MECHANICAL PROPERTIES OF CERAMICS TILES BY REPLACEMENT OF QUARTZ BY RHA AND POFA

HASSAN USMAN JAMO

A thesis submitted in partial fulfilment of the requirements for the award of the Doctor of Philosophy in Science

Faculty of Science, Technology and Human Development
Universiti Tun Hussein Onn Malaysia

DEDICATION

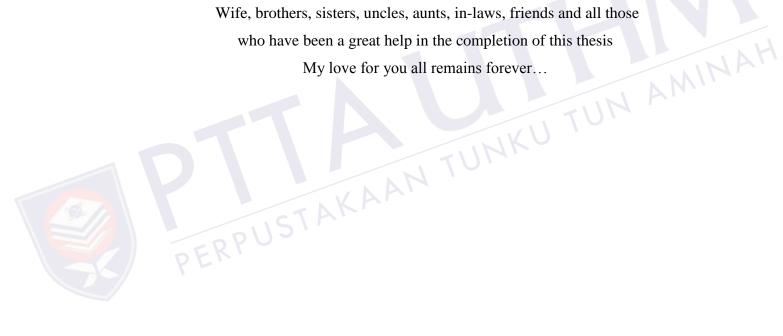
Specially dedicated to:

For my mother, father

Words cannot express everything you have done to me

Thank you

Wife, brothers, sisters, uncles, aunts, in-laws, friends and all those



ACKNOWLEDGEMENT

Praise to Allah, The Omnipotent, The Omnipresent and The Omniscience for making this thesis a reality. I wish to express my gratitude to my main supervisor, Asso. Prof. Dr. Mohamad Zaky Bin Noh, for the inspiration, motivation and professional guidance toward successful completion of my research work. I would like to also thank him for providing me with outstanding research facilities and numerous technical discussions which I found to be very valuable to my research. His constant enthusiasm and insightfulness will be a model for me throughout my career.

I would like to express my great appreciation to my co supervisor Prof. Dr. Hj. Zainal Arifin Hj. Ahmad for giving the opportunity to use equipment at USM, and his constant supervision while I was there to conduct some of the experiments. Numerous technical discussions which I found to be very valuable to my research. His constant enthusiasm and insightfulness will be a model for me throughout my career. Words are not enough to thank him. May Allah (S.W.T) extend his life give him good health, so that more and more students would benefit from his wealthy of experience. I must also not forget to thank his students such as Dr. Ahmad Zahir, Mr. Fahmin, Mr. Rashid and Mr. Joe for guiding me while I was in USM. I gratefully acknowledge the UTHM Graduate Incentive Scheme for their sponsorship throughout my study. My gratitude to Mr. Fazlannuddin Hanur bin Harith, Mr. Mohd Azrul Nizam bin Mustari, Mr. Shahrul Mahadi bin Samsudin, Mr. Mohd Tarmizi bin Nasir, Mr. Anuar bin Ismail, Mr. Ahmad Nasrull bin Mohamed Norsidah binti Harun and Mr. Nooriskandar Sani for their assistance as for their experimental and technical assistant.

My special and deepest appreciation and thanks go to both my parents Alhaji Shehu Jamo and Late Malama Mardhiyya Muhammad. Their constant support, encouragement gives me warmth and strength. They always are there to share my success as well during sad and down time. Their inspiration, understanding, patience and support help me to complete this thesis and no words are adequate to express my appreciation to both of them. I would like to thank my beloved wife, Aisha Muhammad Lawal, for her love and support for encouraging me in all my efforts to this day. I would also like to thank my in-laws Associate Prof. Muhammad Lawal Mayanchi (Capt. RTD) and his wife Hajiya Binta for their support during my studies in Malaysia. My time in UTHM wouldn't be pleasant without accompany of friends and colleagues Malam Abul-fazri Gambo and Malam Mansir D. for proofreading the manuscript of the thesis. Mr. Saffuwan bin Mohamed Johar and Mr. Uzair Majeed.



ABSTRACT

Rice husk ash (RHA) and palm oil fuel ash (POFA) have a great potential to replace the quartz element in porcelain composition. RHA and POFA were mostly used by construction industries, and only a few researchers studied its applications in ceramics industries. It is due to the mechanical properties of porcelain are strongly affected by the generated thermal stress during processing because of the deleterious effects of quartz. In this work, the quartz is being replaced by RHA, POFA and the combination of RHA and POFA. The sample's composition was mixed for 90 minutes and pressed at different mold pressures (31, 61, 91 and 121 MPa) and then sintered at the sintering temperatures ranging from 1000 to 1300 °C at different soaking times ranging from 1 to 3 hours respectively. The samples were measured the physical and mechanical properties and then the microstructure observation. It was found that the RHA, POFA and the combination of RHA and POFA have tremendous effects on the properties of porcelain tiles. For the RHA, the highest bulk density (2.42 g/cm³) and compressive strength (44 MPa) were recorded on 20 wt% of RHA at the sintering temperature of 1200 °C and the soaking time of 2 hours. For the POFA, the highest bulk density (2.45 g/cm³) and compressive strength (46 MPa) were achieved on 15 wt% of POFA at the sintering temperature of 1100 °C and the soaking time of 2 hours. For the combination of RHA and POFA, the highest bulk density (2.43 g/cm³) and compressive strength (45 MPa) were recorded on 20 wt% of RHA and POFA at the sintering temperature of 1200 °C and the soaking time of 2 hours. It was observed that the microstructure was enhanced by increasing the sintering temperature, mould pressure and soaking time. It can be concluded that samples containing POFA attained vitrification stage at lower temperature and exhibited higher mechanical properties. Thermal expansion and thermal conductivity measurement are some of the areas that could be explored for further research.

ABSTRAK

Abu sekam padi (ASP) dan abu bakar kelapa sawit (ABKS) mempunyai potensi yang besar untuk menggantikan kuartza dalam kompossisi porselin. KebanyakanASP dan ABKS digunakan oleh industri pembinaan, kecuali hanya beberapa penyelidik sahaja yang mengkaji penggunaannya dalam industri seramik. Ini adalah disebabkan oleh sifat mekanik porselin yang amat dipengaruhi oleh tekanan terma yang terhasil semasa pemprosesan kerana wujud kesan yang merosakkan kuartza. Dalam kajian ini, kuartza digantikan dengan ASP, ABKS dan kombinasi ASP dan ABKS. Komposisi sampel dicampurkan selama 90 minit dan ditekan pada tekanan acuan yang berbeza (31, 61, 91 dan 121 MPa) dan disinter pada suhu di antara 1000 °C hingga 1300 °C pada masa rendaman yang berbeza, masing-masing antara 1 hingga 3 jam. Sampel itu diukur sifat fizikal dan mekanikal dan kemudiannya dibuat pemerhatian mikrostruktur. Hasil kajian menunjukkan bahawa ASP, ABKS dan kombinasi ASP dan ABKS mempunyai kesan yang sangat baik pada sifat-sifat porselin. Untuk ASP, ketumpatan pukal (2.42 g/cm³) dan kekuatan mampatan(44 MPa) tertinggi dicatatkan pada 20 % berat ASP pada suhu sinter 1200 °C dan pada masa rendaman 2 jam. Untuk ABKS, ketumpatan pukal (2.45 g/cm³) dan kekuatan mampatan (46 MPa) tertinggi telah dicapai pada 15% berat ABKS pada suhu sinter 1100 °C dan masa rendaman 2 jam. Untuk kombinasi ASP dan POFA, ketumpatan pukal (2.43 g/cm³) dan kekuatan mampatan (45 MPa) dicatatkan pada 20% berat ASP dan ABKS pada suhu sinter 1200 °C dan pada masa rendaman 2 jam. Ini dapat diperhatikan bahawa peningkatan dalam suhu sinter, tekanan acuan dan merendam masa meningkatkan mikrostruktur. Kesimpulannya, sampel yang mengandungi ABKS mencapai peringkat pengkacaan pada suhu yang lebih rendah dan mempamerkan sifat mekanikal yang lebih tinggi. Pengukuran pengembangan dan kekonduksian terma adalah beberapa bidang yang boleh diterokai untuk penyelidikan selanjutnya.

TABLE OF CONTENT

	SUI	PERVISO (OR'S DECLARATION	ii
	TIT	LE		iii
	STU	JDENT'S	S DECLARATION	iv
	DE	DICATI	ION	v
	AC	KONOW	VLEDGEMENT	vi
	ABS	STRACT	Γ	viii
	ABS	SRAK		ix
	TAI	BLE OF	CONTENT	X
	LIS	T OF FI	GURES	xvi
	LIS	T OF TA	ABLES	xxvii
	LIS	T OFAB	BBREVIATIONS	xxix
CHAPTER 1	INT	RODUC	BBREVIATIONS	1
	1.1	Backgro	ound of the study	1
	1.2	Problen	n statement	3
	1.3	Objecti	ves of the study	5
	1.4	Scope of	of the study	5
CHAPTER 2	LIT	ERATU	RE REVIEW	7
	2.1	Introdu	ction on porcelain ceramic tile	7
		2.1.1	Raw materials of porcelain	7
		2.1.2	Strength consideration	9
		2.1.3	Sinteing of porcelain	10
		2.1.4	Sintering stages	12
	2.2	Charact	terisation	27
		2.2.1	X-ray diffraction	27
		2.2.2	Microstructure observation	32
	2.3	Physica	al properties	37

				xii
		2.3.1	Shrinkage	37
		2.3.2	Percentage of porosity	42
		2.3.3	Bulk density	45
	2.4	Mecha	anical properties	50
		2.4.1	Bending strength	50
		2.4.2	Vickers hardness	55
		2.4.3	Compressive strength	57
	2.5	Rice h	nusk ash (RHA)	59
		2.5.1	High silica content with amorphous	
			characteristics	59
		2.5.2	Disposal problems	60
	2.6	Origin	of palm oil	61
		2.6.1	Palm oil fuel ash (POFA)	62
		2.6.2	Environmental issues	64
		2.6.3	Summary	65
CHAPTER 3	RES	EARC	H METHODOLOGY	66
	3.1		uction	66
	3.1		rch Design	68
	3.2	3.2.1	Effects of substitution of quartz by RHA	68
			Effects of substitution of quartz by POFA	68
		3.2.3		00
		3.2.3	combination of RHA and POFA	69
	3.3	Exper	imental design variables	69
		3.3.1	Experimental parameters	69
	3.4	Prepar	ration of raw materials	78
		3.4.1	RHA	78
		3.4.2	POFA	79
	3.5	Samp	le preparation	79
		3.5.1	Sample composition	79
		3.5.2	Mixing	80
		3.5.3	Dry pressing	80
		3.5.4	Drying	81
			· ·	

				xiii
	3.6	Physic	cal measurement	81
		3.6.1	Volume shrinkage	81
		3.6.2	Porosity and bulk density	82
	3.7	X-ray	Fluorescence characterisation measurement	83
		3.7.1	X-ray diffraction (XRD) measurement	83
	3.8	Scann	ing electron microscopy (SEM)	84
	3.9	Mecha	anical measurement	84
		3.9.1	Vickers hardness	84
		3.9.2	Bending strength	85
		3.9.3	Compressive strength	86
	3.10	Partic	le size analysis	87
	The	powde	r was diluted using water, the solution was plced	
		into tl	he machine, the machine stirred the solution for 5	
		to 7 r	minutes in a chamber having two lasers. With the	
		help	of a formula built in the machine the sizes of the	
		particl	es were analysed	87
	3.11	Etchin	ag	87
CHAPTER 4			AND DISCUSSIONS	88
	4.1	Introd	x-ray fluorescence (XRF) measurement X-ray diffraction (XRD) analysis	88
	4.2	Analy	sis of raw materials	88
		4.2.1	X-ray fluorescence (XRF) measurement	88
		4.2.2	X- ray diffraction (XRD) analysis	89
		4.2.3	Particle size analysis	91
			4.2.3.1 Porcelain raw material	91
			4.2.3.2 RHA	91
			4.2.3.3 POFA	91
		4.2.4	Scanning electron microscopy (SEM)	95
	4.3	Phase	I: Study of RHA	96
		4.3.1	Effects of sintering temperature, mould	
			pressure and soaking on volume shrinkage	96
		4.3.2	Effects of sintering temperature, mould	
			pressure and soaking time on percentage	

	4.3.3	Effects of sintering temperature, mould	
		pressure and soaking time on bulk density	103
	4.3.4	Effects of RHA, sintering temperature, mould	
		pressure and soaking time on Vickers hardness	106
	4.3.5	Effects of sintering temperature, mould	
		pressure and soaking time on bending strength	109
	4.3.6	Effects of sintering temperature, mould	
		pressure and soaking time on compressive	
		strength	112
	4.3.7	Effects of sintering temperature, mould	
		pressure and soaking time on XRD	116
	4.3.8	Effects of sintering temperature, mould	
		pressure and soaking time on microstructure	119
	4.3.9	Summary	122
4.4	Phase	II Study of POFA, sintering temperature, mould	
	pressu	re and soaking	123
	4.4.1	Effects of sintering temperature, mould	
		pressure and soaking time on volume	
		shrinkage	123
	4.4.2	Effects of sintering temperature, mould	
		pressure and soaking time on percentage of	
		porosity	125
	4.4.3	Effects of sintering temperature and soaking	
		time on bulk density	128
	4.4.4	Effects of sintering temperature and soaking	
		time on Vickers hardness	132
	4.4.5	Effects of sintering temperature and soaking	
		time on bending strength	135
	4.4.6	Effects of sintering temperature and soaking	
		time on compressive strength	139
	4.4.7	Effects of sintering temperature and soaking	
		time on XRD	142
	4.4.8	Effects of sintering temperature and soaking	
		time on microstructure	145

		4.4.9	Summary	148
	4.5	Phase	III: Study of RHA and POFA	150
		4.5.1	Effects of sintering temperature and soaking	
			time on volume shrinkage	150
		4.5.2	Effects of sintering temperature and soaking	
			time on percentage of porosity	152
		4.5.3	Effects of sintering temperature and soaking	
			time on bulk density	155
		4.5.4	Effects of sintering temperature and soaking	
			time on Vickers hardness	158
		4.5.5	Effects of sintering temperature and soaking	
			time on bending strength	161
		4.5.6	Effects of sintering temperature and soaking	
			time on compressive strength	164
		4.5.7	Effects of sintering temperature and soaking	
			time on XRD	167
		4.5.8	Effects of sintering temperature and soaking	
			time on microstructure	171
		4.5.9	Summary	173
		4.5.10	Relationship between percentage of porosity	
			and compressive strength	174
		4.5.11	Discussion on comparison between samples	
			containing RHA, POFA and the combination	
			of RHA and POFA	177
CHAPTER 5	CON	NCLUS	ION AND RECOMMENDATIONS	181
	5.1	Conclu	usion	181
	5.2	Recon	nmendation	183
REFERENCES	}			184

LIST OF FIGURES

Figure 2.1:	Stages of sintering (a) free particles, (b) necking between	
	particles, (c) formation of grain boundary, and (d) densification	
	process and pores elimination (Randal, 1991)	13
Figure 2.2:	XRD curves of fired body mixes (Prasad et al., 2001)	28
Figure 2.3:	XRD results of porcelains, sintered at 1350 °C and 1400 °C.	
	(Karamanov et al., 2006)	29
Figure 2.4:	X-ray diffraction patterns of POFA20 tile and sintered at	
	different temperatures (Tonnayopas et al., 2009)	30
Figure 2.5:	X-ray diffraction patterns: (a) 20G, 1150 °C; (b) 15G, 1200 °C;	
	(c) 20F10G, 1200 $^{\circ}$ C; and (d) 25 F5G, 1220 $^{\circ}$ C (Luz and	
	Ribeiro, 2007)	31
Figure 2.6:	X-ray diffractogram of the fired porcelain body at different	
	mould pressure (MP) (Pérez et al., 2012)	32
Figure: 2.7	SEM images of the pellets after (a) 1000°C, (b) 1100°C, (c)	
	1200°C, (d)1300°C, (e) 1400°C	33
Figure 2.8:	Backscatered image of the fractured surfaces of POFA-GFQW	
	tiles (Tonnayopas et al., 2009)	34
Figure 2.9:	SEM of light weight porcelain stoneware Garcia-Ten et al.,	
	(2012)	35
Figure 2.10:	SEM observations on polished surfaces of the porcelain	
	stoneware (Martín-Márquez et al., 2010a)	36
Figure 2.11:	SEM micrographs on polished surfaces of the porcelain	
	stoneware at different mould pressures: 10 MPa (b) 20 MPa,	
	(c) 30 MPa and (d) 30 MPa (Preze et al., 2012)	37

		xvii
Figure 2.12:	Effect of RHA+SF on the fired shrinkage of body mixes fired at	
	different test temperature (Prasad et al., 2003)	38
Figure 2.13:	Linear shrinkage % of the understudied compositions as	
	function of maximum firing temperature (Leonelli et al., 2001)	39
Figure 2.14:	Linear shrinkage as function of temperature (Martín-Márquez	
	et al., 2008)	40
Figure 2.15:	Linear shrinkage as a function of the firing temperatures for	
	the tested samples (Tucci et al., 2007)	40
Figure 2.16:	Firing shrinkage as function of firing temperature (de'Gennaro	
	et al., 2003)	41
Figure 2.17:	Dilatometric sintering curves of C0 (standard sample) and C1	
	(sample with cathode ray TV) compositions (de'Gennaro et al.,	
	2003)	42
Figure 2.18:	Influence of firing temperature on porosity and irreversible	
	stain retention (evaluated as E) for samples obtained from the	
	as-received industrial powder (Sanchez et al. 2001).	43
Figure 2.19:	Total porosity (PT), closed porosity (PC), and open porosity	
	(PO) porosities behaviour vs. temperature for laboratory fired	
	standard sample (C0). (Andreola et al. 2008)	43
Figure 2.20:	The variations of total porosity as function of soaking time for	
	STD milled at 8 h and sintered at three different temperatures.	
	(Salem et al., 2010)	44
Figure 2.21:	Apparent porosity as a function of temperature (Dana et al.,	
	2004).	45
Figure 2.22:	Bulk densities of porcelain compositions (Kamseu et al., 2007)	46
Figure 2.23:	Bulk density versus oil palm fuel ash POFA content	
	(Tonnayopas et al., 2009)	46
Figure 2.24:	Absolute and bulk densities of green tiles at different MP	
	(Pérez et al., 2011)	47
Figure 2.25:	Variation in bulk density of C1 and C2 bodies with heating	
	temperature (Das and Dana, 2003)	48
Figure 2.26:	Bulk density versus the sintering temperature. NC: Samples	
	without calcinations, C: calcined samples (Kitouni and Harabi,	
	2011)	48

Figure 2.27:	Effect of MP and firing temperature on the maximum density	
	(Youssef and Ghazal, 2011)	49
Figure 2.28:	Bending strength of porcelain compositions (Kamseu et al.,	
	2007)	50
Figure 2.29:	Effect of RHA on the fired M.O.R. of body mixes fired at	
	different test temperatures. (Prasad et al., 2001)	51
Figure 2.30:	Bending strength of porcelain compositions (Kamseu et al.,	
	2007)	52
Figure 2.31:	Bending strength versus oil palm fuel ash (OPFA) content	
	(Tonnayopas et al., 2009)	53
Figure 2.32:	Flexural strength and tensile and versus sintering temperature	
	for calcined porcelain samples (Kitouni and Harabi, 2011)	54
Figure 2.33:	Effect of temperature and soaking time on the MOR of	
	samples (Youssef and Ghazal, 2011)	54
Figure 2.34:	Vickers hardness versus POFA content (Tonnayopas et al.,	
	2009)	55
Figure 2.35:	Effect of MP and soaking time on Rockwell Hardness for	
	samples fired at 1250 °C (Youssef and Ghazal, 2011)	56
Figure 2.36:	Variation of Vickers hardness with Weight % RHA (Prasad	
	and Krishna, 2011)	56
Figure 2.37:	Vickers hardness versus RHA content (Fadzil et al., 2008)	57
Figure 2.38:	Effect of firing temperature and soaking time on the cold	
	crushing strength of samples (Youssef and Ghazal, 2011)	58
Figure 2.39:	Dry compressive strength of fired briquettes as a function of	
	OBM content as well as firing temperatures (El-mahllawy and	
	Osman, 2010)	58
Figure 2.40:	Map of the largest palm oil producing countries in the world	
	(Zuur, 2004)	61
Figure 33.1:	Overall flow Chart for the experiment	67
Figure 4.1:	The XRD patterns of raw materials (where c = cristobalite, cc =	
	calcite, $k = kaolin$, $m = mullite$, $mu = muscovite$, $p =$	
	portlandite, $q = quartz$)	90
Figure 4.2:	Result of particle size analysis of porcelain raw material	92
Figure 4.3:	Result of particle size analysis of RHA	93

		xix
Figure 4.4:	Result of particle size analysis of POFA	94
Figure 4.5:	Surface morphology of (a) RHA (b) POFA (c) ceramic	
	(porcelain) raw material. Phase 1: Substitution of quartz by	
	RHA	95
Figure 4.6:	Effect of temperature on shrinkage of the samples with	
	different percentage of RHA (mould pressure =91 MPa,	
	soaking time =2 hours)	99
Figure 4.7:	Effect of mould pressure on shrinkage of the samples with	
	different percentage of RHA (temperature =1200 °C, soaking	
	time =2 hours)	99
Figure 4.8:	Effect of soaking time shrinkage of the samples with different	
	percentage of RHA (temperature =1200 °C, mould pressure =	
	91 MPa)	99
Figure 4.9:	Effect of temperature on percentage of porosity of the samples	
	with different percentage of RHA (mould pressure =91 MPa,	
	soaking time =2 hours)	101
Figure 4.10:	Effect of mould pressure on percentage of porosity the samples	
	of sample with different percentage of RHA (temperature	
	=1200 °C, soaking time =2 hours)	101
Figure 4.11:	Effect of soaking time on percentage of porosity of the samples	
	with different percentage of RHA (temperature =1200 °C,	
	mould pressure = 91 MPa)	101
Figure 4.12:	Effect of temperature on bulk density of the samples with	
	different percentage of RHA (mould pressure =91 MPa,	
	soaking time =2 hours)	104
Figure 4.13:	Effect of mould pressure on bulk density of the samples with	
	different percentage of RHA (temperature =1200 °C, soaking	
	time =2 hours)	104
Figure 4.14:	Effect of soaking time on bulk density of body mixes with	
	different percentage of RHA (temperature =1200 °C, mould	
	pressure = 91 MPa)	104
Figure 4.15:	Effect of temperature on Vickers micro-hardness of sample	
	with different percentage of RHA (mould pressure =91 MPa,	
	soaking time =2 hours)	108

Figure 4.16:	Effect of mould pressure on hardness of the samples with	
	different percentage of RHA (temperature =1200 °C, soaking	
	time =2 hours)	108
Figure 4.17:	Effect of soaking time on Vickers hardness of the samples with	
	percentage of RHA (temperature =1200 °C, mould pressure =	
	91 MPa)	108
Figure 4.18:	Effect of temperature on bending strength of the samples with	
	different percentage of RHA (mould pressure =91 MPa,	
	soaking time =2 hours)	111
Figure 4.19:	Effect of mould pressure on bending strength of the samples	
	with different percentage of RHA (temperature =1200 °C,	
	soaking time =2 hours)	111
Figure 4.20:	Effect of soaking time on bending strength of the samples with	
	different percentage of RHA (temperature =1200 °C, mould	
	pressure = 91 MPa)	111
Figure 4.21:	Effect of temperature on compressive strength of the samples	
	with different percentage of RHA (mould pressure =91 MPa,	
	soaking time =2 hours)	114
Figure 4.22:	Effect of mould pressure on compressive strength of the	
	samples with different percentage RHA (temperature =1200	
	°C, soaking time =2 hours)	114
Figure 4.23:	Effect of soaking time compressive strength of the samples	
	with different percentage of RHA (temperature =1200 °C,	
	mould pressure = 91 MPa)	114
Figure 4.24:	The XRD curves of the samples containing 20 wt% of RHA	
	sintered at different temperatures (mould pressure =91 MPa,	
	soaking time =2 hours)	116
Figure 4.25:	The XRD curves of sintered sample containing 20 wt% RHA	
	pressed at different mould pressure (temperature =1200 °C,	
	soaking time =2 hours)	117
Figure 4.26:	The XRD curves of the samples containing 20 wt% RHA	
	sintered at different soaking times (temperature =1200 °C,	
	mould pressure = 91 MPa)	118

Figure 4.27:	The SEM of the samples containing 20 wt% of RHA sintered	
	at different temperatures of (a) 1000 °C (b) 1100 °C (c) 1200	
	°C (d) 1300 °C. All micrograph were taken with 1000X	
	magnification (mould pressure =91 MPa, soaking time =2	
	hours)	119
Figure 4.28:	SEM of the samples sintered containing 20 wt% RHA pressed	
	at mould pressure of (a) 31 MPa (b) 61 MPa (c) 91 MPa (d)	
	121MPa: All micrographs were taken with1000X	
	magnification	120
Figure 4.29:	SEM of the samples sintered containing 20 wt% RHA sintered	
	at a soaking time of (a) 1 Hour (b) 2 Hours (c) 3 Hours. All	
	micrograph were taken with 1000X magnification (temperature	
	=1200 °C, mould pressure = 91 MPa)	121
Figure 4.30:	Effect of temperature on shrinkage of the samples with	
	different percentage of POFA (mould pressure =91 MPa,	
	soaking time =2 hours)	124
Figure 4.31:	Effect of mould pressure on shrinkage of the samples with	
	different percentage of POFA (temperature =1100 °C, soaking	
	time =2 hours)	124
Figure 4.32:	Effect of soaking time of shrinkage of body mixes with	
	different percentage of POFA (temperature =1100 °C, mould	
	pressure = 91 MPa)	124
Figure 4.33:	Effect of temperature on percentage of porosity of the samples	
	with different percentage of POFA wt% (mould pressure =91	
	MPa, soaking time =2 hours)	127
Figure 4.34:	Effect of mould pressure on percentage of the samples of	
	sample with different percentage of POFA (temperature =1100	
	°C, soaking time =2 hours)	127
Figure 4.35:	Effect of soaking time of percentage of porosity of the samples	
	with different percentage of POFA (temperature =1100 °C,	
	mould pressure = 91 MPa)	127
Figure 4.36:	Effect of temperature on bulk density of the samples with	
	different percentage of POFA (mould pressure =91 MPa,	
	soaking time =2 hours)	130

Figure 4.37:	Effect of mould pressure on bulk density of the samples with	
	different percentage of POFA (temperature =1100 °C, soaking	
	time =2 hours)	130
Figure 4.38:	Effect of soaking time on bulk density of the samples with	
	different percentage of POFA (temperature =1100 °C, mould	
	pressure = 91 MPa)	130
Figure 4.39:	Effect of temperature on Vickers hardness of the samples with	
	different percentage of POFA (mould pressure =91 MPa,	
	soaking time =2 hours)	134
Figure 4.40:	Effect of mould pressure on Vickers hardness of the samples	
	with different percentage of RHA (temperature =1100 °C,	
	soaking time =2 hours)	134
Figure 4.41:	Effect of soaking time on Vickers hardness of the the samples	
	with different percentage of POFA (temperature =1100 °C,	
	mould pressure = 91 MPa)	134
Figure 4.42:	Effect of temperature on bending strength of body the samples	
	different percentage of POFA (mould pressure =91 MPa,	
	soaking time =2 hours)	137
Figure 4.43:	Effect of mould pressure on bending strength of the samples	
	with different percentage of POFA (temperature =1100 °C,	
	soaking time =2 hours)	137
Figure 4.44:	Effect of soaking time on bending strength of the the samples	
	with different percentage of POFA (temperature =1100 °C,	
	mould pressure = 91 MPa)	137
Figure 4.45:	Effect of temperature on compressive strength of the samples	
	with different percentage of POFA (mould pressure =91 MPa,	
	soaking time =2 hours)	140
Figure 4.46:	Effect of mould pressure on compressive strength of the	
	samples with	140
Figure 4.47:	Effect of soaking time on compressive strength of the samples	
	with different percentage of POFA (temperature =1100 °C,	
	mould pressure = 91 MPa)	140

Figure 4.48:	The XRD curves of the samples containing 15 wt% of POFA	
	sintered at different temperatures (mould pressure =91 MPa,	
	soaking time =2 hours)	143
Figure 4.49:	The XRD curves of the samples containing 15 wt% POFA	
	pressed at different mould pressure (temperature =1100 °C,	
	soaking time =2 hours)	144
Figure 4.50:	The XRD curves of the samples containing 15 wt% POFA	
	sintered at different soaking times ((temperature =1100 °C,	
	mould pressure = 91 MPa)	145
Figure 4.51:	SEM of the sample containing 15 wt% POFA sintered at a	
	temperature of (a) $1000^{\rm o}$ C (b) $1100^{\rm o}$ C (c) $1200^{\rm o}$ C (c) $1280^{\rm o}$	
	°C. All micrograph were taken with 1000X magnification	
	(mould pressure =91 MPa, soaking time =2 hours)	146
Figure 4.52:	SEM of the samples sintered containing 15 wt% POFA pressed	
	at pressed at mould pressure of (a) 31 MPa (b) 61 MPa (c) 91	
	MPa (d) 121 MPa. All micrograph were taken with 1000X	
	magnification (temperature =1100 °C, soaking time =2 hours)	147
Figure 4.53:	SEM of the samples containing 15 wt% of POFA sintered at a	
	soaking time of (a) 1hour (b) 2hours (c) 3hours. 1000X	
	magnification (temperature =1100 °C, mould pressure = 91	
	MPa)	148
Figure 4.54:	Effect of temperature on shrinkage of the samples with	
	different percentage of RHA and POFA (mould pressure =91	
	MPa, soaking time =2 hours)	151
Figure 4.55:	Effect of mould pressure on shrinkage of the samples with	
	different percentage of RHA and POFA (temperature =1200	
	°C, soaking time =2 hours)	151
Figure 4.56:	Effect of soaking time on shrinkage of the samples with	
	different percentage of RHA and POFA (temperature =1200	
	°C, mould pressure =91 MPa)	151
Figure 4.57:	Effect of temperature on pecentage of porosity of the samples	
	with different percentage of RHA and POFA (mould pressure	
	=91 MPa, soaking time =2 hours)	154

Figure 4.58:	Effect of mould pressure on percentage of the samples of	
	sample with different percentage of RHA and POFA wt%	
	(temperature =1200 °C, soaking time =2 hours)	154
Figure 4.59:	Effect of soaking time on the porosity of of the samples with	
	different percentage of RHA and POFA (temperature =1200	
	°C, mould pressure =91 MPa)	154
Figure 4.60:	Effect of temperature on bulk density of the samples with	
	different percentage of RHA and POFA (mould pressure =91	
	MPa, soaking time =2 hours)	157
Figure 4.61:	Effect of mould pressure on bulk density of the samples with	
	different percentage of RHA and POFA (temperature =1200	
	°C, soaking time =2 hours)	157
Figure 4.62:	Effect of soaking time on the bulk density of the samples with	
	different percentage of RHA and POFA (temperature =1200	
	°C, mould pressure =91 MPa)	157
Figure 4.63:	Effect of temperature on Vickers micro-hardness of sample	
	with different percentage of RHA (mould pressure =91 MPa,	
	soaking time =2 hours)	159
Figure 4.64:	Effect of mould pressure on Vickers hardness of the samples	
	with different percentage of RHA and POFA (temperature	
	=1200 °C, soaking time =2 hours)	159
Figure 4.65:	Effect of soaking time on Vickers hardness of the samples	
	with different percentage of RHA and POFA (temperature	
	=1200 °C, mould pressure = 91 MPa)	159
Figure 4.66:	Effect of temperature on bending strength of the samples with	
	different percentage of RHA and POFA (mould pressure =91	
	MPa, soaking time =2 hours)	163
Figure 4.67:	Effect of mould pressure bending strength of the samples with	
	different percentage of RHA and POFA (temperature =1200	
	°C, soaking time =2 hours)	163
Figure 4.68:	Effect of soaking time on bending strength of the samples with	
	different percentage of RHA and POFA (temperature =1200	
	°C, mould pressure = 91 MPa)	163

Figure 4.69:	Effect of temperature on compressive strength of the samples	
	with different percentage of RHA and POFA (mould pressure	
	=91 MPa, soaking time =2 hours)	165
Figure 4.70:	Effect of mould pressure on compressive strength of the	
	samples with different percentage of RHA and POFA	
	(temperature =1200 °C, soaking time =2 hours)	165
Figure 4.71:	Effect of soaking time on com pressive of the samples with	
	different percentage of RHA and POFA (temperature =1200	
	°C, mould pressure = 91 MPa)	165
Figure 4.72:	The XRD curves of the samples containing 20wt% of RHA	
	and POFA sintered at different temperatures (mould pressure	
	=91 MPa, soaking time =2 hours)	168
Figure 4.73:	The XRD curves of the samples containing 20 wt% of RHA	
	and POFA sintered at different mould pressure (temperature	
	=1200 °C, soaking time =2 hours)	169
Figure 4.74:	The XRD curves of the samples containing 20 wt% of RHA	
	and POFA sintered at different soaking time (temperature	
	=1200 °C, mould pressure =91 MPa)	170
Figure 4.75:	SEM of the samples containing 20 wt% RHA and POFA	
	sintered at a temperature of (a) 1000°C (b) 1200°C (c) 1300°C.	
	All micrograph were taken with 1000X magnification (mould	
	pressure =91 MPa, soaking time =2 hours)	171
Figure 4.76:	SEM of the samples containing 20 wt% RHA and POFA	
	pressed at a mould pressure of (a) 31 MPa (b) 61 MPa (c) 91	
	MPa (d) 121 MPa. All micrograph were taken with 1000X	
	magnification (temperature =1200 °C, soaking time =2 hours)	172
Figure 4.77:	SEM of the samples containing RHA and POFA sintered at a	
	soaking times (a) 1 hour (b) 2 hours (c) 3 hours. All	
	micrograph were taken with 1000X magnification (temperature	
	=1200 °C, mould pressure =91 MPa)	173
Figure 4.78:	Relationship between the compressive strength and percentage	
	of porosity (RHA wt %)	175
Figure 4.79:	Relationship between the compressive strength and percentage	
	of porosity (POFA wt %)	176

Figure 4.80: Relationship between the compressive strength and percentage of porosity (RHA and POFA wt %)

177



LIST OF TABLES

Table 3.1:	RHA wt% Vs temperature	/(
Table 3.2:	RHA wt% Vs mould pressure	71
Table 3.3:	RHA wt% Vs soaking time	72
Table 3.4:	POFA wt% Vs temperature	73
Table 3.5:	POFA wt% Vs mould pressure	74
Table 3.6:	POFA wt% Vs soaking time	75
Table 3.7:	RHA and POFA wt% Vs temperature	76
Table 3.8:	RHA and POFA wt% Vs mould pressure	77
Table 3.9:	RHA and POFA wt% Vs soaking time	78
Table 3.10:	The composition with the substitution of quartz by RHA (wt %)	79
Table 3.11:	The composition with the substitution of quartz by POFA (wt %)	80
Table 3.12:	The composition with the substitution of quartz by RHA and	
	POFA (wt %)	80
Table 4.1:	X-Ray Fluorescence (XRF) Analysis	89
Table 4.2:	XRD quantitative analysis of the samples containing 20 wt%	
	RHA sintered at different temperatures (mould pressure =91	
	MPa, soaking time =2 hours)	116
Table 4.3:	XRD quantitative analysis of the samples containing 20 wt%	
	RHA pressed at different mould pressure (temperature =1200	
	°C, soaking time =2 hours)	117
Table 4.4:	XRD quantitative analysis of the samples containing 20wt%	
	RHA sintered at different soaking times (temperature =1200	
	°C, mould pressure = 91 MPa)	118

		xxviii
Table 4.5:	XRD quantitative analysis of the samples containing 15 wt% of	
	POFA sintered at different temperatures (mould pressure =91	
	MPa, soaking time =2 hours)	143
Table 4.6:	XRD quantitative analysis of the samples containing 15 wt%	
	POFA pressed at different mould pressure (temperature =1100	
	°C, soaking time =2 hours)	144
Table 4.7:	XRD quantitative analysis of the samples containing 15 wt%	
	POFA sintered at different soaking times (temperature =1100	
	°C, mould pressure = 91 MPa)	145
Table 4.8:	XRD quantitative analysis of the samples containing 20 wt% of	
	RHA and POFA sintered at different temperatures (mould	
	pressure =91 MPa, soaking time =2 hours)	168
Table 4.9:	XRD quantitative analysis of the samples containing 20 wt% of	
	RHA and POFA pressed at different mould pressure	
	(temperature =1200 °C, soaking time =2 hours)	169

Table 4.10: XRD quantitative analysis of the samples containing 20 wt%

PERPUSTAKAAN

RHA and POFA sintered at different soaking times

(temperature =1200 °C, mould pressure =91 MPa)

170

REFERENCES

- Abadir, M. F., Sallam, E. H., & Bakr, I. M. (2002). Preparation of porcelain tiles from Egyptian raw materials, 28, 303–310.
- Abbas, A., & Ansumali, S. (2010). Global potential of rice husk as a renewable feedstock for ethanol biofuel production. *BioEnergy Research*, *3*(4), 328-334.
- Abdul Awal, A. S. M., & Warid Hussin, M. (2011). Effect of palm oil fuel ash in controlling heat of hydration of concrete. Procedia Engineering, 14, 2650-2657.
- Afroz, R., Hassan, M. N., & Ibrahim, N. A. (2003). Review of air pollution and health impacts in Malaysia. *Environmental Research*, 92(2), 71-77.
- Ahmad, M. H., Omar, R. C., Malek, M. A., Noor, N., & Thiruselvam, S. (2008). Compressive strength of palm oil fuel ash concrete. In *In International Conference on Construction and Building Technology (ICCBT)*.
- Ahmadi, M. A., Alidoust, O., Sadrinejad, I., & Nayeri, M. (2007). Development of mechanical properties of self compacting concrete contain rice husk ash. *International Journal of Computer, Information, and Systems Science, and Engineering*, 1(4), 259-262.
- Ahmed, A. E., & Adam, F. (2009). The benzylation of benzene using aluminium, gallium and iron incorporated silica from rice husk ash. *Microporous and Mesoporous Materials*, 118(1), 35-43.
- Ajiwe, V. I. E., Okeke, C. A., & Akigwe, F. C. (2000). A preliminary study of manufacture of cement from rice husk ash. *Bioresource Technology*, 73(1), 37-39.
- Al-Hilli, M. F., & Al-Rasoul, K. T. (2013). Characterization of alumino-silicate glass/kaolinite composite. *Ceramics International*.
- Almeida, M. I., Almeida, N. G., Carvalho, K. L., Gonçalves, G. A. A., Silva, C. N., Santos, E. A., ... & Vargas, E. A. (2012). Co-occurrence of aflatoxins B1, B2, G1 and G2, ochratoxin A, zearalenone, deoxynivalenol, and citreoviridin in rice in Brazil. *Food Additives & Contaminants: Part A*, 29(4), 694-703.
- Altwair, N. M., Johari, M. A. M., & Hashim, S. F. S. (2011). Strength activity index and microstructural characteristics of treated palm oil fuel ash. structure, 5, 6.

- Altwair, N. M., Johari, M. A., Hashim, S. F. S., & Zeyad, A. M. (2013). Mechanical Properties of Engineered Cementitious Composite with Palm Oil Fuel Ash as a Supplementary Binder. *Advanced Materials Research*, 626, 121-125.
- Alves, H. J., Freitas, M. R., Melchiades, F. G., & Boschi, A. O. (2011). Dependence of surface porosity on the polishing depth of porcelain stoneware tiles. *Journal of the European Ceramic Society*, 31(5), 665-671.
- Amick, J. A. (1982). Purification of rice hulls as a source of solar grade silicon for solar cells. *Journal of the Electrochemical Society*, 129(4), 864-866.
- Amick, J.A. (1985). Purification of Rice Hulls as a Source of Solar Grade Silicon for Solar Cells. Journal of Elecrochemical Society, 34,pp. 269-273.
- Amorós, J. L., Orts, M. J., García-Ten, J., Gozalbo, A., & Sánchez, E. (2007). Effect of the green porous texture on porcelain tile properties. *Journal of the European Ceramic society*, 27(5), 2295-2301.
- Amrutkar, P. P., Patil, S. B., Todarwal, A. N., Wagh, M. A., Kothawade, P. D., & Surawase, R. K. (2011). Design and evaluation of taste masked chewable dispersible tablet of lamotrigine by melt granulation. *International Journal of Drug Delivery*, 2(2).
- An, D., Guo, Y., Zou, B., Zhu, Y., & Wang, Z. (2011). A study on the consecutive preparation of silica powders and active carbon from rice husk ash. *biomass and bioenergy*, 35(3), 1227-1234.
- Andreola, F., Barbieri, L., Karamanova, E., Lancellotti, I., & Pelino, M. (2008). Recycling of CRT panel glass as fluxing agent in the porcelain stoneware tile production. *Ceramics International*, 34(5), 1289-1295.
- Armesto, L., Bahillo, A., Veijonen, K., Cabanillas, A., & Otero, J. (2002). Combustion behaviour of rice husk in a bubbling fluidised bed. *Biomass and Bioenergy*, 23(3), 171-179.
- Awal A.S. and Hussain M.W.(1997). The effectiveness of palm oil fuel ash in preventing expansion due to aikali-silica reaction. Cement Concrete Com, 19(4), pp. 96-105
- Awal, A. S. M. A., & Nguong, S. K. (2010). A Short-Term Investigation on High Volume Palm Oil Fuel Ash (POFA) Concrete. *Proceedings of the 35th Conference on our World in Concrete and Structure*, 185-192.
- Bansal, V., Ahmad, A., & Sastry, M. (2006). Fungus-mediated biotransformation of amorphous silica in rice husk to nanocrystalline silica. *Journal of the American Chemical Society*, *128*(43), 14059-14066.
- Basegio, T., Haas, C., Pokorny, A., Bernardes, A. M., & Bergmann, C. P. (2006). Production of materials with alumina and ashes from incineration of chromium tanned leather shavings: environmental and technical aspects. *Journal of hazardous materials*, 137(2), 1156-1164.

- Beheri, H. H., Mohamed, K. R., & El-Bassyouni, G. T. (2012). Mechanical and microstructure of reinforced hydroxyapatite/calcium silicate nano-composites materials. Materials & Design.
- Bondioli, F., Andreola, F., Barbieri, L., Manfredini, T., & Ferrari, A. M. (2007). Effect of rice husk ash (RHA) in the synthesis of (Pr, Zr) SiO< sub> 4</sub> ceramic pigment. Journal of the European Ceramic Society, 27(12), 3483-3488.
- Bouzidi, N., Bouzidi, A., Gaudon, P., Merabet, D., & Blanchart, P. (2013). Porcelain containing anatase and rutile nanocrystals. Ceramics International, 39(1), 489-495.
- Bragança, S. R., Vicenzi, J., Guerino, K., & Bergmann, C. P. (2006). Recycling of iron foundry sand and glass waste as raw material for production of whiteware. *Waste management & research*, 24(1), 60-66.
- Bragança, S.R. and Bergmann, C.P. (2003) "A View of Whitewares Mechanical strength and microstructure" Ceramics International 29, pp 801-806.
- Bragança, S.R. and Bergmann, C.P. (2004) "Traditional and glass powder porcelain: Technical and Microstructure analysis" Journal of the European Ceramic Society 24, pp 2383-2388.
- Bribiesca, S., Equihua, R., & Villaseñor, L. (1999). Photoacoustic thermal characterization of electrical porcelains: effect of alumina additions on thermal diffusivity and elastic constants. *Journal of the European Ceramic Society*, 19(11), 1979-1985.
- Brindley, G. W., & Nakahira, M. (1959). The KaoIinite-Mullite Reaction Series: I, A Survey of Outstanding Problems. *Journal of the American Ceramic Society*, 42(7), 311-314.
- Buchanan, R.C. (1991) Ceramic Materials for Electronics (R.C. Buchanan, Ed.) Chap. 1. Dekker, New York.
- Bui, D. D., Hu, J., & Stroeven, P. (2005). Particle size effect on the strength of rice husk ash blended gap-graded Portland cement concrete. *Cement and Concrete Composites*, 27(3), 357–366.
- Callister, D. W. And Rethwich G.D. (2008). Fundamentals of Materials Science and Engineering 3rd ed. U.S.A.,: John Wiley & Sons, Inc.
- Carty, W. M., & Senapati, U. (1998). Porcelain—raw materials, processing, phase evolution, and mechanical behavior. *Journal of the American Ceramic Society*, 81(1), 3-20.
- Celik, H. (2011). Effect of spray-dried powder granularity on porcelain tile properties. *Journal of Ceramic Processing Research*, 12(4), 483-487.
- Chakraverty, A., Mishra, P., & Banerjee, H. D. (1988). Investigation of combustion of raw and acid-leached rice husk for production of pure amorphous white silica. *Journal of Materials Science*, 23(1), 21-24.

- Chandara, C., Mohd Azizli, K. A., Ahmad, Z. A., Saiyid Hashim, S. F., & Sakai, E. (2012). Heat of hydration of blended cement containing treated ground palm oil fuel ash. *Construction and Building Materials*, 27(1), 78-81.
- Chandara, C., Sakai, E., Azizli, K. A. M., Ahmad, Z. A., & Hashim, S. F. S. (2010). The effect of unburned carbon in palm oil fuel ash on fluidity of cement pastes containing superplasticizer. *Construction and Building Materials*, 24(9), 1590–1593.
- Chandrasekhar, S.; Satyanarayana, K. G.; Pramada, P. N.; Raghavan, P. and Gupta, T. N. (2003). Review Processing, Properties and Application of Reactive Silica from Rice Husk An Overview. *Journal of Materials Science*. 38: 3159-3168.
- Chatveera, B., & Lertwattanaruk, P. (2011). Durability of conventional concretes containing black rice husk ash. *Journal of environmental management*, 92(1), 59–66.
- Chaudhry, A. A., Yan, H., Gong, K., Inam, F., Viola, G., Reece, M. J., ... & Darr, J. A. (2011). High-strength nanograined and translucent hydroxyapatite monoliths via continuous hydrothermal synthesis and optimized spark plasma sintering. *Acta Biomaterialia*, 7(2), 791-799.
- Chavalparit, O. (2006a). Clean technology for the crude palm oil industry in Thailand. Wageneningen University.
- Chavalparit, O., Rulkens, W. H., Mol, A. P. J., & Khaodhair, S. (2006b). Options for environmental sustainability of the crude palm oil industry in Thailand through enhancement of industrial ecosystems. *Environment, Development and Sustainability*, 8(2), 271-287.
- Chen, H., Zhao, C., & Ren, Q. (2012). Feasibility of CO₂/SO₂ uptake enhancement of calcined limestone modified with rice husk ash during pressurized carbonation. *Journal of environmental management*, 93(1), 235–44.
- Cheng, Y., Lu, M., Li, J., Su, X., Pan, S., Jiao, C., & Feng, M. (2012). Synthesis of MCM-22 zeolite using rice husk as a silica source under varying-temperature conditions. *Journal of colloid and interface science*, 369(1), 388-394.
- Chiew, Y. L., & Cheong, K. Y. (2011). A review on the synthesis of SiC from plant-based biomasses. *Materials Science and Engineering: B*, 176(13), 951-964.
- Chindaprasirt, P, & Rukzon, S. (2008). Strength, porosity and corrosion resistance of ternary blend Portland cement, rice husk ash and fly ash mortar, 22, 1601–1606.
- Chindaprasirt, P., Homwuttiwong, S., & Jaturapitakkul, C. (2007b). Strength and water permeability of concrete containing palm oil fuel ash and rice husk–bark ash. Construction and Building Materials, 21(7), 1492-1499.
- Chindaprasirt, P., Jaturapitakkul, C., & Rattanasak, U. (2009). Influence of fineness of rice husk ash and additives on the properties of lightweight aggregate. *Fuel*, 88(1), 158–162.

- Chindaprasirt, P., Kanchanda, P., Sathonsaowaphak, a., & Cao, H. T. (2007a). Sulfate resistance of blended cements containing fly ash and rice husk ash. *Construction and Building Materials*, 21(6), 1356–1361.
- Chmelík, F., Trník, A., Štubňa, I., & Pešička, J. (2011). Creation of microcracks in porcelain during firing. *Journal of the European Ceramic Society*, 31(13), 2205-2209.
- Chouhan, R. K., Kujur, B., Amritphale, S. S., & CHANDRA, N. (2000). Effect of temperature of ashing of rice husk on the compressive strength of lime-rice husk silica mortar. *Silicates industriels*, (5-6), 67-71.
- Constâncio, C., Franco, L., Russo, A., Anjinho, C., Pires, J., Vaz, M. F., & Carvalho, A. P. (2010). Studies on polymeric conservation treatments of ceramic tiles with Paraloid B-72 and two alkoxysilanes. *Journal of Applied Polymer Science*, *116*(5), 2833-2839.
- Cordeiro, G. C., Toledo Filho, R. D., & Fairbairn, E. D. M. R. (2009). Use of ultrafine rice husk ash with high-carbon content as pozzolan in high performance concrete. *Materials and structures*, 42(7), 983-992.
- Cordeiro, G. C., Toledo Filho, R. D., Tavares, L. M., Fairbairn, E. D. M. R., & Hempel, S. (2011). Influence of particle size and specific surface area on the pozzolanic activity of residual rice husk ash. *Cement and Concrete Composites*, 33(5), 529–534.
- da Silva, L. H., Feitosa, S. A., Valera, M. C., de Araujo, M. A., & Tango, R. N. (2012). Effect of the addition of silanated silica on the mechanical properties of microwave heat-cured acrylic resin. *Gerodontology*, 29(2), e1019-e1023.
- Daguano, J. K., Suzuki, P. A., Strecker, K., Fernandes, M. H., & Santos, C. (2012). Evaluation of the micro-hardness and fracture toughness of amorphous and partially crystallized 3CaO· P₂O₅–SiO₂–MgO bioglasses. Materials Science and Engineering: A, 533, 26-32.
- Dana, K., Das, S., & Das, S. K. (2004). Effect of substitution of fly ash for quartz in triaxial kaolin–quartz–feldspar system. *Journal of the European Ceramic Society*, 24(10), 3169-3175.
- Dana, K., Dey, J., & Das, S. K. (2005). Synergistic effect of fly ash and blast furnace slag on the mechanical strength of traditional porcelain tiles. *Ceramics international*, 31(1), 147-152.
- Das, K.S. & Dana, K. (2003) Differences in densification behaviour of K- and Na-Feldspar-containing porcelain bodies. *Thermochimica Acta* 406, pp 199-206.
- De Noni, A., Hotza, D., Soler, V. C., & Vilches, E. S. (2009). Effect of quartz particle size on the mechanical behaviour of porcelain tile subjected to different cooling rates. *Journal of the European Ceramic Society*, 29(6), 1039-1046.

- De Noni, A., Hotza, D., Soler, V. C., & Vilches, E. S. (2010). Influence of composition on mechanical behaviour of porcelain tile. Part I: Microstructural characterization and developed phases after firing. *Materials Science and Engineering: A*, 527(7), 1730-1735.
- de'Gennaro, R., Cappelletti, P., Cerri, G., de'Gennaro, M., Dondi, M., Guarini, G., ... & Naimo, D. (2003). Influence of zeolites on the sintering and technological properties of porcelain stoneware tiles. *Journal of the European Ceramic Society*, 23(13), 2237-2245.
- Derevyagina, A. A., Derevyagin, G. F., & Gaidash, B. I. (1980). Using kyanite—Sillimanite concentrate for making high-tension porcelain. Glass and Ceramics, 37(5), 252-255.
- Deshmukh, P., Bhatt, J., Peshwe, D., & Pathak, S. (2012). Determination of Silica Activity Index and XRD, SEM and EDS Studies of Amorphous SiO2 Extracted from Rice Husk Ash. *Transactions of the Indian Institute of Metals*, 65(1), 63-70.
- Devaraj, A. R., Cheeseman, C. R., Boccaccini, A. R., & Deegan, D. (2010). Glass-Ceramic Tiles Prepared by Pressing and Sintering DC Plasma-Vitrified Air Pollution Control Residues. *International Journal of Applied Ceramic Technology*, 7(6), 925-934.
- Duku, M. H., Gu, S., & Hagan, E. B. (2011). Biochar production potential in Ghana—A review. *Renewable and Sustainable Energy Reviews*, 15(8), 3539-3551.
- Ece, O. I. and Nakagawa, Z. (2002) Bending Strength of Porcelains *Ceramics International* 28, pp 131-140.
- Eldagal, A., & Elmukhtar, O. (2008). Study on the behaviour of high strength palm oil fuel ash (POFA) concrete (Doctoral dissertation, Universiti Teknologi Malaysia, Faculty of Civil Engineering).
- El-Dakroury, a., & Gasser, M. S. (2008). Rice husk ash (RHA) as cement admixture for immobilization of liquid radioactive waste at different temperatures. *Journal of Nuclear Materials*, 381(3), 271–277.
- El-Maghraby, H. F., El-Omla, M. M., Bondioli, F., & Naga, S. M. (2011). Granite as flux in stoneware tile manufacturing. *Journal of the European Ceramic Society*, 31(12), 2057-2063.
- El-mahllawy, M. S., & Osman, T. A. (2010). Influence of Oil Well Drilling Waste on the Engineering Characteristics of Clay Bricks, *6*(7), 48–54.
- Engelthaler, A. Z. and Engena, M. Y (January, 1972) Ceramic raw materials in Uganda, preliminary report *African Ceramics Company* Ltd, Kampala, Uganda.
- Esposito, L., Salem, A., Tucci, A., Gualtieri, A., & Jazayeri (2005). The use of nepheline-syenite in a body mix for porcelain stoneware tiles *Ceramics International* 31, pp 233-240.

- Fadzil, A. M., Azmi, M. M., Hisyam, A. B., & Azizi, M. K. (2008). Engineering Properties of Ternary Blended Cement Containing Rice Husk Ash and Fly Ash as Partial Cement Replacement Materials. *A* (10), 125-134.
- Farzadnia, N., Abang Ali, A. A., & Demirboga, R. (2011). Incorporation of mineral admixtures in sustainable high performance concrete. *International Journal of Sustainable Construction Engineering and Technology*, 2(1).
- Ferreira-Leitao, V., Gottschalk, L. M. F., Ferrara, M. A., Nepomuceno, A. L., Molinari, H. B. C., & Bon, E. P. (2010). Biomass residues in Brazil: availability and potential uses. *Waste and Biomass Valorization*, *1*(1), 65-76.
- Foo, K. Y., & Hameed, B. H. (2009). Utilization of rice husk ash as novel adsorbent: A judicious recycling of the colloidal agricultural waste. *Advances in Colloid and Interface Science*, 152(1-2), 39–47.
- Frías, M., Vigil, R., García, R., Rodríguez, O., Goñi, S., & Vegas, I. (2012). Evolution of mineralogical phases produced during the pozzolanic reaction of different metakaolinite by-products: Influence of the activation process. *Applied Clay Science*, 56, 48-52.
- Furlani, E., Brückner, S., Minichelli, D., & Maschio, S. (2008). Synthesis and characterization of ceramics from coal fly ash and incinerated paper mill sludge. *Ceramics International*, 34(8), 2137-2142.
- Gadde, B., Bonnet, S., Menke, C., & Garivait, S. (2009). Air pollutant emissions from rice straw open field burning in India, Thailand and the Philippines. *Environmental Pollution*, 157(5), 1554-1558.
- Galos, K. (2011). Composition and ceramic properties of ball clays for porcelain stoneware tiles manufacture in Poland. *Applied clay science*, 51(1), 74-85.
- García Ten, J., Orts, M. J., Saburit, A., & Silva, G. (2010). Thermal conductivity of traditional ceramics. Part I: Influence of bulk density and firing temperature. *Ceramics International*, 36(6), 1951-1959.
- Garcia-Ten, J., Saburit, A., Bernardo, E., & Colombo, P. (2012). Development of lightweight porcelain stoneware tiles using foaming agents. *Journal of the European Ceramic Society*, 32(4), 745-752.
- Gastaldini, a. L. G., Isaia, G. C., Gomes, N. S., & Sperb, J. E. K. (2007). Chloride penetration and carbonation in concrete with rice husk ash and chemical activators. *Cement and Concrete Composites*, 29(3), 176–180.
- Gastaldini, a. L. G., Isaia, G. C., Hoppe, T. F., Missau, F., & Saciloto, A. P. (2009). Influence of the use of rice husk ash on the electrical resistivity of concrete: A technical and economic feasibility study. *Construction and Building Materials*, 23(11), 3411–3419.

- Ghasemi, Z., Younesi, H., & Kazemian, H. (2011). Synthesis of nanozeolite sodalite from rice husk ash without organic additives. *The Canadian Journal of Chemical Engineering*, 89(3), 601-608.
- Ghosh, T. B., Nandi, K. C., Acharya, H. N., & Mukherjee, D. (1991). XPS studies of magnesium silicide obtained from rice husk. *Materials Letters*, 11(1), 6-9.
- Givi, A. N., Rashid, S. A., Aziz, F. N. A., & Salleh, M. A. M. (2010). Assessment of the effects of rice husk ash particle size on strength, water permeability and workability of binary blended concrete. Construction and Building Materials, 24(11), 2145-2150.
- Gopal, J. (2001) The Development of Malaysia's Palm Oil Refining Industry: Obstacles, Policy and Performce. Ph.D. Thesis of the University of London and Diploma of Imperial College, England.
- Grimshaw, R. W., & Searle, A. B. (1971). *The chemistry and physics of clays and allied ceramic materials*. Wiley-Interscience.
- Guo, X., Wang, S., Wang, Q., Guo, Z., & Luo, Z. (2011). Properties of bio-oil from fast pyrolysis of rice husk. Chinese *Journal of Chemical Engineering*, 19(1), 116-121.
- Habeeb, G. A., & Fayyadh, M. M. (2009). Rice Husk Ash Concrete: the Effect of RHA Average Particle Size on Mechanical Properties and Drying Shrinkage, 3(3), 1616–1622.
- Hand, R.J., Stevens, S.J. & Sharp, J.H. (1998). "Characterisation of fired silicas", *Thermochimica Acta*, 318, 115-123.
- Handreck, K. A., & Jones, L. H. P. (1968). Studies of silica in the oat plant. *Plant and Soil*, 29(3), 449-459.
- Hanna, S. B., Farag, L. M., & Mansour, N. A. (1984). Pyrolysis and combustion of treated and untreated rice hulls. *Thermochimica Acta*, 81, 77-86.
- Harish, K. V., Rangaraju, P. R., & Vempati, R. K. (2010). Fundamental investigations into performance of carbon-neutral rice husk ash as supplementary cementitious material. Transportation Research Record: Journal of the Transportation Research Board, 2164(1), 26-35.
- Haslinawati, M. M., Matori, K. A., Wahab, Z. A., Sidek, H. A. A., & Zainal, A. T. (2009). Effect of temperature on ceramic from rice husk ash. *International Journal of Basic & Applied Sciences*, 9(9), 111-117.
- Hassan, J. U., Noh, M. Z., & Ahmad, Z. A. (2014). Effects of Palm Oil Fuel Ash Composition on the Properties and Morphology of Porcelain-palm Oil Fuel Ash Composite. Jurnal Teknologi, 70(5).
- He, D., Wang, P., Liu, P., & Liu, X. (2012). Interface Bond Mechanism of EVA-Modified Mortar and Porcelain Tile. *Journal of Materials in Civil Engineering*, 25(6), 726-730.

- Heim, W. F., & McDowell, H. L. (1982). U.S. Patent No. 4,342,632. Washington, DC: U.S. Patent and Trademark Office.
- Hojamberdiev, M., Eminov, A., & Xu, Y. (2011). Utilization of muscovite granite waste in the manufacture of ceramic tiles. *Ceramics International*, *37*(3), 871-876.
- Höland, W., Rheinberger, V., Apel, E., van't Hoen, C., Höland, M., Dommann, A., ... & Graf-Hausner, U. (2006). Clinical applications of glass-ceramics in dentistry. *Journal of Materials Science: Materials in Medicine*, 17(11), 1037-1042.
- Houston, D. F. (1972). Rice Chemistry and Technology, American Association of Cereal Chemist. *Inc, Minnesota*.
- Hunt, L. P., Dismukes, J. P., Amick, J. A., Schei, A., & Larsen, K. (1984). Rice hulls as a raw material for producing silicon. Journal of the Electrochemical Society, 131(7), 1683-1686.
- Hussain, R. R., & Ishida, T. (2010). Development of numerical model for FEM computation of oxygen transport through porous media coupled with micro-cell corrosion model of steel in concrete structures. *Computers & structures*, 88(9), 639-647.
- Hussain, R. R., & Ishida, T. (2011). Investigation of volumetric effect of coarse aggregate on corroding steel reinforcement at the interfacial transition zone of concrete. *KSCE Journal of Civil Engineering*, 15(1), 153-160.
- Iqbal, Y and Lee, W.E (2000) "Microstructural evolution in Triaxial porcelain" J. Am. Ceram. Soc. 83 [12] pp 3121-27.
- Iqbal, Y., & Lee, W. E. (1999). Fired porcelain microstructures revisited. *Journal of the American Ceramic Society*, 82(12), 3584-3590.
- Iqbal, Y., & Lee, W. E. (2000). Microstructural evolution in triaxial porcelain. *Journal of the American Ceramic Society*, 83(12), 3121-3127.
- Isaia, G. ., Gastaldini, a. L. ., & Moraes, R. (2003). Physical and pozzolanic action of mineral additions on the mechanical strength of high-performance concrete. *Cement and Concrete Composites*, 25(1), 69–76. Ismail, S., & Waliuddints, A. M. (1996). Effect of rice husk ash on high strength concrete, 10(I), 521–526.
- Ishak, Z. M., Bakar, A. A., Ishiaku, U. S., Hashim, A. S., &Azahari, B. (1997). An investigation of the potential of rice husk ash as a filler for epoxidized natural rubber—II.Fatigue behaviour.European polymer journal, 33(1), 73-79.
- Islam, R. A., Chan, Y.C, and Islam M. F. (2004) Structure-property relationship in high-tension ceramic insulator fired at high temperature *Materials Science and Engineering* B106, pp 132-140.
- Ismail, M. E., Budiea, A., Hussin, M. W., & Muthusamy, K. (2010). Effect of POFA fineness towards durability of POFA cement based high strength concrete. *The Indian Concrete Journal*.

- James, J., & Subba Rao, M. (1986). Reactivity of rice husk ash. *Cement and concrete research*, 16(3), 296-302.
- James, K. A. C., Bray, J. J., Morgan, I. G., & Austin, L. (1970). The effect of colchicine on the transport of axonal protein in the chicken. *Biochem. J*, 117, 767-771.
- Jamo, H. U., Noh, M. Z., & Ahmad, Z. A. (2013). Structural analysis and surface morphology of a treated palm oil fuel ash.
- Jamo, H. U., Noh, M. Z., & Ahmad, Z. A. (2014). Influence of Temperature on the Substitution of Quartz by Rice Husk Ash (RHA) in Porcelain Composition. Applied Mechanics and Materials, 465, 1297-1303.
- Jaturapitakkul, C., Kiattikomol, K., Tangchirapat, W., & Saeting, T. (2007). Evaluation of the sulfate resistance of concrete containing palm oil fuel ash. *Construction and Building Materials*, 21(7), 1399-1405.
- Johari, M., Zeyad, A. M., Muhamad Bunnori, N., & Ariffin, K. S. (2012). Engineering and transport properties of high-strength green concrete containing high volume of ultrafine palm oil fuel ash. *Construction and Building Materials*, 30, 281-288.
- Kalapathy, U., Proctor, A., & Shultz, J. (2000). A simple method for production of pure silica from rice hull ash. *Bioresource Technology*, 73(3), 257-262.
- Kamseu, E., Leonelli, C., Boccaccini, D. N., Veronesi, P., Miselli, P., Pellacani, G., & Melo, U. C. (2007). Characterisation of porcelain compositions using two china clays from Cameroon. *Ceramics international*, *33*(5), 851-857.
- Kapur, P. C. (1985). Production of reactive bio-silica from the combustion of rice husk in a tube-in-basket (TiB) burner. *Powder technology*, 44(1), 63-67.
- Karamanov, A., Karamanova, E., Ferrari, A. M., Ferrante, F., & Pelino, M. (2006). The effect of fired scrap addition on the sintering behaviour of hard porcelain. Ceramics international, 32(7), 727-732.
- Karim, M. R., Zain, M. E. M., Jamil, M., Lai, F. C., & Islam, M. N. (2011). Strength development of mortar and concrete containing fly ash: A review. *International Journal of Physical*. *Sciences*, 6(17), 4137-4153.
- Kaya, G., Karasu, B., & Cakir, A. (2011). Characterisation of diopside-based glass-ceramic porcelain tile glazes containing borax solid wastes. *Journal of Ceramic Processing Research*, 12(2), 135-139.
- Ke, S., Cheng, X., Wang, Y., Wang, Q., & Wang, H. (2012). Dolomite, wollastonite and calcite as different CaO sources in anorthite-based porcelain. *Ceramics International*.
- Kingery, W. D. (1986). The development of European porcelain. In *High-Technology Ceramics: Past, Present, and Future-The Nature of Innovation and Change in Ceramic Technology* (Vol. 3, pp. 153-180). The American Ceramic Society, Inc..
- Kitouni, S., & Harabi, A. (2011). Sintering and mechanical properties of porcelains prepared from algerian raw materials. *Cerâmica*, *57*(344), 453-460.

- Klankaw, P., Chawengkijwanich, C., Grisdanurak, N., & Chiarakorn, S. (2012). The hybrid photocatalyst of TiO< sub> 2</sub>—SiO< sub> 2</sub> thin film prepared from rice husk silica. Superlattices and Microstructures, 51(3), 343-352.
- Kobayashi, K., Komine, F., Blatz, M. B., Saito, A., Koizumi, H., & Matsumura, H. (2009). Influence of priming agents on the short-term bond strength of an indirect composite veneering material to zirconium dioxide ceramic. *Quintessence international (Berlin, Germany: 1985)*, 40(7), 545.
- Kobayashi, Y. (1987). Strength and weibull distribution of alumina strengthened whiteware bodies. Ceramic Society Japan, 95(9), 887.
- Koloor, S. S. R., Kashani, J., & Kadir, M. A. (2011, January). Simulation of Brittle Damage for Fracture Process of Endodontically Treated Tooth. In *5th Kuala Lumpur International Conference on Biomedical Engineering 2011* (pp. 210-214). Springer Berlin Heidelberg.
- Kordani, N., Asadi, A., & Jabbari, A. (2012). Optimization of fracture behavior of alumina/silicon carbide nano ceramic. *Neural Computing and Applications*, 1-7.
- Kroehong, W., Sinsiri, T., Jaturapitakkul, C., & Chindaprasirt, P. (2011). Effect of palm oil fuel ash fineness on the microstructure of blended cement paste. *Construction and Building Materials*, 25(11), 4095-4104.
- Kurama, S., & Kurama, H. (2008). The reaction kinetics of rice husk based cordierite ceramics. *Ceramics International*, *34*(2), 269–272.
- LaSalvia, J. C., Campbell, J., Swab, J. J., & McCauley, J. W. (2010). Beyond hardness: ceramics and ceramic-based composites for protection. *JOM*, 62(1), 16-23.
- Lee, D. J., Jang, J. J., Park, H. S., Kim, Y. C., Lim, K. H., Park, S. B., & Hong, S. H. (2012). Fabrication of biomorphic SiC composites using wood preforms with different structures. *Ceramics International*, 38(4), 3089-3095.
- Lee, W. E. (1994). *Ceramic microstructures*: property control by processing. Springer.
- Lee, W. E., & Iqbal, Y. (2001). Influence of mixing on mullite formation in porcelain. *Journal of the European Ceramic Society*, 21(14), 2583-2586.
- Lee, W. E., & Iqbal, Y. (2001). Influence of mixing on mullite formation in porcelain. Journal of the European Ceramic Society, 21(14), 2583-2586.
- Lee, W. E., Souza, G. P., McConville, C. J., Tarvornpanich, T., & Iqbal, Y. (2008). Mullite formation in clays and clay-derived vitreous ceramics. *Journal of the European Ceramic Society*, 28(2), 465-471.
- Leonelli, C., Bondioli, F., Veronesi, P., Romagnoli, M., Manfredini, T., Pellacani, G. C., & Cannillo, V. (2001). Enhancing the mechanical properties of porcelain stoneware tiles: a microstructural approach. *Journal of the European Ceramic Society*, 21(6), 785-793.

- Leonelli, C., Bondioli, F., Veronesi, P., Romagnoli, M., Manfredini, T., Pellacani, G. C., & Cannillo, V. (2001). Enhancing the mechanical properties of porcelain stoneware tiles: a microstructural approach. *Journal of the European Ceramic Society*, 21(6), 785-793.
- Li, Y., Ding, X., Guo, Y., Rong, C., Wang, L., Qu, Y., & Wang, Z. (2011). A new method of comprehensive utilization of rice husk. Journal of hazardous materials, 186(2), 2151-2156.
- Li, Y., Zhao, C., Ren, Q., Duan, L., Chen, H., & Chen, X. (2009). Effect of rice husk ash addition on CO₂ capture behavior of calcium-based sorbent during calcium looping cycle. *Fuel Processing Technology*, 90(6), 825–834.
- Li, Z., Tang, S., Deng, X., Wang, R., & Song, Z. (2010). Contrasting effects of elevated CO₂ on Cu and Cd uptake by different rice varieties grown on contaminated soils with two levels of metals: Implication for phytoextraction and food safety. *Journal of hazardous materials*, 177(1), 352-361.
- Liou, T. H. (2004). Preparation and characterization of nano-structured silica from rice husk. *Materials Science and Engineering*: A, 364(1), 313-323.
- Liu, J., Zehnder, A. J., & Yang, H. (2009). Global consumptive water use for crop production: The importance of green water and virtual water. *Water Resources Research*, 45(5).
- Lopez, S. Y. R., Yobanny, S., Serrato Rodriguez, J., & Sugita Sueyoshi, S. (2011). Determination of the activation energy for densification of porcelain stoneware. *Journal of Ceramic Processing Research*, 12(3), 228-232.
- Lundin, S. T. (1964). Microstructure of porcelain. *Microstructure of Ceramic Materials*. *NBS Misc*. Publ, 257, 93-106.
- Luz, A. P., & Ribeiro, S. (2007). Use of glass waste as a raw material in porcelain stoneware tile mixtures. *Ceramics international*, 33(5), 761-765.
- Ma, X., Zhou, B., Gao, W., Qu, Y., Wang, L., Wang, Z., & Zhu, Y. (2012). A recyclable method for production of pure silica from rice hull ash. *Powder*
- Maity, S., & Sarka, B. K. (1996). Development of High-Strength Whiteware Bodies, 2219(96).
- Maity, S., & Sarkar, B. K. (1996). Development of high-strength whiteware bodies. *Journal of the European Ceramic Society*, 16(10), 1083-1088.
- Maity, S., & Sarkar, B. K. (1996). Development of high-strength whiteware bodies. *Journal of the European Ceramic Society*, 16(10), 1083-1088.
- Maity, S., & Sarkar, B. K. (1996). Development of high-strength whiteware bodies. *Journal of the European Ceramic Society*, 16(10), 1083-1088.
- Malaysian Palm Oil Board (MPOB). Economic and statistic; 2013 http://econ.mpob.gov.my/economy/Overview_2009.pdf [accessed 12.05. 2012].

- Mane, V. S., Deo Mall, I., & Chandra Srivastava, V. (2007). Kinetic and equilibrium isotherm studies for the adsorptive removal of Brilliant Green dye from aqueous solution by rice husk ash. Journal of Environmental Management, 84(4), 390-400.
- Manfredini, T., Pellacani, G. C., Romagnoli, M., & Pennisi, L. (1995). Porcelainized stoneware tile. *American Ceramic Society Bulletin*, 74(5), 76-79
- Mansaray, K. G., & Ghaly, A. E. (1997). Physical and thermochemical properties of rice husk. *Energy Sources*, 19(9), 989-1004.
- Martin, J. I. (1938). *The desilification of rice husk and a study of the products obtained*. Lousiana State University, USA: Master Thesis.
- Martínez, J. D., Pineda, T., López, J. P., & Betancur, M. (2011). Assessment of the rice husk lean-combustion in a bubbling fluidized bed for the production of amorphous silica-rich ash. *Energy*, *36*(6), 3846-3854.
- Martín-Márquez, J., De la Torre, A. G., Aranda, M. A., Rincón, J. M., & Romero, M. (2009). Evolution with temperature of crystalline and amorphous phases in porcelain stoneware. *Journal of the American Ceramic Society*, 92(1), 229-234.
- Martín-Márquez, J., Rincón, J. M., & Romero, M. (2008). Effect of firing temperature on sintering of porcelain stoneware tiles. *Ceramics International*, *34*(8), 1867-1873.
- Martín-Márquez, J., Rincón, J. M., & Romero, M. (2010). Mullite development on firing in porcelain stoneware bodies. *Journal of the European Ceramic Society*, 30(7), 1599-1607.
- Martín-Márquez, J., Rincón, J. M., & Romero, M. (2010a). Effect of microstructure on mechanical properties of porcelain stoneware. *Journal of the European Ceramic Society*, 30(15), 3063-3069.
- Martín-Márquez, J., Rincón, J. M., & Romero, M. (2010b). Mullite development on firing in porcelain stoneware bodies. *Journal of the European Ceramic Society*, 30(7), 1599-1607.
- Matteucci, F., Dondi, M., & Guarini, G. (2002). Effect of soda-lime glass on sintering and technological properties of porcelain stoneware tiles. *Ceramics International*, 28(8), 873-880.
- Mehta, P. K., & Pitt, N. (1976). Energy and industrial materials from crop residues. *Resource Recovery and Conservation*, 2(1), 23-38.
- Mondragon, R., Jarque, J. C., Julia, J. E., Hernandez, L., & Barba, A. (2012). Effect of slurry properties and operational conditions on the structure and properties of porcelain tile granules dried in an acoustic levitator. *Journal of the European Ceramic Society*, 32(1), 59-70.
- Montoya, N., Serrano, F. J., Reventós, M. M., Amigo, J. M., & Alarcón, J. (2010). Effect of TiO< sub> 2</sub> on the mullite formation and mechanical properties of alumina porcelain. *Journal of the European Ceramic Society*, 30(4), 839-846.

- Moradkhani, A. R., Baharvandi, H. R., Vafaeesefat, A., & Tajdari, M. (2012). Microstructure and Mechanical Properties of Al2O3-SiC Nanocomposites with 0.05% MgO and Different SiC Volume Fraction. *International Journal of Advanced Design and Manufacturing Technology*, 5(3), 99-105.
- Mukhopadhyay, T. K., Das, M., Ghosh, S., Chakrabarti, S., & Ghatak, S. (2003). Microstructure and thermo mechanical properties of a talc doped stoneware composition containing illitic clay. *Ceramics international*, 29(5), 587-597.
- Mukhopadhyay, T. K., Ghosh, S., Ghatak, S., & Maiti, H. S. (2006). Effect of pyrophyllite on vitrification and on physical properties of triaxial porcelain. *Ceramics international*, 32(8), 871-876.
- Mukhopadhyay, T. K., Ghosh, S., Ghatak, S., & Maiti, H. S. (2006). Effect of pyrophyllite on vitrification and on physical properties of triaxial porcelain. *Ceramics international*, 32(8), 871-876.
- Naga, S. M., Bondioli, F., Wahsh, M. M. S., & El-Omla, M. (2012). Utilization of granodiorite in the production of porcelain stoneware tiles. *Ceramics International*, 38(8), 6267-6272.
- Naivinit, W., Le Page, C., Trébuil, G., & Gajaseni, N. (2010). Participatory agent-based modeling and simulation of rice production and labor migrations in Northeast Thailand. *Environmental Modelling & Software*, 25(11), 1345-1358.
- Naskar, M. K., & Chatterjee, M. (2004). A novel process for the synthesis of cordierite (Mg2Al4Si5O18) powders from rice husk ash and other sources of silica and their comparative study. Journal of the European Ceramic Society, 24(13), 3499-3508.
- Natarajan, E., Öhman, M., Gabra, M., Nordin, A., Liliedahl, T., & Rao, A. N. (1998). Experimental determination of bed agglomeration tendencies of some common agricultural residues in fluidized bed combustion and gasification. *Biomass and Bioenergy*, 15(2), 163-169.
- Nayak, J. P., Kumar, S., & Bera, J. (2010). Sol–gel synthesis of bioglass-ceramics using rice husk ash as a source for silica and its characterization. Journal of Non-Crystalline Solids, 356(28), 1447-1451.
- Neklyudova, T. L., & Maslennikova, G. N. (2011). Structural particulars of porcelain articles manufactured by slip casting. *Glass and Ceramics*, 68(1), 52-55.
- Ngo, S. P. (2006). Production of amorphous silica from rice husk in fluidised bed system (Doctoral dissertation, Universiti Teknologi Malaysia, Faculty of Chemical Engineering and Natural Resources Engineering).
- Norton, F.H. (1970) Fine Ceramics, Technology and Applications. McGraw-Hill Book Co. New York.

- Okoronkwo, E. A., Imoisili, P. E., & Olusunle, S. O. O. (2013). Extraction and characterization of Amorphous Silica from Corn Cob Ash by Sol-Gel Method. *Chemistry and Materials Research*, *3*(4), 68-72.
- Olgun, A., Erdogan, Y., Ayhan, Y., & Zeybek, B. (2005). Development of ceramic tiles from coal fly ash and tincal ore waste. *Ceramics International*, 31(1), 153-158.
- Olupot, P. W. (2006). Assessment of ceramic raw materials in Uganda for electrical porcelain (Doctoral dissertation, KTH).
- Pelaez-Vargas, A., Dussan, J. A., Restrepo-Tamayo, L. F., Paucar, C., Ferreira, J. A., & Monteiro, F. J. (2011). The effect of slurry preparation methods on biaxial flexural strength of dental porcelain. *The Journal of prosthetic dentistry*, 105(5), 308-314.
- Pérez, J. M., Rincón, J. M., & Romero, M. (2012). Effect of moulding pressure on microstructure and technological properties of porcelain stoneware. *Ceramics International*, 38(1), 317-325.
- Pinheiro, B. C. A., & Holanda, J. N. F. (2013). Reuse of solid petroleum waste in the manufacture of porcelain stoneware tile. *Journal of environmental management*, 118, 205-210.
- Porion, P., Busignies, V., Mazel, V., Leclerc, B., Evesque, P., & Tchoreloff, P. (2010).

 Anisotropic Porous Structure of Pharmaceutical Compacts Evaluated by PGSTENMR in Relation to Mechanical Property Anisotropy. *Pharmaceutical research*,
 27(10), 2221-2233.
- Prasad, C. S., Maiti, K. N., & Venugopal, R. (2001). Effect of rice husk ash in whiteware compositions. *Ceramics International*, 27(6), 629-635.
- Prasad, C. S., Maiti, K. N., & Venugopal, R. (2003). Effect of substitution of quartz by rice husk ash and silica fume on the properties of whiteware compositions. *Ceramics international*, 29(8), 907-914.
- Prasad, C.S, Maiti, K.N. and Venugopal R. (2002) Effect of silica fume addition on the properties of whiteware compositions *Ceramics International* 28 pp 9-15
- Prasad, R., & Pandey, M. (2012). Rice husk ash as a renewable source for the production of value added silica gel and its application: an overview. *Bulletin of Chemical Reaction Engineering & Catalysis*, 7(1), 1-25.
- Prasad, S. D., & Krishna, R. A. (2011). Production and mechanical properties of A356. 2/RHA composites. *International Journal of Advanced Science and Technology*, 33, 51-58.
- Prasara-A, J. (2009). Comparative Life Cycle Assessment of Rice Husk Utilization in Thailand (Doctoral dissertation, RMIT University).
- Prasetyoko, D., Ramli, Z., Endud, S., & Hamdan, H. (2006a). Conversion of rice husk ash to zeolite beta, 26, 1173–1179.

- Rahaman, N. 2003. Ceramic Processing and Sintering, M. Dekker.
- Raimondo, M., Dondi, M., Zanelli, C., Guarini, G., Gozzi, A., Marani, F., & Fossa, L. (2010). Processing and properties of large-sized ceramic slabs. *Boletín de la Sociedad Española de Cerámica y Vidrio*, 49(4), 289-295.
- Rambaldi, E., Carty, W. M., Tucci, A., & Esposito, L. (2007). Using waste glass as a partial flux substitution and pyroplastic deformation of a porcelain stoneware tile body. *Ceramics international*, 33(5), 727-733.
- Rambaldi, E., Tucci, A., Esposito, L., Naldi, D., & Timellini, G. (2010). Nano-oxides to improve the surface properties of ceramic tiles. Boletín de la Sociedad Española de Cerámica y Vidrio, 49(4), 253-258.
- Ramesh, S., Tolouei, R., Hamdi, M., Purbolaksono, J., Y Tan, C., Amiriyan, M., & D Teng, W. (2011). Sintering Behavior of Nanocrystalline Hydroxyapatite Produced by Wet Chemical Method. *Current Nanoscience*, 7(6), 845-849.
- Randal, M. G. 1991. Engineered Materials Handbook: Ceramics and Glasses, Ohio.
- Rangaraju, P., & Harish, K. V. Optimization of Use of Rice Husk Ash for use as SCM in Cementitious Mortars.
- Rincón, J. Ma.,(1992). Principes of nucleation and controlled crystallization of glasses, Polym. Plast. Technol. Eng., 31, 309-357.
- Rodríguez de Sensale, G. (2006). Strength development of concrete with rice-husk ash. *Cement and Concrete Composites*, 28(2), 158–160.
- Romero, M., Martín-Márquez, J., & Rincón, J. M. (2006). Kinetic of mullite formation from a porcelain stoneware body for tiles production. *Journal of the European Ceramic Society*, 26(9), 1647-1652.
- Rosa, V., Yoshimura, H. N., Pinto, M. M., Fredericci, C., & Cesar, P. F. (2009). Effect of ion exchange on strength and slow crack growth of a dental porcelain. *dental materials*, 25(6), 736-743.
- Rukzon, S., Chindaprasirt, P., & Mahachai, R. (2009). Effect of grinding on chemical and physical properties of rice husk ash. International Journal of Minerals, Metallurgy and Materials, 16(2), 242-247.
- Safiuddin, M., Abdus Salam, M., & Jumaat, M. Z. (2011). Utilization of palm oil fuel ash in concrete: a review. *Journal of Civil Engineering and Management*, 17(2), 234-247.
- Safiuddin, M., Jumaat, M. Z., Salam, M. A., Islam, M. S., & Hashim, R. (2010). Utilization of solid wastes in construction materials. International *Journal of Physical Sciences*, 5(13), 1952-1963.

- Saidur, R., Abdelaziz, E. A., Demirbas, A., Hossain, M. S., & Mekhilef, S. (2011). A review on biomass as a fuel for boilers. *Renewable and Sustainable Energy Reviews*, 15(5), 2262-2289.
- Salem, A., Jazayeri, S. H., Rastelli, E., & Timellini, G. (2010). Kinetic model for isothermal sintering of porcelain stoneware body in presence of nepheline syenite. Thermochimica Acta, 503, 1-7.
- Sanchez, E., Ibanez, M. J., García-Ten, J., Quereda, M. F., Hutchings, I. M., & Xu, Y. M. (2006). Porcelain tile microstructure: Implications for polished tile properties. *Journal of the European Ceramic Society*, 26(13), 2533-2540.
- Sanchez, E., Orts, M. J., Garcia-Ten, J., & Cantavella, V. (2001). Porcelain tile composition effect on phase formation and end products. *American Ceramic Society Bulletin*, 80(6), 43-49.
- Santos, T., Costa, L. C., Hennetier, L., Valente, M. A., Monteiro, J., & Sousa, J. (2012). Microwave processing of porcelain tableware using a multiple generator configuration. *Applied Thermal Engineering*.
- Sarangi, M., Bhattacharyya, S., & Behera, R. C. (2009). Effect of temperature on morphology and phase transformations of nano-crystalline silica obtained from rice husk. *Phase Transitions*, 82(5), 377-386.
- Saravanan, K., Yuvakkumar, R., Rajendran, V., & Paramasivam, P. (2012). Influence of sintering temperature and pH on the phase transformation, particle size and antireflective properties of RHA nano silica powders. *Phase Transitions*, 85(12), 1109-1124.
- Sata, V., Jaturapitakkul, C., & Kiattikomol, K. (2007). Influence of pozzolan from various by-product materials on mechanical properties of high-strength concrete. *Construction and Building Materials*, 21(7), 1589–1598.
- Sata, V., Jaturapitakkul, C., & Rattanashotinunt, C. (2010). Compressive strength and heat evolution of concretes containing palm oil fuel ash. *Journal of Materials in Civil Engineering*, 22(10), 1033-1038.
- Sata, V., Jaturapitakkul, C., & Rattanashotinunt, C. (2010). Compressive strength and heat evolution of concretes containing palm oil fuel ash. *Journal of Materials in Civil Engineering*, 22(10), 1033-1038.
- Sata, V., Tangpagasit, J., Jaturapitakkul, C., & Chindaprasirt, P. (2012). Effect of W/B ratios on pozzolanic reaction of biomass ashes in Portland cement matrix. *Cement and Concrete Composites*, *34*(1), 94-100.
- Schey, J.A. (2000). Introduction to Manufacturing Processes 3rd ed. U.S.A.,: McGraw Hill Higher Education pp476-489.
- Schroeder, E. J. (1978) Inexpensive high strength electrical porcelain Am. Ceram. Soc. Bull. 57, 526.

- Schuller, K.H. (1964) Reactions between mullite and glassy in porcelain, Trans. J. Brit. Ceram. Soc. 63 pp 103.
- Sharma, P., Kaur, R., Baskar, C., & Chung, W. J. (2010). Removal of methylene blue from aqueous waste using rice husk and rice husk ash. *Desalination*, 259(1), 249-257.
- Shibib, K. S., Qatta, H. I., & Hamza, M. S. (2012). Enhancement in thermal and mechanical properties of bricks. *Thermal Science*, (00), 43-43.
- Shu, Z., Garcia-Ten, J., Monfort, E., Amoros, J. L., Zhou, J., & Wang, Y. X. (2012). Cleaner duction of porcelain tile powders. Granule and green compact characterization. *Ceramics International*, 38(1), 517-526.
- Shui, A., Xi, X., Wang, Y., & Cheng, X. (2011). Effect of silicon carbide additive on microstructure and properties of porcelain ceramics. *Ceramics International*, *37*(5), 1557-1562.
- Singer, F. Singer, S.S. (1971). Industrial Ceramics, Chapman and Hall, London
- Siqueira, E. J., Yoshida, I. V. P., Pardini, L. C., & Schiavon, M. A. (2009). Preparation and characterization of ceramic composites derived from rice husk ash and polysiloxane. Ceramics International, 35(1), 213-220.
- Snel, M. D., Snijkers, F., Luyten, J., & Kodentsov, A. (2007). Aqueous tape casting of reaction bonded aluminium oxide (RBAO). *Journal of the European Ceramic Society*, 27(1), 27-33.
- Sokolar, R., & Smetanova, L. (2010). Dry pressed ceramic tiles based on fly ash—clay body: influence of fly ash granulometry and pentasodium triphosphate addition. Ceramics international, 36(1), 215-221.
- Sokolar, R., & Vodova, L. (2011). The effect of fluidized fly ash on the properties of dry pressed ceramic tiles based on fly ash–clay body. *Ceramics International*, *37*(7), 2879-2885.
- Sokolar, R., & Vodova, L. (2011). The effect of fluidized fly ash on the properties of dry pressed ceramic tiles based on fly ash–clay body. *Ceramics International*, 37(7), 2879-2885.
- Sonuparlak, M. Sarikaya, and I. A. Aksay, (1987). "Spinel Phase Formation during 980°C Exothermic Reaction in the Kaolinite-to-Mullite Reaction Se- ries," J. Am. Ceram. Soc., 70 [III 83742., 70 [III 83742]]
- Sooksaen, P. (2011). Effect of alumina content on the properties of ceramic foams crystallized with nano-sized mullite crystals. In *Materials Science Forum* (Vol. 694, pp. 456-460).
- Soraru, G. D., & Tassone, P. (2004). Mechanical durability of a polymer concrete: a Vickers indentation study of the strength degradation process. *Construction and Building Materials*, 18(8), 561-566.

- Standard A.S.T.M. (2011). C773-88(2011) Standard Test Method for Compressive (Crushing) Strength of Fired Whitewares Materials , ASTM, West Conshohocken, PA.
- Standard, A. S. T. M. (2006). C373-88(2006). Standard test method for water absorption, bulk density, apparent density and the apparent specific gravity of fired whiteware products. *American Society for Testing Materials*, West Conshohocken, PA.
- Standard, A. S. T. M. (2008). C1161-02 (2008) . Standard test method for flexural strength of advanced ceramics at ambient temperature. West Conshocken, PA.
- Standard, A.S.T.M.(2008).C1327. Standard test method for Vickers indentation hardness of advanced ceramics. Materials , ASTM, West Conshohocken, PA tion for Testing Materials; 1999.
- Stathis, G., Ekonomakou, A., Stournaras, C.J, and Ftikos, C. (2004) Effect of firing conditions, filler grain size and quartz content on bending strength and physical properties of sanitary ware porcelain *Journal of the European Ceramic Society* 24, pp 2357-2366.
- Stathis, G., Ekonomakou, A., Stournaras, C.J, and Ftikos, C. (2004) Effect of firing conditions, filler grain size and quartz content on bending strength and physical properties of sanitary ware porcelain *Journal of the European Ceramic Society* 24, pp 2357-2366.
- Sumran, M., & Kongkachuichay, P. (2013). Synthesis of Silica from Rice Husk by Onestep Combustion, Fluidized Bed Combustion, and Alkaline Extraction. *KKU Engineering Journal*, 30(2), 165-172.
- Suvaci, E., & Tamsu, N. (2010). The role of viscosity on microstructure development and stain resistance in porcelain stoneware tiles. *Journal of the European Ceramic Society*, 30(15), 3071-3077.
- Tai, Weon-Pil. Kimura, K. and Jinnai K. (2002) "A new approach to anorthite porcelain bodies using non-plastic raw materials" Journal of the *European Ceramic Society* 22, pp 463-470.
- Tangchirapat, W., & Jaturapitakkul, C. (2010). Strength, drying shrinkage, and water permeability of concrete incorporating ground palm oil fuel ash. *Cement and Concrete Composites*, 32(10), 767-774.
- Tangchirapat, W., Jaturapitakkul, C., & Chindaprasirt, P. (2009). Use of palm oil fuel ash as a supplementary cementitious material for producing high-strength concrete. *Construction and Building Materials*, 23(7), 2641-2646.
- Tarvornpanich, T., Souza, G. P., & Lee, W. E. (2008). Microstructural Evolution in Clay-Based Ceramics II: Ternary and Quaternary Mixtures of Clay, Flux, and Quartz Filler. Journal of the American Ceramic Society, 91(7), 2272-2280.

- Tasić, Ź. D. (1993). Improving the microstructural and physical properties of alumina electrical porcelain with Cr2O3, MnO2 and ZnO additives. *Journal of materials science*, 28(21), 5693-5701.
- Tay, J. H. (1990). Ash from oil-palm waste as a concrete material. *Journal of Materials in Civil Engineering*, 2(2), 94-105.
- Tay, J. H., & Show, K. Y. (1995). Use of ash derived from oil-palm waste incineration as a cement replacement material. *Resources, conservation and recycling*, 13(1), 27-36.
- Tenorio Cavalcante, P. M., Dondi, M., Ercolani, G., Guarini, G., Melandri, C., Raimondo, M., & Rocha e Almendra, E. (2004). The influence of microstructure on the performance of white porcelain stoneware. *Ceramics international*, 30(6), 953-963.
- Thompson, M., Fahrenholtz, W. G., & Hilmas, G. (2011). Effect of starting particle size and oxygen content on densification of ZrB2. *Journal of the American Ceramic Society*, 94(2), 429-435.
- Thurnauer, H. (1954) Dielectric Materials and Applications (A.R.V. Hippel, Ed.)
- Tian, X., Günster, J., Melcher, J., Li, D., & Heinrich, J. G. (2009). Process parameters analysis of direct laser sintering and post treatment of porcelain components using Taguchi's method. *Journal of the European Ceramic Society*, 29(10), 1903-1915.
- Tolouei, R., Singh, R., Sopyan, I., Tan, C. Y., Amiriyan, M., & Teng, W. D. (2011). Dependence of the Fracture Toughness on the Sintering Time of Dense Hydroxyapatite Bioceramics. In *Materials Science Forum* (Vol. 694, pp. 391-395).
- Tomizaki, M. F., & Shinkai, S. (1995). Influence of the grain size of alumina in the glaze of fast-fired porcelain tiles. Nippon seramikkusu kyokai gakujutsu ronbunshi, 103(4), 325-329.
- Tonnayopas D, Nilrat F, Putto K, Tantiwitayawanich J (2006). Effect of Oil Palm Fiber Fuel Ash on Compressive Strength of Hardened Concrete. In: Proceedings of the Fourth Thailand Materials Science and Technology Conference, Pathumthani, Thailand. pp.1-3.
- Tonnayopas, D., Kooptarnond, K., & Masae, M. (2009). Novel Ecological Tiles Made with Granite Fine Quarry Waste and Oil Palm Fiber Ash. *Thammasat Int. J. Sc. Tech*, 14(1).
- Torres, Y., Pavón, J. J., & Rodríguez, J. A. (2012). Processing and characterization of porous titanium for implants by using NaCl as space holder. *Journal of Materials Processing Technology*, 212(5), 1061-1069.
- Tucci, A., Esposito, L., Malmusi, L., & Rambaldi, E. (2007). New body mixes for porcelain stoneware tiles with improved mechanical characteristics. *Journal of the European Ceramic Society*, 27(2), 1875-1881.

- Tucci, A., Esposito, L., Rastelli, E., Palmonari, C., & Rambaldi, E. (2004). Use of sodalime scrap-glass as a fluxing agent in a porcelain stoneware tile mix. *Journal of the European Ceramic Society*, 24(1), 83-92.
- Tucci, A., Esposito, L., Rastelli, E., Palmonari, C., & Rambaldi, E. (2004). Use of sodalime scrap-glass as a fluxing agent in a porcelain stoneware tile mix. *Journal of the European Ceramic Society*, 24(1), 83-92.
- Tuck, C. O., Pérez, E., Horváth, I. T., Sheldon, R. A., & Poliakoff, M. (2012). Valorization of biomass: deriving more value from waste. *Science*, *337*(6095), 695-699.
- Usman, J., Arifin, Z., & Zaky, M. (2012). The effects of Rice Husk Ash on Porcelain composition, 2012(Rafss).
- Van Tuan, N., Ye, G., Van Breugel, K., Fraaij, A. L., & Bui, D. D. (2011). The study of using rice husk ash to produce ultra high performance concrete. Construction and Building Materials, 25(4), 2030-2035.
- Vandeperre, L. J., Wang, J., & Clegg†, W. J. (2004). Effects of porosity on the measured fracture energy of brittle materials. *Philosophical Magazine*, 84(34), 3689-3704.
- Vargas-Gonzalez, L., Speyer, R. F., & Campbell, J. (2010). Flexural strength, fracture toughness, and hardness of silicon carbide and boron carbide armor ceramics.

 International Journal of Applied Ceramic Technology, 7(5), 643-651.
- Vásquez, V., Özcan, M., Nishioka, R., Souza, R., Mesquita, A., & Pavanelli, C. (2008). Mechanical and Thermal Cycling Effects on the Flexural Strength of Glass Ceramics Fused to Titanium, 27(1).
- Vempati, R. K. (2002). U.S. Patent No. 6,444,186. Washington, DC: U.S. Patent and Trademark Office.
- Vempati, Rajan K., et al. "Surface analyses of pyrolysed rice husk using scanning force microscopy." Fuel 74.11 (1995): 1722-1725.
- Vieira, C. M. F., & Monteiro, S. N. (2007). Evaluation of a Plastic Clay from the State of Rio de Janeiro as a Component of Porcelain Tile Body, 1–7.
- Vieira, C. M. F., Peçanha Jr, L. A., & Monteiro, S. N. (2006). Effect of kaolinitic clays from the State of Rio de Janeiro in the composition of whiteware floor tile bodies. *Cerâmica*, 52(322), 138-145.
- Vinh, P. P., van Tol, I. A., van Paassen, I. L., & Ye, G. (2012). Utilization of Rice husk ash in GeoTechnology. GeoScience.
- Wansom, S., Janjaturaphan, S., & Sinthupinyo, S. (2010). Characterizing pozzolanic activity of rice husk ash by impedance spectroscopy. *Cement and Concrete Research*, 40(12), 1714–1722. Xu, W., Lo, T. Y., & Memon, S. A. (2012).

- Microstructure and reactivity of rich husk ash. *Construction and Building Materials*, 29, 541–547.
- Warshaw, S. I., & Seider, R. (1967). Comparison of strength of triaxial porcelains containing alumina and silica. *Journal of the American Ceramic Society*, 50(7), 337-343.
- Wei, Q., Gao, W., & Sui, X. (2010). Synthesis of low-temperature, fast, single-firing body for porcelain stoneware tiles with coal gangue. *Waste Management & Research*, 28(10), 944-950.
- Wetzel, A., Zurbriggen, R., & Herwegh, M. (2010). Spatially resolved evolution of adhesion properties of large porcelain tiles. *Cement and Concrete Composites*, 32(5), 327-338.
- Whittenberger, R. T. (1945). Silicon absorption by rye and sunflower. American Journal of Botany, 539-549.
- Wongpa, J., Kiattikomol, K., Jaturapitakkul, C., & Chindaprasirt, P. (2010). Compressive strength, modulus of elasticity, and water permeability of inorganic polymer concrete. Materials & Design, 31(10), 4748-4754.
- Wu, C. Z., Yin, X. L., Yuan, Z. H., Zhou, Z. Q., & Zhuang, X. S. (2010). The development of bioenergy technology in China. *Energy*, 35(11), 4445-4450.
- Wu, J., Leung, P. L., Li, J. Z., Stokes, M. J., & Li, M. T. W. (2000). EDXRF studies on blue and white Chinese Jingdezhen porcelain samples from the Yuan, Ming and Qing dynasties, (March 1999), 239–244.
- Yalc, N., & Sevinc, V. (2001). Studies on silica obtained from rice husk, 27, 219–224.
- Yan, X., Akiyama, H., Yagi, K., & Akimoto, H. (2009). Global estimations of the inventory and mitigation potential of methane emissions from rice cultivation conducted using the 2006 Intergovernmental Panel on Climate Change Guidelines. *Global Biogeochemical Cycles*, 23(2).
- Youssef, M. M., & Ghazal, H. B. G. (2011). Characterization and Reuse of Kiln Rollers Waste in the Manufacture of Ceramic Floor Tiles. *Journal of American Science*, 7(5).
- Yuvakkumar, R., Elango, V., Rajendran, V., & Kannan, N. (2012). High-purity nano silica powder from rice husk using a simple chemical method. *Journal of Experimental Nanoscience*, (ahead-of-print), 1-10.
- Zain, M. F. M., Islam, M. N., Mahmud, F., & Jamil, M. (2011). Production of rice husk ash for use in concrete as a supplementary cementitious material. *Construction and Building Materials*, 25(2), 798–805.
- Zain, M. F. M., Islam, M. N., Mahmud, F., & Jamil, M. (2011). Production of rice husk ash for use in concrete as a supplementary cementitious material. Construction and building materials, 25(2), 798-805.

- Zanelli, C., Raimondo, M., Guarini, G., & Dondi, M. (2011). The vitreous phase of porcelain stoneware: Composition, evolution during sintering and physical properties. *Journal of Non-Crystalline Solids*, 357(16), 3251-3260.
- Zarone, F., Russo, S., & Sorrentino, R. (2011). From porcelain-fused-to-metal to zirconia: clinical and experimental considerations. *Dental materials*, 27(1), 83-96.
- Zhang, H., Guo, T. W., Song, Z. X., Wang, X. J., & Xu, K. W. (2006). The effect of ZrSiN diffusion barrier on the bonding strength of titanium porcelain, 1–4.
- Zhang, H., Zhao, X., Ding, X., Lei, H., Chen, X., An, D., & Wang, Z. (2010). A study on the consecutive preparation of d-xylose and pure superfine silica from rice husk. *Bioresource technology*, 101(4), 1263-1267.
- Zou, C., Zhang, C., Li, B., Wang, S., & Cao, F. (2012). Microstructure and properties of porous silicon nitride ceramics prepared by gel-casting and gas pressure sintering. *Materials & Design*.
- Zuda, L., Drchalová, J., Rovnaník, P., Bayer, P., Keršner, Z., & Černý, R. (2010). Alkaliactivated aluminosilicate composite with heat-resistant lightweight aggregates exposed to high temperatures: Mechanical and water transport properties. *Cement and Concrete Composites*, 32(2), 157-163.790. (n.d.)
- Zuur, K. (2004) Sustainability through Environmental Technology: A Case Study on Malaysia/ Thai Palm Oil Industry. MSc. Thesis, Wageningen University.