

KAJIAN POTENSI TANAH LIAT, ZEOLIT DAN KARBON TERAKTIF
SEBAGAI PENJERAP KOMPOSIT UNTUK MERAHWAT LARUT
RESAPAN

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Tesis ini dikemukakan sebagai
memenuhi syarat penganugerahan
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DEDIKASI

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(Hj. Ridzuan bin Mat Deris & Hj. Rafiah binti Deraman)

isteri yang tercinta,

(Noorlailaafihiah binti Ahmad Zainuri)

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PTTA
PERPUSTAKAAN TUNKU AMINAH

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“Segala Puji dan Syukur bagi Allah Tuhan Semesta Alam, Selawat dan Salam
ke atas Junjungan Besar Nabi Muhammad S.A.W”*

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ABSTRAK

Larut resap domestik mengandungi kepekatan bahancemar $\text{NH}_3\text{-N}$ dan COD yang sangat tinggi. Kajian ini menggunakan campuran bahan penjerap semulajadi terdiri daripada tanah liat, zeolit dan karbon teraktif bagi menjerap bahancemar tersebut. Ujian pencirian penjerap komposit telah dilakukan dengan menggunakan pendarkilau sinar-x (XRF), mikroskop imbasan elektron (SEM), luas permukaan Brunauer Emmett Teller (BET), titratan Boehm, pH di caj titik sifar (pHzpc) kelajuan goncangan, masa sentuhan, saiz partikel dan dos penjerap. Kelajuan goncangan adalah pada 200 rpm, masa sentuhan 120 minit, pH 7 dan saiz partikel 2.36-3.35 mm. Sementara ujian nisbah optimum dijalankan melalui penjerapan kelompok. Keputusan analisa kajian mendapati, tanah liat dan zeolit boleh dikategorikan sebagai penjerap hidrofilik dan karbon teraktif pula penjerap hidrofobik dengan nisbah optimum pada nisbah 5:3 sesuai dengan tingkah laku penjerapan $\text{NH}_3\text{-N}$ dan COD ke atas penjerap. Keputusan ujian XRF komposit penjerap menunjukkan kehadiran kalsium oksida dan silika oksida yang tinggi sebagai sebatian utama. Ujian SEM pula menunjukkan penjerap komposit mempunyai permukaan liang yang heterogen dan kasar. Kajian perbandingan menunjukkan kapasiti penjerapan penjerap komposit terhadap $\text{NH}_3\text{-N}$ dan COD adalah lebih baik daripada zeolit dan karbon teraktif. Kajian kinetik penjerapan mendapati penjerap komposit mengikut hampir kesemua model yang dikaji namun model pseudo-tertib kedua adalah paling dominan untuk keseluruhan parameter.

ABSTRACT

Domestic leachate contains very high concentrations of NH₃-N and COD contaminants. This study uses a mixture of natural adsorbents consisting of clay, zeolite and activated carbon to adsorb the pollutants. Composite adsorbent characterization tests were performed using x-ray fluorescence (XRF), scanning electron microscope (SEM), Brunauer Emmett Teller surface area (BET), Boehm titration, pH at zero-point charge (pH_{zpc}) shake speed, contact time, size particles and adsorbent dose. While the optimum ratio test is carried out through batch adsorption. Shake speed was at 200 rpm, contact time 120 minutes, pH 7 and particle size ranging from 2.36mm-3.35 mm. The results of the study analysis found that clay and zeolite can be categorized as hydrophilic adsorbents and activated carbon is a hydrophobic adsorbent with an optimum ratio of 5: 3 in accordance with the adsorption behavior of NH₃-N and COD on the adsorbent. The XRF test results of the adsorbent composite showed the presence of high levels of calcium oxide and silica oxide as the main compounds. SEM test showed that the composite adsorbent has a heterogeneous and rough pore surface. Comparative studies showed that the adsorption capacity of composite adsorbents against NH₃-N and COD was better than that of zeolite and activated carbon. The adsorption kinetic study found that composite adsorbents according to almost all models studied however second-order pseudo-models were the most dominant for the overall parameters.

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SENARAI SINGKATAN

<i>AC</i>	-	Activated Carbon
<i>APHA</i>	-	American Public Health Association
<i>ASTM</i>	-	American Society for Testing and Materials
<i>ATR</i>	-	Attenuated Total Reflection
<i>AWWA</i>	-	American Water Works Association
<i>BET</i>	-	Brunauer, Emmet, Teller
<i>BOD</i>	-	Permintaan Oksigen Biokimia
<i>BS</i>	-	British Standards
<i>CCD</i>	-	Charge-Coupled Device
<i>COD</i>	-	Permintaan Oksigen Kimia
<i>EBCT</i>	-	Empty Bed Contact Time
<i>FA</i>	-	Asid Fulvik
<i>Fe</i>	-	Ferum
<i>FKAAS</i>	-	Fakulti Kejuruteraan Awam dan Alam Sekitar
<i>FTIR</i>	-	Spektroskopi Inframerah Transformasi Fourier
<i>GAC</i>	-	Granular Activated Carbon
<i>HA</i>	-	Asid Humik
<i>HCl</i>	-	Asid Hidroklorik
<i>JAS</i>	-	Jabatan Alam Sekitar
<i>KPK</i>	-	Kapasiti Pertukaran Kation
<i>KT</i>	-	Karbon Teraktif
<i>LOI</i>	-	Loss of Ignition
<i>MPRC</i>	-	Pusat Penyelidikan Pencemar Mikro
<i>MSW</i>	-	Municipal Solid Waste
<i>MTZ</i>	-	Mass Transfer Zone
<i>NaCl</i>	-	Natrium Klorida
<i>NaOH</i>	-	Natrium Hidroksida

<i>NH₃-N</i>	-	Ammonia-Nitrogen
<i>OPC</i>	-	Simen Portland Biasa
<i>PAC</i>		Powdered Activated Carbon
<i>pH_{ZPC}</i>	-	pH Caj Titik Sifar
<i>Pt.Co</i>	-	Platinum Cobalt
<i>RBC</i>	-	Penyentuh Biologi Berputar
<i>RECESS</i>	-	Pusat Penyelidikan Tanah Lembut
<i>RPM</i>	-	Pusingan Per Minit
<i>SEM</i>	-	Mikroskop Imbasan Elektron
<i>SS</i>	-	Pepejal Terampai
<i>TL</i>	-	Tanah Liat
<i>TPSR</i>	-	Tapak Pelupusan Simpang Renggam
<i>UTHM</i>	-	Universiti Tun Hussein Onn Malaysia
<i>VFA</i>	-	Asid Lemak Meruap
<i>WCAT</i>	-	Water Contact Angle Test
<i>WDPT</i>	-	Water Drop Penetration Test
<i>WEF</i>	-	Water Environment Federation
<i>XRF</i>	-	Analisis Pendarkilau Sinar-X
<i>ZEO</i>	-	Zeolit



SENARAI SIMBOL

C_b	-	Kepekatan efluen atau kepekatan bulus	mg/L
C_e	-	Kepekatan pada keseimbangan	mg/L
C_L	-	Had kepekatan	mg/L
C_o	-	Kepekatan awal	mg/L
C_t	-	Kepekatan bulus pada masa t	mg/L
H	-	Ketinggian bahan penjerap penjerap dalam turus	mg/L
k_1	-	Pemalar kadar pseudo-tertib pertama	cm atau m
k_2	-	Pemalar kadar pseudo-tertib kedua	(min ⁻¹)
k_{AB}	-	Pemalar Adams-Bohart	(g/mg min)
K_F	-	Pemalar Freundlich	(L/mg-min)
k_{TH}	-	Pemalar Thomas	mgg ⁻¹ (gm ⁻³) ⁿ
k_{YN}	-	Pemalar Yoon-Nelson	(mL/min-mg)
k_i	-	Pemalar kadar pembauran intra-partikel	(min ⁻¹)
K_L	-	Pemalar Langmuir	(mg/g min ^{0.5})
n	-	Pekali dalam persamaan Freundlich	m ³ g ⁻¹
N_o	-	Kapasiti penjerapan atau kapasiti bulus	-
Q	-	Kadar aliran volumetrik	mg/L
q_e	-	Amaun penjerapan pada keseimbangan	mL/min
q_m	-	Amaun penjerapan maksimum	mg/g
q_t	-	Amaun penjerapan pada masa t	mg/g
t_b	-	Masa bulus	mg/g
t_e	-	Masa tepu	minit
V_e	-	Isipadu turus kosong	minit
V_0	-	Halaju linear	cm ³ atau m ³
X	-	Jisim penjerap yang digunakan	cm/min
Z	-	Kedalaman lapisan katil turus	cm atau m

BAB 1

Pengenalan

1.1 Pendahuluan

Malaysia menuju ke arah negara maju dan melalui fasa pembangunan yang pesat dengan kewujudan pelbagai sektor seperti sektor perindustrian, pertanian, perkhidmatan dan pembinaan. Dengan perkembangan pelbagai industri tersebut, populasi serta kawasan perbandaran akan meningkat berikutan daripada migrasi penduduk luar bandar ke kawasan bandar. Seiring dengan perkembangan ekonomi yang pesat dan peningkatan jumlah populasi setempat, Malaysia berdepan dengan cabaran dalam pengurusan sisa pepejal domestik dan industri. Peningkatan bahan sisa pepejal, kekurangan tanah untuk tapak pelupusan, kos tanah untuk tapak pelupusan dan kos pengangkutan sisa pepejal ke tapak pelupusan adalah antara cabaran yang perlu ditangani oleh Pihak Berkuasa Tempatan (PBT) atau mana-mana pihak pengendali sisa pepejal di sesuatu kawasan.

Pengurusan sisa pepejal merupakan suatu cabaran baru dan berterusan di Malaysia dalam usaha membangunkan Malaysia sebagai negara maju menjelang abad ke-21. Pengurusan yang efektif perlu selari dengan komitmen Malaysia dalam Deklarasi Rio pada tahun 1992 yang menekankan tahap kesihatan awam yang baik, kehidupan manusia yang selesa dan seimbang serta bebas daripada masalah alam sekitar. Sementara deklarasi Rio, 2012-pula dipetik sebagai “Mengawal dan menguruskan sisa pepejal merupakan tunjang kemajuan dalam pembangunan sesebuah negara maju”.

Pengurusan sisa pepejal memerlukan kemudahan yang secukupnya untuk mengatasi masalah pelupusan sisa pepejal. Perancangan sistem pengurusan sisa pepejal bagi suatu kawasan perlu mengambil kira rancangan pembangunan, kadar

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