuclides in Kinta District , Perak , Malaysia. *Journal of Environmental Radioactivity*, 100(5), 368–374. <https://doi.org/10.1016/j.jenvrad.2009.01.001>
- Lee, Y. H. (1985). Aluminium Speciation in Different Water Types. *Ecological Bulletins*, (37), 109–119. Retrieved from <http://www.jstor.org/stable/20112940>
- Lenntech. (2018). Silicon and water: reaction mechanisms, environmental impact and health effects. Retrieved June 19, 2018, from <https://www.lenntech.com/periodic/water/silicon/silicon-and-water.htm>
- Linsinger, T. (2010). *Application Note 1: Comparison of a measurement result with the certified value*. Geel, Belgium.
- Liu, S.-H., & Wang, H. P. (2004). In situ speciation studies of copper-humic substances in a contaminated soil during electrokinetic remediation. *Journal of Environmental Quality*, 33(4), 1280–1287. <https://doi.org/10.2134/jeq2004.1280>
- Lu, X., Zhao, C., Chen, C., & Liu, W. (2012). Radioactivity level of soil around Baqiao coal-fired power plant in China. *Radiation Physics and Chemistry*, 81(12), 1827–1832. <https://doi.org/10.1016/j.radphyschem.2012.07.013>
- Ludlum Measurement. (2012). *LUDLUM MODEL 19 MICRO R METER*. Sweetwater, Texas: Ludlum Measurement, Inc.

- Lukman, S., Essa, M. H., Mu'azu, N. D., Bukhari, A., & Basheer, C. (2013). Adsorption and desorption of heavy metals onto natural clay material: Influence of initial pH. *Journal of Environmental Science and Technology*, 6(1), 1–15. <https://doi.org/10.3923/jest.2013.1.15>
- Ma, Y., & Hooda, P. S. (2010). Chromium, nickel and cobalt. In P. S. Hooda (Ed.), *Trace elements in soils* (1st ed.). London: Blackwell Publishing Ltd.
- Maini, G., Sharman, A. K., Knowles, C. J., Sunderland, G., & Jackman, S. A. (2000). Electrokinetic remediation of metals and organics from historically contaminated soil. *Journal of Chemical Technology & Biotechnology*, 75(8), 657–664. [https://doi.org/10.1002/1097-4660\(200008\)75:8<657::AID-JCTB263>3.0.CO;2-5](https://doi.org/10.1002/1097-4660(200008)75:8<657::AID-JCTB263>3.0.CO;2-5)
- Makoundi, C., Zaw, K., Large, R. R., Meffre, S., Chun, K. L., & Teh, G. H. (2014). Geology, geochemistry and metallogenesis of the Selinsing gold deposit, central Malaysia. *Gondwana Research*, 26(1), 241–261. <https://doi.org/10.1016/j.gr.2013.08.023>
- Martin, B. R. (2012). *Statistics for Physical Sciences : An Introduction* (1st ed.). Amsterdam ; Elsevier, Academic Press.
- Masi, M., Ceccarini, A., & Iannelli, R. (2017). Model-based optimization of field-scale electrokinetic treatment of dredged sediments. *Chemical Engineering Journal*, 328, 87–97. <https://doi.org/10.1016/j.cej.2017.07.004>
- Mat Ripin, S. N., Hasan, S., Kamal, M. L., & Mohd Hashim, N. (2014). Analysis and pollution assessment of heavy metal in soil, Perlis. *Malaysian Journal of Analytical Sciences*, 18(1), 155–161. <https://doi.org/10.1023/A:1007826405826>
- Matschullat, J. (2000). Arsenic in the geosphere-a review. *Sci. Total Environ.*, 249, 297–312.
- Mazlan, A. Z., Hussain, H., & Mohamed Zawawi, M. A. (2016). Potential dermal exposure assessment of farmers to herbicide imazapic in an agriculture area. In M. Y. Abbas & S. K. Syed Othman Thani (Eds.), *ASLI (Annual Serial Landmark International) Conferences on QoL 2016. AMER International Conference on Quality of Life, AicQoL2016 Medan, Indonesia, 25–27 February 2016* (Vol. 234, pp. 144–153). Medan: Elsevier Ltd. <https://doi.org/10.1016/j.sbspro.2016.10.229>
- Mc Laughlin, J. P. (2015). Some characteristics and effects of natural radiation. *Radiation Protection Dosimetry*, (Advance Access), 1–6. <https://doi.org/10.1093/rpd/ncv206>

- Merry, R. H. (2009). Acidity and alkalinity of soils. In A. Sabljic (Ed.), *Environmental and Ecological Chemistry* (Vol. 2, pp. 115–131). UNESCO and Encyclopedia of Life Support Systems (EOLSS). Retrieved from <http://www.eolss.net/sample-chapters/c06/e6-13-03-01.pdf>
- Mohd Kusin, F., Mohd Azani, N. N., Syed Hasan, S. N. M., & Sulong, N. A. (2018). Distribution of heavy metals and metalloid in surface sediments of heavily-mined area for bauxite ore in Pengerang, Malaysia and associated risk assessment. *CATENA*, 165(June), 454–464. <https://doi.org/10.1016/j.catena.2018.02.029>
- Mohd Radzi, N. A. S., Abu Bakar, N. K., Emenike, C. U., & Abas, M. R. (2015). Polycyclic aromatic hydrocarbons (PAHs): contamination level and risk assessment in urban areas, Kuala Lumpur, Malaysia. In *International Conference on Business, Economics, Energy and Environmental Sciences (ICBEEES) 19–21 September 2014, Kuala Lumpur, Malaysia* (pp. 1–20). Kuala Lumpur: Balaban Desalination Publications. <https://doi.org/10.1080/19443994.2015.1021103>
- Moore, J. D., & Donaldson, J. A. (2016). Human-Scale Economics: Economic Growth and Poverty Reduction in Northeastern Thailand. *World Development*, 85(September), 1–15. <https://doi.org/10.1016/j.worlddev.2016.04.004>
- Morera-Gómez, Y., Cartas-Aguila, H. A., Alonso-Hernández, C. M., & Nuñez-Duwartes, C. (2016). Validation of an efficiency calibration procedure for a coaxial n-type and a well-type HPGe detector used for the measurement of environmental radioactivity. *Nuclear Instruments and Methods in Physics Research, Section A: Accelerators, Spectrometers, Detectors and Associated Equipment*, 818, 51–56. <https://doi.org/10.1016/j.nima.2016.02.039>
- MOSTI. (2010). *National Coastal Resources and Marine Environment Profile of Malaysia*. Putrajaya, Malaysia. Retrieved from <http://www.mynodc.gov.my/index>
- Mothersill, C., & Seymour, C. (2014). Implications for human and environmental health of low doses of ionising radiation. *Journal of Environmental Radioactivity*, 133, 5–9. <https://doi.org/10.1016/j.jenvrad.2013.04.002>
- Mulligan, C. N., Yong, R. N., & Gibbs, B. F. (2013). Remediation Technologies for Soils and Groundwater. *Remediation Technologies for Soils and Groundwater*, 60, 193–207. <https://doi.org/10.1061/9780784408940>
- Mulware, S. J. (2012). Trace elements and carcinogenicity: a subject in review. *3 Biotech*, 3(2), 85–96. <https://doi.org/10.1007/s13205-012-0072-6>

- Naudé, W., Szirmai, A., & Lavopa, A. (2013). *Industrialization Lessons from BRICS: A Comparative Analysis* (IZA Discussion Paper No. 7543). Bonn. Retrieved from [https://papers.ssrn.com/sol3/papers.cfm?abstract\\_id=2314802](https://papers.ssrn.com/sol3/papers.cfm?abstract_id=2314802)
- Noli, F., & Tsamos, P. (2016). Concentration of heavy metals and trace elements in soils, waters and vegetables and assessment of health risk in the vicinity of a lignite-fired power plant.pdf. *Science of The Total Environment*, 563–564(1 September 2016), 377–385. <https://doi.org/doi:10.1016/j.scitotenv.2016.04.098>
- NRC. (1972). *Lead: Airborne lead in perspective*. Washington DC: National Academy of Sciences.
- Oh, T. H., Hasanuzzaman, M., Selvaraj, J., Teo, S. C., & Chua, S. C. (2018). Energy policy and alternative energy in Malaysia: Issues and challenges for sustainable growth – An update. *Renewable and Sustainable Energy Reviews*, 81(Part 2), 3021–3031. <https://doi.org/10.1016/j.rser.2017.06.112>
- Omar, W., Paramananthan, S., Harun, M. H., Abdul Aziz, N., & Kushairi, A. (2013). Malaysian unified peat classification technique. *MPOB Information Series*.
- Onnittan, A., Sillanpaa, M., Cameselle, C., & Reddy, K. R. (2009). Field Applications of Electrokinetic Remediation of Soils Contaminated with Heavy Metals. In K. R. Reddy & C. Cameselle (Eds.), *Electrochemical Remediation Technologies for Polluted Soils, Sediments and Groundwater* (p. 760). Hoboken, New Jersey: Wiley.
- Osmond, D. L., Line, D. E., Gale, J. A., Gannon, R. W., Knott, C. B., Bartenhagen, K. A., ... Lehning, D. W. (1995). Water Resource Characterization DSS - Heavy Metals. Retrieved May 14, 2018, from <http://www.water.ncsu.edu/watershedss/info/hmetals.html>
- Ottosen, L. M., Hansen, H. K., & Jensen, P. E. (2009). Electrokinetic removal of heavy metals. In K. R. Reddy & C. Cameselle (Eds.), *Electrochemical Remediation Technologies for Polluted Soils, Sediments and Groundwater* (1st ed., pp. 97–120). New Jersey: John Wiley & Sons.
- Ottosen, L. M., Hansen, H. K., Laursen, S., & Villumsen, A. (1997). Electrodialytic Remediation of Soil Polluted with Copper from Wood Preservation Industry. *Environ. Sci. Technol.*, 31(6), 1711–1715. <https://doi.org/10.1021/es9605883>
- Pejabat Setiausaha Eksekutif AELB. (2011). *Laporan Tahunan 2011*. Dengkil. Retrieved from [wwwaelb.gov.my](http://wwwaelb.gov.my)
- Perkin Elmer. (2001). The 30-Minute Guide to ICP-MS. *Technical Note: ICP-Mass*

- Spectrometry*, 1–8. <https://doi.org/10.1016/j.pnucene.2012.01.005>
- Perkin Elmer. (2019). Multi-element Calibration Standard 3, Matrix per Volume: 5% HNO<sub>3</sub> per 125 mL. Retrieved January 1, 2019, from <http://www.perkinelmer.com/product/icp-ms-cal-std-3-n9300233>
- Prion, S., & Haerling, K. A. (2014). Making Sense of Methods and Measurement: Pearson Product-Moment Correlation Coefficient. *Clinical Simulation in Nursing*, 10(11), 587–588. <https://doi.org/10.1016/j.ecns.2014.07.010>
- Pulford, I. D. (2010). Gold and Uranium. In P. S. Hooda (Ed.), *Trace elements in soils* (1st ed., p. 551). London: Blackwell Publishing Ltd.
- Qing, X., Yutong, Z., & Shenggao, L. (2015). Assessment of heavy metal pollution and human health risk in urban soils of steel industrial city (Anshan), Liaoning, Northeast China. *Ecotoxicology and Environmental Safety*, 120, 377–385. <https://doi.org/10.1016/j.ecoenv.2015.06.019>
- Ramli, A. T. (1997). Environmental Terrestrial Gamma Dose and its Relationship with Soil Type and Underlying Geological Formations in Pontian District, Malaysia. *Appl. Radiat. Isot.*, 48(3), 407–412.
- Ramli, A. T., Hussein, A. W. M. A., & Lee, M. H. (2001). Geological influence on terrestrial gamma radiation dose rate in the Malaysian State of Johore. *Appl. Radiat. Isot.*, 54, 327–333.
- Reddy, K. R., & Cameselle, C. (2009). Overview of Electrochemical Remediation Technologies. In K. R. Reddy & C. Cameselle (Eds.), *Electrochemical Remediation Technologies for Polluted Soils, Sediments and Groundwater* (1st ed., pp. 1–28). Hoboken, New Jersey: Wiley.
- Reiche, S., Blume, R., Zhao, X. C., Su, D., Kunkes, E., Behrens, M., & Schlögl, R. (2014). Reactivity of mesoporous carbon against water - An in-situ XPS study. *Carbon*, 77(October), 175–183. <https://doi.org/10.1016/j.carbon.2014.05.019>
- Reyes-Gutiérrez, L. R., Romero-Guzmán, E. T., Cabral-Prieto, A., & Rodríguez-Castillo, R. (2007). Characterization of Chromium in Contaminated Soil Studied by SEM, EDS, XRD and Mössbauer Spectroscopy. *Journal of Minerals & Materials Characterization & Engineering*, 7(1), 59–70.
- Rice, E. W., Baird, R. B., & Eaton, A. D. (2017). Standard Methods for the Examination of Water and Wastewater. Retrieved May 5, 2019, from <https://www.bookdepository.com/Standard-Methods-for-the-Examination-of-Water-and-Wastewater-E.W.-Rice/Examination-of-Water-and-Wastewater>

- Rice, E. W., Baird, R. B., Eaton, A. D., & Clesceri, L. S. (2012). *Standard Methods for the Examination of Water and Wastewater*. (L.S. Clesceri, Ed.) (22nd ed.). American Public Health Association, American Water Works Association, Water Environment Federation.
- Rouessac, F., & Rouessac, A. (2013). *Chemical Analysis: Modern Instrumentation Methods and Techniques* (2nd ed.). Wiley. Retrieved from <https://books.google.com.my/books?id=4XmjFLkJGygC>
- Ruiz-Serrano, D., Flores-Acosta, M., Conde-Barajas, E., Ramírez-Rosales, D., Yáñez-Limón, J. M., & Ramírez-Bon, R. (2010). Study by XPS of different conditioning processes to improve the cation exchange in clinoptilolite. *Journal of Molecular Structure*, 980(1), 149–155. <https://doi.org/10.1016/j.molstruc.2010.07.007>
- Sahoo, S. K., Kierepko, R., Sorimachi, A., Omori, Y., Ishikawa, T., Tokonami, S., ... Ramola, R. C. (2016). Natural radioactivity level and elemental composition of soil samples from a high background radiation area on eastern coast of India (Odisha). *Radiation Protection Dosimetry*, 171(2), 172–178. <https://doi.org/10.1093/rpd/ncw052>
- Saim, A. bin A. (2017). Unsur Surih. Retrieved March 5, 2018, from <http://www.myhealth.gov.my/unsur-surih/>
- Sakawi, Z., Syed Abdullah, S. M., Ariffin, M. R., Ismail, L., & Jaafar, O. (2012). Analysis of heavy metal concentration in the vicinity of a landfill site. *Research Journal of Applied Sciences*, 7(7), 349–353. <https://doi.org/10.3923/rjasci.2012.349.353>
- Saleh, M. A., Ramli, A. T., Alajerami, Y., & Aliyu, A. S. (2013). Assessment of environmental  $^{226}\text{Ra}$ ,  $^{232}\text{Th}$  and  $^{40}\text{K}$  concentrations in the region of elevated radiation background in Segamat District, Johor, Malaysia. *Journal of Environmental Radioactivity*, 124, 130–140. <https://doi.org/10.1016/j.jenvrad.2013.04.013>
- Saleh, M. A., Ramli, A. T., Alajerami, Y., Aliyu, A. S., & Basri, N. A. (2013). Radiological study of Mersing District, Johor, Malaysia. *Radiation Physics and Chemistry*, 85, 107–117. <https://doi.org/10.1016/j.radphyschem.2012.12.045>
- Saleh, M. A., Ramli, A. T., Alajeramie, Y., Hashim, S., Aliyu, A. S., & Basri, N. A. (2013). Terrestrial gamma radiation and its statistical relation with geological formation in the Mersing district, Johor, Malaysia. *Radiation Protection Dosimetry*, (Advanced Access), 1–7.

- Saleh, M. A., Ramli, A. T., Hamzah, K., Alajerami, Y., Moharib, M., & Saeed, I. (2015). Prediction of terrestrial gamma dose rate based on geological formations and soil types in the Johor State, Malaysia. *Journal of Environmental Radioactivity*, 148, 111–122. <https://doi.org/10.1016/j.jenvrad.2015.05.019>
- Sanusi, M. S. M., Ramli, A. T., Gabdo, H. T., Garba, N. N., Heryanshah, A., Wagiran, H., & Said, M. N. (2014). Isodose mapping of terrestrial gamma radiation dose rate of Selangor state, Kuala Lumpur and Putrajaya, Malaysia. *Journal of Environmental Radioactivity*, 135, 67–74. <https://doi.org/10.1016/j.jenvrad.2014.04.004>
- Schmeling, M., & Aldstadt, J. H. (2009). Fundamental Environmental Chemistry. In *Environmental and Ecological Chemistry* (pp. 1–35). UNESCO and Encyclopedia of Life Support Systems (EOLSS).
- Schweitzer, L., & Noblet, J. (2018). Water Contamination and Pollution. In *Green Chemistry An Inclusive Approach* (pp. 261–290). Elsevier. Retrieved from <https://doi.org/10.1016/B978-0-12-809270-5.00011-X>
- Shackelford, C. D., & Daniel, D. E. (1991). Diffusion in saturated soil. I: Background. *Journal of Geotechnical Engineering*, 117(3), 467–484. Retrieved from <http://www.engr.colostate.edu/ce/homepages/shackel/Papers/4.pdf>
- Shafie, N. A., Aris, A. Z., & Haris, H. (2015). Geoaccumulation and distribution of heavy metals in the urban river sediment, 29(3), 368–377.
- Shen, Z., Chen, X., Jia, J., Qu, L., & Wang, W. (2007). Comparison of electrokinetic soil remediation methods using one fixed anode and approaching anodes. *Environmental Pollution*, 150(2), 193–199. <https://doi.org/10.1016/j.envpol.2007.02.004>
- Singh, V., & Agrawal, H. M. (2012). Qualitative soil mineral analysis by EDXRF , XRD and AAS probes. *Radiation Physics and Chemistry*, 81(12), 1796–1803. <https://doi.org/10.1016/j.radphyschem.2012.07.002>
- Skuce, M. (2011). *Characterization of nickel speciation in soils near Kalgoorlie, Western Australia*. Queen's University. Retrieved from [http://www.queensu.ca/ensc/undergraduate/courses/ensc501/pastprojects/Skuce\\_ENSC501.pdf](http://www.queensu.ca/ensc/undergraduate/courses/ensc501/pastprojects/Skuce_ENSC501.pdf)
- Slessarev, E. W., Lin, Y., Bingham, N. L., Johnson, J. E., Dai, Y., Schimel, J. P., & Chadwick, O. A. (2016). Water balance creates a threshold in soil pH at the global scale. *Nature*, 540, 567–569. <https://doi.org/10.1038/nature20139>
- Smith, K. A., & Paterson, J. E. (1995). Heavy Metals in Soils. In B. J. Alloway (Ed.),

- Manganese and cobalt* (2nd ed., pp. 255–244). London: Blackie Academic & Professional.
- Sohrabi, M. (2013). World high background natural radiation areas: Need to protect public from radiation exposure. *Radiation Measurements*, 50. <https://doi.org/10.1016/j.radmeas.2012.03.011>
- Spa Plating. (2017a). Electrode Handle. Retrieved January 11, 2017, from <http://www.goldn.co.uk/electrode-handle/>
- Spa Plating. (2017b). Platinum Electrode. Retrieved January 11, 2017, from <http://www.goldn.co.uk/platinum-electrode/>
- Spa Plating. (2017c). Stainless Steel Electrode. Retrieved January 11, 2017, from <http://www.goldn.co.uk/stainless-steel-electrode/>
- Staff of the Soil Survey Division. (1968). Reconnaissance Soil Map of Peninsular Malaysia. Sheet 1. Series L 40A. Retrieved December 14, 2011, from [http://eusoils.jrc.ec.europa.eu/ESDB\\_Archive/EuDASM/Asia/images/maps/download/MY3004\\_2SO.jpg](http://eusoils.jrc.ec.europa.eu/ESDB_Archive/EuDASM/Asia/images/maps/download/MY3004_2SO.jpg)
- Statistics Solutions. (2018). Pearson's Correlation Coefficient. Retrieved September 19, 2018, from <http://www.statisticssolutions.com/pearsons-correlation-coefficient/>
- Steimle, R. (1995). *In Situ Remediation Technology: Electrokinetics*. Washington, DC.
- Su, C., Jiang, L., & Zhang, W. (2014). A review on heavy metal contamination in the soil worldwide: Situation, impact and remediation techniques. *Environmental Skeptics and Critics*, 3(2), 24–38. Retrieved from [www.iaees.org](http://www.iaees.org)
- Suhana, J., & Rashid, M. (2016). Naturally occurring radionuclides in particulate emission from a coal fired power plant: A potential contamination? *Journal of Environmental Chemical Engineering*, 4(4), 4904–4910. <https://doi.org/10.1016/j.jece.2016.07.015>
- Suied, A. A., Ahmad Tajudin, S. A., Zakaria, M. N., & Madun, A. (2018). Potential Electrokinetic Remediation Technologies of Laboratory Scale into Field Application- Methodology Overview. In *Journal of Physics: Conference Series* (Vol. 995, pp. 1–10). Institute of Physics Publishing. <https://doi.org/10.1088/1742-6596/995/1/012083>
- Sultan, K., & Shazili, N. A. (2009). Distribution and geochemical baselines of major , minor and trace elements in tropical topsoils of the Terengganu River basin ,

- Malaysia, 103, 57–68. <https://doi.org/10.1016/j.gexplo.2009.07.001>
- Swagelok. (2016). Swagelok Fittings and Valves. Retrieved January 11, 2017, from <https://www.swagelok.com/en/product>
- Tack, F. M. G. (2010). Trace Elements: General Soil Chemistry, Principles and Processes. In P. S. Hooda (Ed.), *Trace elements in soils* (1st ed., p. 9). Blackwell Publishing Ltd.
- Thakali, S., Herbert, E. A., & Di Toro, D. M. (2006). A terrestrial biotic ligand model. 2. Application to Ni and Cu toxicities to plants, invertebrates, and microbes in soil. *Environ. Sci. Technol.*, 40, 7094–7100.
- Tousi, E. T., Firoozabadi, M. M., & Shiva, M. (2016). Determination of the thorium potential in Shah-Kooh area in Iran by NAA and comparison with the results of ICP and XRF techniques. *Measurement: Journal of the International Measurement Confederation*, 90, 20–24. <https://doi.org/10.1016/j.measurement.2016.04.020>
- Tufail, M., Akhtar, N., & Waqas, M. (2006). Measurement of terrestrial radiation for assessment of gamma dose from cultivated and barren saline soils of Faisalabad in Pakistan. *Radiation Measurements*, 41(4), 443–451. <https://doi.org/10.1016/j.radmeas.2005.10.007>
- UN Environment. (2017). *Waste Management In ASEAN Countries*. Osaka. Retrieved from [https://wedocs.unep.org/bitstream/handle/20.500.11822/21134/waste\\_mgt\\_asean\\_summary.pdf](https://wedocs.unep.org/bitstream/handle/20.500.11822/21134/waste_mgt_asean_summary.pdf)
- Underwood, E. J. (1977). *Trace Elements in Human and Animal Nutrition* (4th edn). New York: Academic Press.
- United Nations. (2015). The World Population Prospects: 2015 Revision. Retrieved December 25, 2015, from <https://www.un.org/en/development/desa/publications/world-population-prospects-2015-revision.html>
- UNSCEAR. (2000). *Sources and effects of ionizing radiation (Volume 1: Sources)*. New York. Retrieved from [https://www.unscear.org/unscear/en/publications/2000\\_1.html](https://www.unscear.org/unscear/en/publications/2000_1.html)
- USAEC. (2000). *In-situ electrokinetic remediation of metal contaminated soils technology status report*.
- USEPA. (2006). In situ treatment technologies for contaminated soil. *United States Environmental Protection Agency*, (November), 1–35. Retrieved from [www.epa.gov/tio/tsp](http://www.epa.gov/tio/tsp)

- Vaezzadeh, V., Zakaria, M. P., & Chui, W. B. (2017). Aliphatic hydrocarbons and triterpane biomarkers in mangrove oyster (*Crassostrea belcheri*) from the west coast of Peninsular Malaysia. *Marine Pollution Bulletin*, 124, 33–42. <https://doi.org/10.1016/j.marpolbul.2017.07.008>
- Velizarova, E., Ribeiro, A. B., & Ottosen, L. M. (2002). A comparative study on Cu, Cr and As removal from CCA-treated wood waste by dialytic and electrodialytic processes. *Journal of Hazardous Materials*, 94(2), 147–160. [https://doi.org/10.1016/S0304-3894\(02\)00063-8](https://doi.org/10.1016/S0304-3894(02)00063-8)
- Vendemiatto, M. A., & Enzweiler, J. (2001). Routine Control of Accuracy in Silicate Rock Analysis by X-Ray Fluorescence Spectrometry. *Geostandards and Geoanalytical Research*, 25(2–3), 283–291. <https://doi.org/10.1111/j.1751-908X.2001.tb00604.x>
- Villarreal, R., Lozano, L. A., Melani, E. M., Salazar, M. P., Otero, M. F., & Soracco, C. G. (2019). Diffusivity and sorptivity determination at different soil water contents from horizontal infiltration. *Geoderma*, 338, 88–96.
- Virkutyte, J., Sillanpää, M., & Latostenmaa, P. (2002). Electrokinetic soil remediation - critical overview. *Science of The Total Environment*, 289(1), 97–121. [https://doi.org/10.1016/S0048-9697\(01\)01027-0](https://doi.org/10.1016/S0048-9697(01)01027-0)
- Wang, A., Bong, C. W., Xu, Y., Hassan, M. H. A., Ye, X., Bakar, A. F. A., ... Loh, K. H. (2017). Assessment of heavy metal pollution in surficial sediments from a tropical river-estuary-shelf system: A case study of Kelantan River, Malaysia. *Marine Pollution Bulletin*, 125(1–2), 492–500. <https://doi.org/10.1016/j.marpolbul.2017.08.010>
- Watts, J. F., & Wolstenholme, J. (2003). *An introduction to surface analysis by XPS and AES*. Chichester: John Wiley.
- Wedepohl, K. H. (1995). The composition of the continental crust. *Geochimica et Cosmochimica Acta*, 59(7), 1217–1232.
- WHO. (2011). Nuclear accidents and radioactive contamination of foods. Retrieved December 25, 2015, from [https://www.who.int/foodsafety/fs\\_management/radionuclides\\_and\\_food\\_300311.pdf](https://www.who.int/foodsafety/fs_management/radionuclides_and_food_300311.pdf)
- Wu, M. Z., Reynolds, D. A., Fourie, A., & Thomas, D. G. (2013). Optimal Field Approaches for Electrokinetic In Situ Oxidation Remediation. *Groundwater Monitoring & Remediation*, 33(1), 62–74. <https://doi.org/10.1111/j.1745-6592.2012.01410.x>

- Yanagi, T. (2011). *Chemical Composition of Continental Crust and the Primitive Mantle. Arc Volcano of Japan: Generation of Continental Crust from the Mantle* (1st ed.). Fukuoka: Springer Japan. <https://doi.org/10.1007/978-4-431-53996-4>
- Yap, C. K., Ismail, A., Tan, S. G., & Omar, H. (2002). Concentrations of Cu and Pb in the offshore and intertidal sediments of the west coast of Peninsular Malaysia. *Environment International*, 28, 467–479. [https://doi.org/10.1016/S0160-4120\(02\)00073-9](https://doi.org/10.1016/S0160-4120(02)00073-9)
- Yong, Y. S., Lim, Y. A., & Ilankoon, I. M. S. K. (2019). An analysis of electronic waste management strategies and recycling operations in Malaysia: Challenges and future prospects. *Journal of Cleaner Production*, 224(July), 151–166. <https://doi.org/10.1016/j.jclepro.2019.03.205>
- Yuan, G., Soma, M., Seyama, H., Theng, B. K. ., Lavkulich, L. ., & Takamatsu, T. (1998). Assessing the surface composition of soil particles from some Podzolic soils by X-ray photoelectron spectroscopy. *Geoderma*, 86(3–4), 169–181. [https://doi.org/10.1016/S0016-7061\(98\)00049-4](https://doi.org/10.1016/S0016-7061(98)00049-4)
- Zavodska, L., Kosorinova, E., Scerbakova, L., & Lesny, J. (2008). Environmental chemistry of uranium.
- Zhang, P., Liu, Y., Li, Z., Kan, A. T., & Tomson, M. B. (2018). Sorption and desorption characteristics of anionic surfactants to soil sediments. *Chemosphere*, 211(November), 1183–1192. <https://doi.org/10.1016/j.chemosphere.2018.08.051>
- Zhao, F. J., McGrath, S. P., & Merrington, G. (2007). Estimates of ambient background concentrations of trace metals in soils for risk assessment. *Environmental Pollution*, 148, 221–229. <https://doi.org/10.1016/j.envpol.2006.10.041>